Structure and function of DNA

Structure and Function of Nucleic Acid

- Two types of nucleic acid are present in the cell
- Similar primary structure
- > Three basic components
 - 1. A nitrogenous base(Purine, Pyrimidine)
 - 2. Sugar
 - 3. Phosphate group

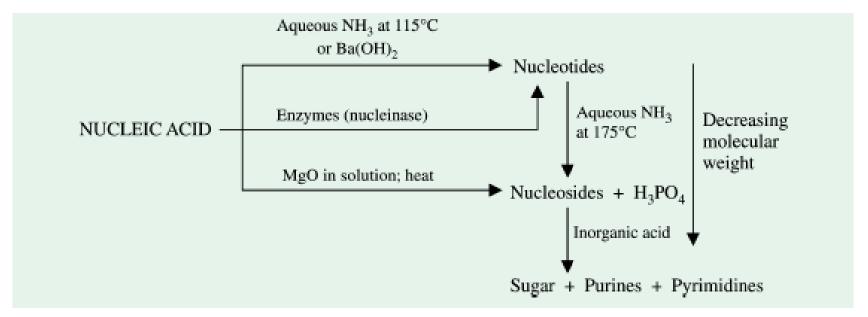
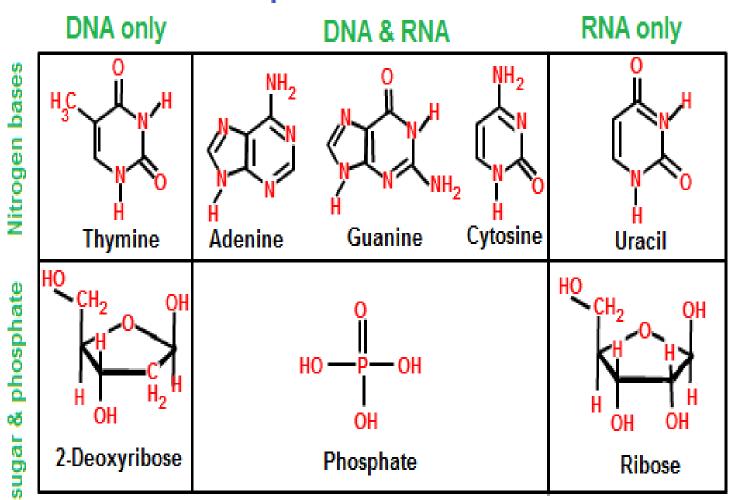
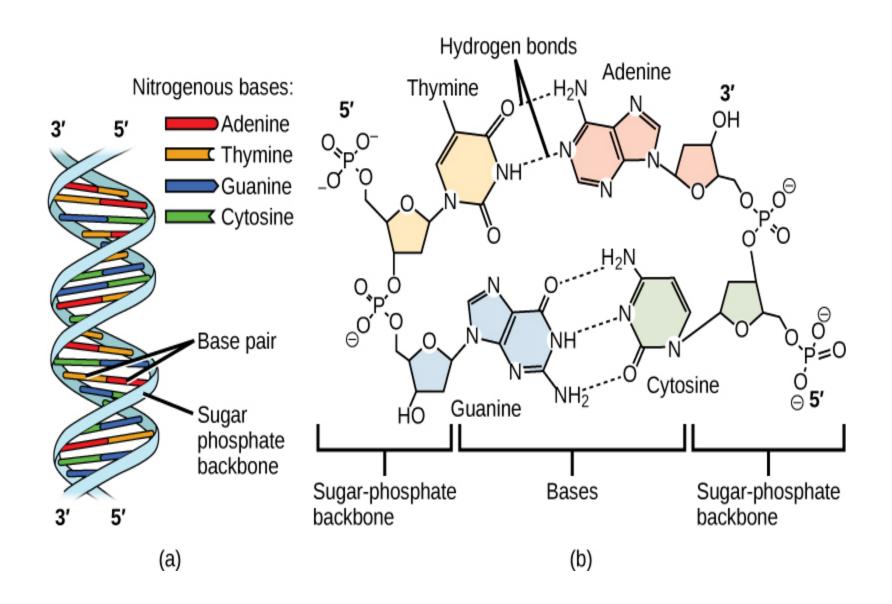
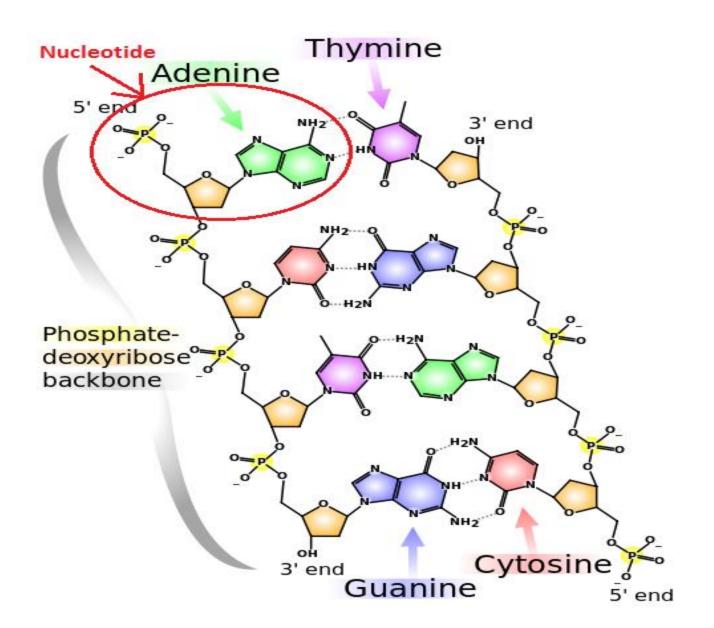


Fig. 15-1. Hydrolytic products of nucleic acid

Components of Nucleic Acids







Nomenclature of Nucleic Acid Components

Base

Nucleoside

Nucleotide

Nucleic acid

Purines

Adenine Adenosine Adenylate RNA

Deoxyadenosine Deoxyadenylate DNA

Guanine Guanosine Guanylate RNA

Deoxy guanosine Deoxyguanylate DNA

Pyrimidines

Cytosine Cytidine Cytidylate RNA

Deoxycytidine Deoxycytidylate DNA

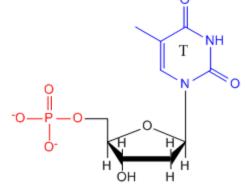
Thymine Thymidine Thymidylate DNA

(deoxythymidine) (deoxythymidylate)

Uracil Uridine Uridylate RNA

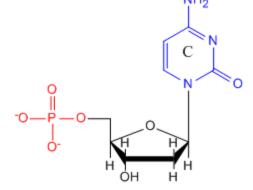
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deoxyadenosine 5'-phosphate

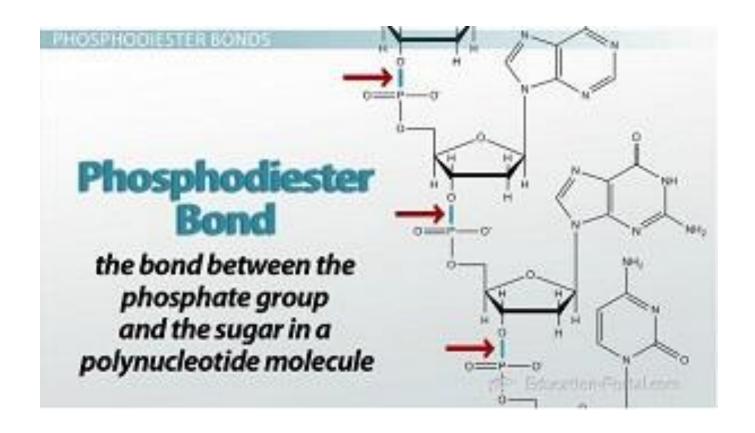


deoxythymidine 5'-phosphate

deoxyguanosine 5'-phosphate



deoxycytosine 5'-phosphate



- In 1869, Friedrich Miescher while studying the nucleus of animal cells found a phosphorus containing material in it. He called it as nuclein.
- Later isolated nuclein from nucleus of pus cells obtained from the used surgical bandages.
- Later it was fractionated into two parts.
 Acidic and basic part

- the base composition of DNA from different organisms vary and is unique for a given organism.
- > However, its composition was constant within a particular species and did not vary from individual to individual or from cell to cell within the same individual.
- > The number of purine nucleotides always equal to pyrimidines nucleotides.
- Watson and crick proposed famous model of DNA. According to this model, the DNA is present in a double stranded form. The two strands are running antiparallel to each other.
- > There are possible three hydrogen bond b/t G and C while only two between bonds can be formed between Agand T.

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- ➤ The hydrogen bonds are formed between the NH2 group of one base and =O of complementary base or between =NH of one and –N of other base.
- ➤ For the formation of stable hydrogen bonds, the N-N distance is 0.30nm and O-N is 0.28-0.29nm

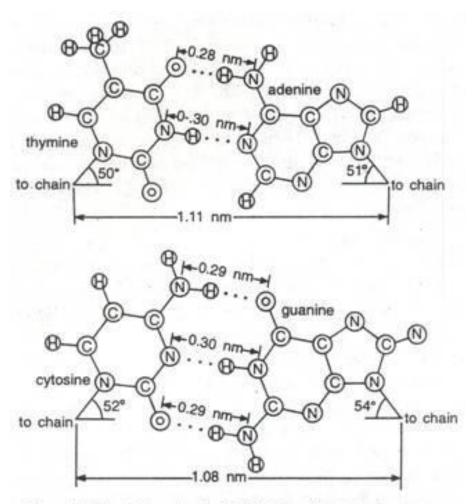
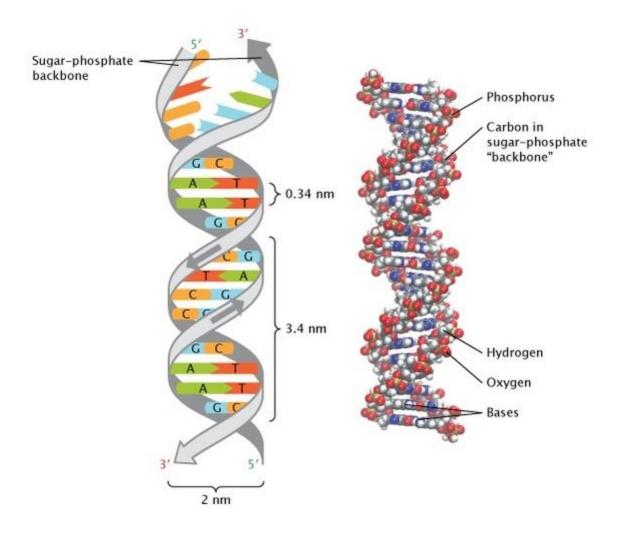


Fig. 3. Two base pairs in DNA. Complimentary bases are T=A and C=G linked with hydrogen bonds.

- two strand of DNA molecule are not present as a linear structure
- double helical structure
- > right hand helix having ten bases in a turn. Each base is thus at an angle of 36° to the adjoining base.
- > The diameter of helix is 20A.
- > Three dimensional structure of DNA has two grooves of different size in each turn. Major and Minor grooves.



Various forms of DNA

- Majority of DNA of the cell has three dimensional structure as represented by Watson and Crick. This form is B-DNA.
- Small amount of DNA having alternate molecular structure with certain differences from the B-DNA are also present.
- ➤ The first alternate structure is that of A form of DNA which varies having 11 base per turn and is more compact.
- > A form of DNA has close resemblance with ds RNA.
- Z-DNA has left handed helix. It contain 12 bases per turn and phosphodiester backbone is present in zigzag manner. Define major and minor groove are not present.

Various forms of DNA

- ➤ In experimental conditions the existence of z-DNA had been shown in the presence of high salt conditions or in presence of specific cations such as spermine and spermidine.
- > It has high degree of negative supercoiling and has certain specific proteins attached to it.
- ➢ Besides, there is usually degree of methylation of C residues at the C5 position in the Z-DNA.
- ➤ the precise role of alternate forms of DNA is not clear. These may play certain regulatory roles at a site near to them as at the distal sites also. May also play some role in stabilization of DNA.

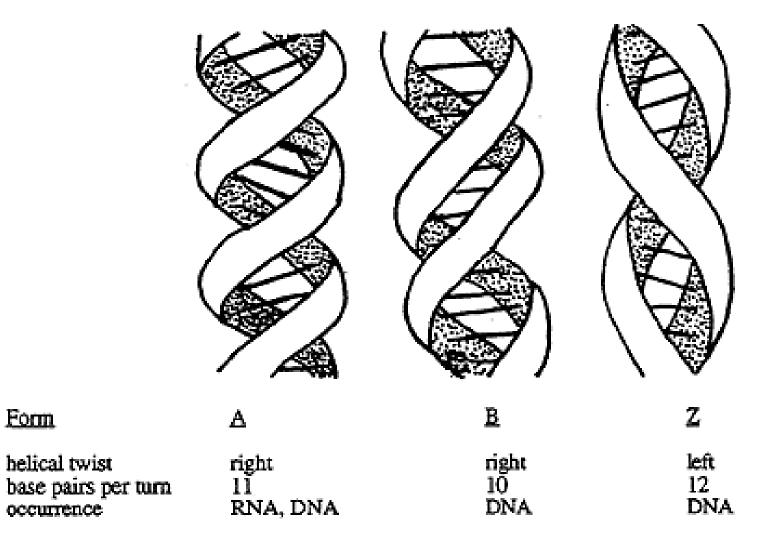


Table 15-6. Comparison of different forms of DNA

Characteristics	A-DNA	B-DNA	C-DNA	Z-DNA
Conditions	75% relative humidity; Na ⁺ , K ⁺ , Cs ⁺ ions	92% relative humidity; Low ion strength	60% relative humidity; Li ⁺ ions	Very high salt concentration
Shape	Broadest	Intermediate	Narrow	Narrowest
Helix sense	Right-handed	Right-handed	Right-handed	Left-handed
Helix diameter	25.5 Å	23.7 Å	19.0 Å	18.4 Å
Rise per base pair ('h')	2.3 Å	3.4 Ã	3.32 Å	3.8 Å
Base pairs per turn of helix ('n')	11	10.4	9.33	12 (= 6 dimers)