

INVESTIGATION OF REACTION MECHANISM

- Actual pathway of reaction is called reaction mechanism
- Mechanism is only suggestive not conclusive
- Mechanism is based on available evidences

TYPES OF MECHANISMS

1. HETEROLYTIC MECHANISMS

If a bond breaks in such a way that both electrons remain with one fragment

- Nucleophilic Reactions
- Electrophilic Reactions

2. HOMOLYTIC OR FREE-RADICAL MECHANISMS

If a bond breaks in such a way that each fragment gets one electron
Free radicals are formed

3. PERICYCLIC MECHANISMS

Electrons move in a closed ring
No intermediates, ions or free radicals are formed

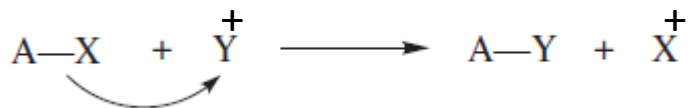
TYPES OF REACTIONS

1. SUBSTITUTIONS

(i) Nucleophilic substitution



(ii) Electrophilic substitution

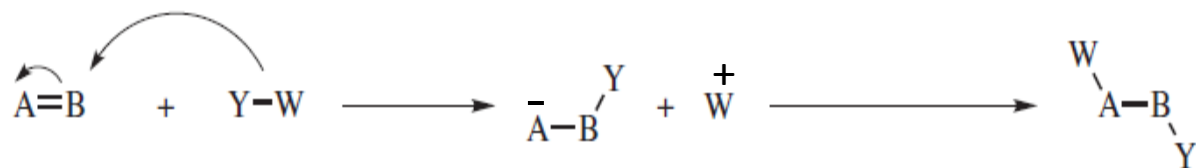


(iii) Free-radical substitution

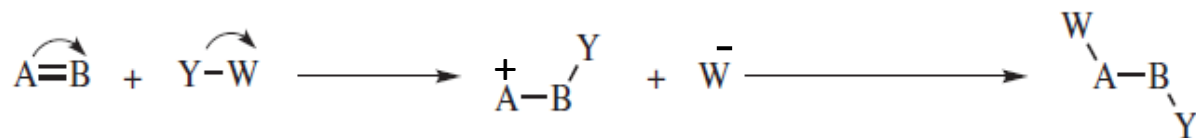


2. ADDITIONS

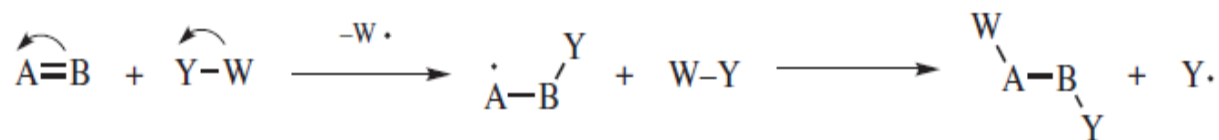
(i) Nucleophilic addition



(ii) Electrophilic addition



(iii) Free-radical addition



(iv) Simultaneous addition

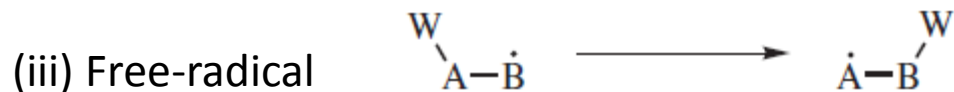
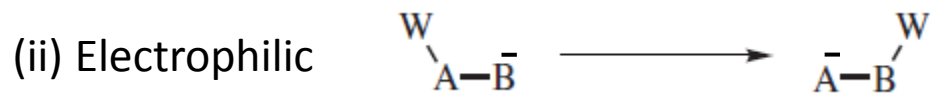
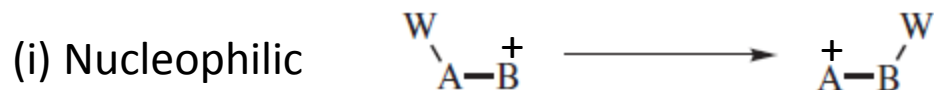


3. ELIMINATIONS



Heterolytic or Pericyclic mechanisms

4. REARRANGEMENTS



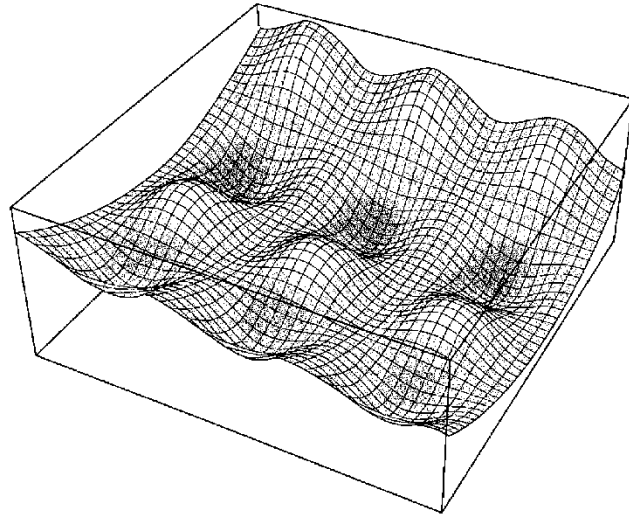
5. REDOX REACTIONS

May be substitutions, eliminations, additions, rearrangements types
May be of some other type

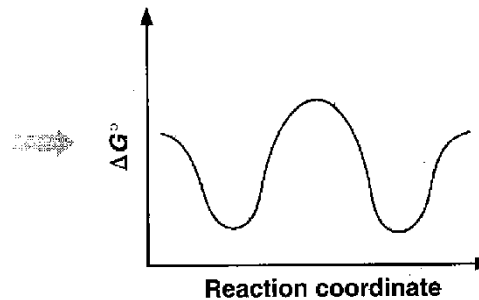
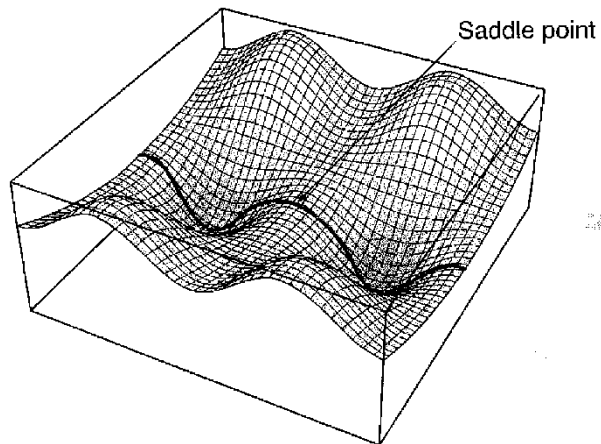
6. COMBINATIONS OF THE ABOVE

Addition-Elimination, Oxidative-Addition, Reductive-Elimination etc.

POTENTIAL ENERGY SURFACE

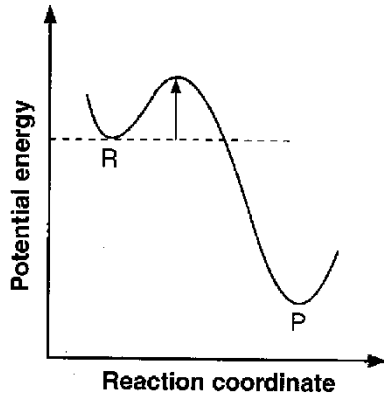


1. Saddle Points
2. Transition State
3. Intermediate
4. Reaction Coordinate diagrams

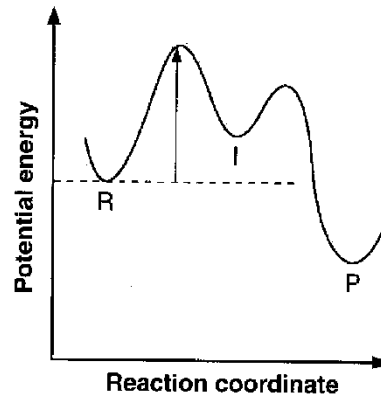


REACTION COORDINATE DIAGRAM

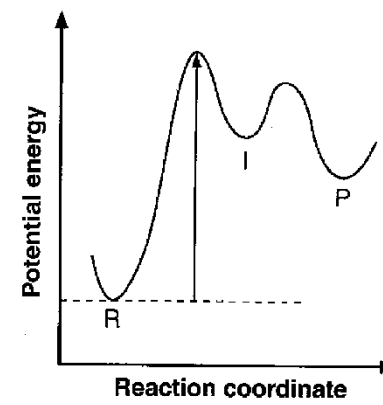
A.



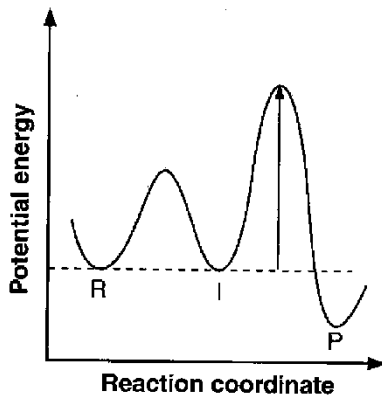
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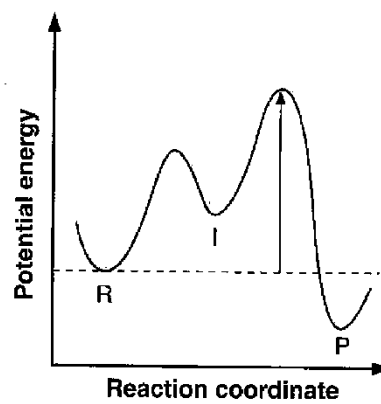
C.



D.



E.



F.

