

Metabolic engineering of microorganisms for biofuel production

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ABSTRACT

Microorganisms directly and indirectly contribute to production of diverse biofuels. Heterotrophic microorganisms are being used for commercial production of biofuels such as biogas and fuel alcohols from organic matter. Photosynthetic microorganisms convert inorganic carbon and water to potential fuels (e.g. fuel alcohols, biohydrogen) and fuel precursors (e.g. biomass, starch, lipids). Only a few microbial processes are used for commercial production of biofuels, but this will certainly change with the enhanced production capabilities being achieved through microbial metabolic engineering. Processes that previously required multiple steps of feedstock pretreatment and subsequent conversion to fuel are being consolidated into single-step microbial processes using metabolically engineered species. Microorganisms with the ability to produce fuels from feedstock they could not use previously, are being engineered. This review discusses some of the metabolic engineering approaches being used to enhance the commercialization potential of microbial biofuels including fuel alcohols, biodiesel and biohydrogen. At present, all biogas production relies on native populations of methanogens and this does not seem likely to change in the near term. Potential fuels from microalgae, cyanobacteria and other photosynthetic bacteria, whether native or engineered, have distant prospects of commercial use. Metabolically engineered yeasts surface displaying various hydrolytic enzymes appear to hold the greatest potential for near term commercial use in generating bioethanol from starch, pretreated lignocellulose and other polysaccharides. The bacterium *Zymomonas mobilis* metabolically engineered to make bioethanol from pentose sugars is already being commercialized. Other similar examples are likely to emerge as more engineered microorganisms become available.

1. Introduction

The global energy consumption continues to increase in keeping up with increasing industrialization and improving quality of life. Fossil fuels are currently our main sources of energy, but renewable energy carriers including biofuels have been receiving increasing attention [1,2]. Biofuels include mainly bioethanol, biodiesel, biogas, and biomass. Displacing a part of the fossil fuels with renewable biofuels has the potential to reduce emissions of greenhouse gases and improve the environmental quality [3–6].

All biofuels represent energy captured from sunlight. More specifically, photosynthesizing microorganisms and higher plants capture sunlight and use it to convert carbon dioxide and water to biochemicals. Some of these biochemicals may be used directly as fuels. In other

cases, they may serve as feedstock for making fuels either through the action of other microorganisms [7] or via chemical processes. For example, starch and cellulose produced via photosynthesis may be converted into bioethanol through the action of non-photosynthesizing microorganisms. Similarly, photosynthetically-produced biomass and oils can be converted into fuels such as biodiesel and gasoline using chemical technologies. Microbial biomass generated via photosynthesis may eliminate the need for arable land for fuel production and, hence, reduce potential impact of biofuels on production of food and fodder [8,9].

Production of biofuels through microbial photosynthesis and microbial bioconversion of an organic feedstock can be greatly improved by genetic and metabolic enhancements of the relevant microorganisms. Traditionally, microbial strains for industrial applications have

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been improved by selection, random mutagenesis and screening processes [10,11]. However, these processes are largely uncontrolled, slow, and unpredictable. In some cases, it may be impossible to obtain a desired phenotype through mutagenesis and selection if the relevant biochemistry is missing in a microorganism [12]. Therefore, the enhancement of microorganisms for various purposes increasingly relies on rational engineering including genetic and metabolic engineering [12,13]. Genetic and metabolic engineering are being used to modify the surfaces of the cells to display novel proteins and other substances for improving biocatalytic capability of the cells [14,15]. For example, microorganisms displaying specific enzymes on the surface may improve enzymatic conversion of starch into sugars for fermentation to fuel alcohols. Moreover, surface display of desired proteins in engineered cells allows easy reproduction of the displayed biocatalyst and easy separation of the product from the catalyst [16,17]. Cells displaying a suite of the relevant surface enzymes and having the relevant intracellular metabolic pathways could greatly improve the ability to convert certain feedstocks to biofuels. Metabolic engineering can provide microorganisms with improved titers and yields of products; with the capacity to use an expanded range of carbon substrates; and with the ability to better withstand high titers of potentially toxic products and adverse environmental conditions [18]. Specific biochemical pathways can be enhanced by manipulating the production and regulation of specific enzymes and deletion of pathways that may be counterproductive. Entirely new metabolic pathways taken from one microorganism may also be engineered into a superior host organism [14]. Overall, metabolic engineering allows precise modification of microbial species without causing an accumulation of unfavorable mutations [14].

Metabolic engineering together with genetic engineering are among core technologies for enhanced production microbial biofuels. Continuing developments in synthetic biology, genetic engineering and manipulation of enzyme expression allows biofuel-producing pathways to be inserted in a microbial host [19]. This review discusses some of the metabolic engineering approaches used in production of main types of biofuels including bioethanol, biobutanol, other alcohols, biodiesel, and biohydrogen. Developments in fuel production using engineered cyanobacteria, other bacteria, microalgae, and yeasts are discussed. The prospects of the use of system metabolic engineering for generating superior biofuel-producers strains are assessed.

2. Metabolic engineering for biofuels production

Product diversity, yield, concentration, and productivity may be enhanced in metabolically-engineered microorganisms for enhanced fuel production [20–22]. Moreover, downstream recovery of a certain biofuel may also be simplified through metabolic engineering. Furthermore, the cost of fuel production may be reduced by engineering microorganisms to use cheap substrates [23]. Some metabolic pathways and other routes used in microbial production of fuel precursors and common fuels are summarized in Fig. 1 [24,25].

2.1. Bacteria

Numerous bacteria are able to convert various substrates into fuels such as biohydrogen, bioalcohols, and biogas. Novel bacteria with enhanced production potential for fuels are continually being sought from nature, in particular from thermophilic environments. Recent advances in the use of bacteria for producing alcohols and hydrogen are discussed in the following sections. Methanogenic archaea and production of biogas have been extensively reviewed elsewhere [26–31].

2.1.1. Bioethanol

2.1.1.1. *Zymomonas mobilis*. Bioethanol is the most commonly produced biofuel and is used as an additive to gasoline. Owing to its clean burning characteristics, fuel ethanol can significantly reduce

emissions of greenhouse gases including carbon dioxide [32]. The anaerobic bacterium *Zymomonas mobilis* is perhaps the best known native producer of bioethanol [33,34]. Compared to yeasts, *Z. mobilis* more efficiently converts glucose to bioethanol, but cannot naturally use most other sugars.

Z. mobilis and the classical ethanologenic yeast *Saccharomyces cerevisiae* both share a natively-expressed homoethanol pathway. However, they differ in terms of their glycolysis; *S. cerevisiae* uses the Embden-Meyerhof-Parnas (EMP) pathway while *Z. mobilis* uses the Entner-Doudoroff (ED) pathway [35]. The ED pathway is 50% more efficient than the EMP pathway leading to less ATP consumption during the ethanologenic process. In addition, *Z. mobilis* is capable of consuming glucose faster than *S. cerevisiae* apparently because of its high cell specific surface area. These factors collectively result in a higher ethanol productivity in *Z. mobilis* in comparison with *S. cerevisiae* [36]. Ethanol production is mainly controlled by the PET operon encoding pyruvate decarboxylase and alcohol dehydrogenase. On a given amount of glucose, *Z. mobilis* generally produces less biomass than yeast [34] and this is considered an advantage. In addition to ethanol, other by-products may also be produced [36].

The inability of *Z. mobilis* to convert pentose sugars to ethanol is its main drawback as cheap pentose sugars are abundantly available in lignocellulosic feedstocks. This problem has been recently addressed by metabolic engineering of pentose utilization capability in *Z. mobilis* [37]. Ethanol production using this technology is being commercialized by DuPont. A recombinant *Z. mobilis* strain TMY-HFPX with multiple improved characteristics was developed and shown to achieve an ethanol concentration of up to 136 g/L in a solution containing 295 g/L glucose (90% of theoretical yield) without requiring exogenous amino acids and vitamins [38]. The enhanced strain reportedly harbored multiple gene modules, i.e., the *xylA/xlyB/tktA/talB* operon for xylose utilization, the *metB/yfdZ* operon for lysine and methionine biosynthesis, the thioesterase gene *tesA* to enhance free fatty acid biosynthesis for increasing ethanol tolerance, a proton-buffering peptide operon for acid stress tolerance, and a small heat shock protein operon for heat stress tolerance [38].

2.1.1.2. *Escherichia coli*. The bacteria *Escherichia coli* and *Bacillus subtilis* have been most widely used to construct ethanol producing strains because the molecular biology of these microorganisms is well understood. They are capable of utilizing diverse substrates and are amenable to metabolic engineering for exclusive production of ethanol [39–44]. *E. coli* was probably the first microorganism to be successfully modified for ethanol production through metabolic engineering [45]. This initial success was further built upon by introducing the relevant foreign genes in *E. coli* and disrupting pathways for competing products. Thus, the *E. coli* W (wild) was transformed into *E. coli* KO11, a novel strain capable of producing ethanol [33]. *E. coli* KO11 contained genes from *Z. mobilis* for encoding pyruvate decarboxylase and alcohol dehydrogenase (PET operon). In addition, the gene for production of fumarate reductase was deleted. *E. coli* KO11 was successfully adapted to produce ethanol at nearly 95% of the theoretical yield in a complex medium and showed increased ethanol tolerance relative to the original host. However, this strain was unable to grow in ethanol concentration of 3.5% and required complex nutritional supplements which contributed to the cost of ethanol production [33].

In a different investigation, ethanol tolerance of the strain KO11 was increased by 10% by adaptive evolution and selection through long term exposure to media supplemented with increasing ethanol content. The adapted strain was LY01. However, both LY01 and its parent KO11 required complex nutritional supplements and this added to the expenses of ethanol production. Therefore, *E. coli* KO11 was further engineered to strain SZ110 to allow ethanol production in less nutritionally-demanding media [45]. The strain SZ110 was engineered by eliminating lactate dehydrogenase gene and inserting pyruvate formate

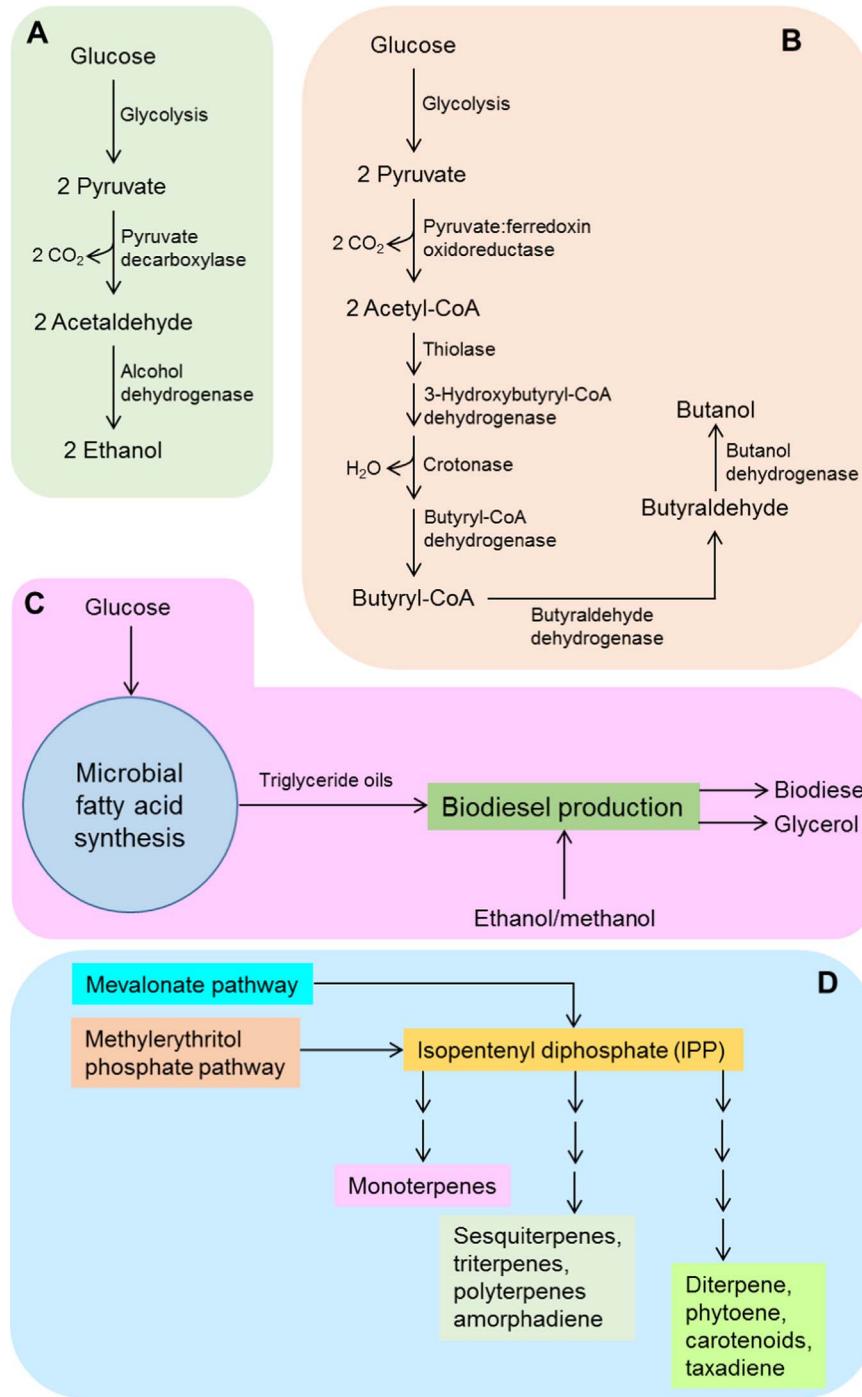


Fig. 1. Major pathways for the production of: ethanol (A); butanol (B); biodiesel (C); and isoprenoid fuel precursors [24] (D).

lyase and PET operon from *Z. mobilis* [33]. Further engineering of SZ110 by transposon-mediated mutagenesis and metabolic evolution provided the strain LY168 [46]. First, the transposon containing promoterless *pdc*, *adhA*, and *adhB* genes were randomly inserted into the chromosome of SZ110. Functional selections were subsequently performed by serial transfers in mineral salts media without antibiotics [47]. Eventually, the combination of genetic engineering and long term adaptation resulted in microbial biocatalysts that produced up to 45 g ethanol/L in 48 h in a simple mineral salts medium. Overall, the strain LY168 successfully grew and produced ethanol in media containing only mineral salts, a high concentration of sugar and the osmoprotectant betaine. The strain LY168 could reportedly produce 0.5 g ethanol/g xylose, or close to the maximum theoretical yield of 0.51, in a minimal medium provided with betaine [33]. The key difference

between the strains LY168 and KO11 was how the expression of the homoethanol pathway was regulated. In strain KO11, the *Z. mobilis pdc* and *adb* genes were regulated by *pfl* promoter with multiple layers of transcriptional control, while in strain LY168, the genes encoding the homoethanol pathway were integrated within the gene encoding 23S ribosomal RNA subunit *rrlE*, concurrent with the direction of transcription.

Recently, the glycolytic pathway of *E. coli* KO11 was engineered by knocking out phosphoglucose isomerase (*pgi*) to channel carbon flux from glucose through the Entner–Doudoroff (ED-P) and pentose phosphate (PP-P) pathways to lactate and acetate production [48]. However, the engineered strain grew quite slowly under non-aerated conditions in minimal media containing 40 g/L glucose. To improve KO11 Δ*pgi*'s capacity to grow, the recombinant strain was evolved for 60 days

Table 1

Ethanol yield of some genetically engineered thermophilic bacteria.

Microorganism	Substrate	Genotype ^a	Mode	Ethanol yield (mol/mol)	Reference
<i>Clostridium thermocellum</i>	Cellobiose	ΔpyrF, Δpta::gapDHp-cat	Batch	0.59	[55]
<i>C. thermocellum</i>	Avicel	ΔpyrF, Δpta::gapDHp-cat	Batch	0.71	[55]
<i>C. thermocellum adhE*(EA) Aldh</i>	Cellobiose	Δhpt, Δldh	Batch	0.37	[56]
<i>C. thermocellum</i>	Avicel	Δhpt, Δldh, Δpta (evolved)	Batch	1.08	[56]
<i>C. thermocellum/Thermoanaerobacterium saccharolyticum</i>	Avicel	Δhpt, Δldh, Δpta (evolved) and Δpta, Δack, Δldh	Batch	1.26	[56]
<i>T. saccharolyticum</i> TD1	Xylose	Δldh	Batch	0.98	[56]
<i>T. saccharolyticum</i> ALK2	Cellobiose	Δpta, Δack, Δldh	Continuous	ND	[57]
<i>T. saccharolyticum</i> HK07	Cellobiose	Δldh, Δhfs	Batch	0.86	[58]
<i>T. saccharolyticum</i> M0355	Cellobiose	Δldh, Δack Δpta	Batch	1.73	[59]
<i>T. saccharolyticum</i> M1051	Cellobiose	Δldh, Δack Δpta, ureABCDEF	Batch	1.73	[57]
<i>Geobacillus thermoglucofermentans</i> TM242	Glucose	Δldh-, pdh up, pfb-	Batch	1.73	[60]
<i>G. thermoglucofermentans</i> TM242	Glucose	Δldh-, pdh up, pfb-	Batch	1.84	[60]
<i>G. thermoglucofermentans</i> TM242	Xylose	Δldh-, pdh up, pfb-	Batch	1.34	[60]
<i>Thermoanaerobacter mathranii</i> BG1L1	Wheat straw	Δldh	Continuous	1.53–1.67	[61]
<i>T. mathranii</i> BG1G1	Glucose + glycerol	Δldh, GldA	Batch	1.68	[62]
<i>T. mathranii</i> BG1G1	Xylose + glycerol	Δldh, GldA	Batch	1.57	[62]
<i>T. mathranii</i> BG1G1	Xylose + glycerol	Δldh, GldA	Continuous	1.53	[62]

^a ack, acetate kinase; GldA, glycerol dehydrogenase A; hfs, hydrogenase; hpt, hypoxanthine phosphoribosyl transferase; pdh, pyruvate decarboxylase; pyrF, orotidine-5-phosphate decarboxylase; pfb, pyruvate formate lyase; ure, urease; ND, not determined.

and renamed KO11 E35. Compared with its parental strain KO11, the KO11 E35 had higher activities of glucose-6-phosphate dehydrogenase and the ED-P enzymes. Subsequently, KO11 E35 was further improved by deleting the *pta*, *ack*, and *ldh* genes leading to a homoethanologenic derivative called KO11 PPAL strain. This strain possessed specific ethanol production rates similar to those achieved by KO11, but diverted less glucose to biomass production. This led to larger yields of ethanol on glucose and biomass. These improved characteristics were ascribed to the strain's capability to channel most of the carbon flux from pyruvate to ethanol as well as increased expression of heterologous pyruvate decarboxylase and alcohol dehydrogenase from *Z. mobilis* [48].

In a different study, the homoethanol producer *E. coli* SE2378 (*Δldh Apfl*) was obtained via mutagenesis and had an ethanol productivity of 2.24 g/(h g-cells) at nearly 80% of the theoretical ethanol yield [46]. Another homoethanol producer *E. coli* SZ420 (*ΔfrdB C Δldh ΔackA ΔfolA- pfl ΔpdhR::pflBp6-aceEF-lpd*) was made by deleting the competing fermentation pathways in *E. coli* B and enabling overexpression of pyruvate dehydrogenase complex (aceEF-lpd, a typical aerobically-expressed operon) under anaerobic conditions [47]. SZ420 could convert xylose to ethanol at nearly 90% of the theoretical yield [47]. Adaptive evolution of nontransgenic *E. coli* KC01 (*ldhA pflB ackA frdB C pdhR::pflBp6-aceEF-lpd*) to obtain a homoethanol producer was also reported [49]. The resulting bacterium converted xylose to ethanol with 94% of theoretical yield [49].

As discussed above, *E. coli* is attractive for metabolic engineering for fuel production owing to the fact that its metabolic pathways are easily deleted or replaced. However, from a bioprocessing point of view, ethanol production using engineered *E. coli* has limitations in comparison with eukaryotic production systems. *E. coli* can be highly susceptible to phage infection and the utility of its spent biomass is limited in view of enteric origins [50].

2.1.1.3. *Bacillus subtilis*. Construction of *B. subtilis* BS35 was reported through disruption of the native lactate dehydrogenase (LDH) gene (*ldh*) by chromosomal insertion of the *Z. mobilis* pyruvate decarboxylase gene (*pdc*) and alcohol dehydrogenase II gene (*adhB*) under the control of the *ldh* native promoter [43]. The BS35 reportedly produced ethanol and butanediol, but the rates of cell growth and glucose consumption were reduced by 65–70% in comparison with the wildtype. BS35 was further modified to BS36 (BS35 *ΔalsS*) by eliminating the butanediol production pathway. This increased the ethanol yield to 89% of the

theoretical value, but rates of cell growth and glucose consumption still remained low. The strain BS37 (BS35 *ΔalsS udhA*⁺) was then developed from BS35 by inactivation of *alsS* through chromosomal integration of *E. coli* transhydrogenase gene (*udhA*). BS37 showed ethanol titers of up to 8.9 g/L in long-term cultivation [44].

2.1.1.4. Thermophilic bacteria. An important problem with mesophilic bacteria such as *E. coli* and *B. subtilis* is their poor ability to hydrolyze carbohydrate polymers, poor tolerance to extreme pH values, and inability to withstand high salt concentrations [7]. Therefore, bioprocesses based on these microorganisms are easily contaminated with other unwanted species, making them poorly suited for use in large-scale production operations.

Although bioethanol is commercially produced from sugars such as sucrose and starch hydrolysates derived from food crops [51,52], this form of production competes with food supply and arable land for production of food and fodder [53]. Therefore, processes capable of using non-food lignocellulosic biomass as a source of sugars for making ethanol are highly desirable. While ethanol production from lignocellulosic biomass is possible, it requires expensive pretreatment of the biomass to release fermentable sugars [54]. Hence, microorganisms with the ability to hydrolyze lignocellulosic materials and simultaneously convert the produced sugars into ethanol are of interest. Genetically engineered thermophilic bacteria are likely to be most useful in this context [34]. A thermophilic operation has the potential to not only enhance the rate of biomass hydrolysis and fermentation of sugars into ethanol, but also to reduce the potential of contamination of the fermentation process by unwanted microbes. Genetically engineered thermophilic bacteria have been extensively investigated for production of ethanol as summarized in Table 1.

2.1.2. Butanol and higher alcohols

Microbial production of primary higher alcohols such as 1-propanol, 1-butanol, and other linear-chain fatty alcohols has been widely investigated [63]. Butanol, a four carbon primary alcohol (C₄H₁₀O), is considered to be superior to ethanol as a fuel because of its higher energy density, lower vapor pressure, and being less hygroscopic and corrosive [64–67]. Moreover, due to relative similarity of its properties to gasoline, it is well compatible with the currently in-place gasoline infrastructure [68]. Therefore, butanol and other higher alcohols have the potential to directly substitute gasoline. Unmodified gasoline car engines fueled with pure butanol have successfully operated over long

periods [69]. Butanol is traditionally produced via the acetone-butanol-ethanol (ABE) fermentation, one of the first large-scale industrial fermentation processes [70]. Member of the bacterial genus *Clostridium* have been widely investigated as natural producers of butanol [68,71]. However, metabolic engineering of clostridial strains is challenging due to a complex metabolic regulation system and difficulties in their genetic manipulation [72]. In addition to clostridial strains, genetically engineered *E. coli* strains are also increasingly attracting attention as butanol producers [73,74].

2.1.2.1. *Clostridium* sp. Fermentative production of butanol by *Clostridium* sp. involves two distinct phases of acidogenesis and solventogenesis [75]. In the acidogenic phase, cells grow and convert the substrate to acetic and butyric acids and, therefore, during this phase pH declines to below 5. In the solventogenesis, the organic acids are transformed into acetone, butanol, and ethanol in the ratio of 3:6:1 [66,68]. Acetone and ethanol are inevitable byproducts of a conventional ABE fermentation. The metabolic engineering of clostridia has been attempted with a focus on selective overproduction of butanol to increase its yield and reduce the cost of removing the unwanted byproducts. This has mainly involved the use of integrative plasmids to inactivate the key genes of the biosynthetic pathways of unwanted byproducts [76,77]. Also, as n-butanol is extremely toxic to host cells, a lot of effort has focussed on developing more robust strains through metabolic engineering or mutagenesis approaches [68].

Clostridium acetobutylicum M5, a derivative of *C. acetobutylicum* ATCC 824, is basically incapable of producing butanol due to a loss of the megaplasmid pSOL1 [78]. This bacterium was later engineered to achieve enhanced butanol production [76,77,79]. For instance, overexpression of the *adhE1* gene under the control of the *ptb* promoter led to butanol production of up to 0.84 g butanol/g ABE. This was much higher than ~0.6 g butanol/g ABE produced by wildtype clostridia. A different production enhancement approach involved the disruption of the *adc* gene encoding acetoacetate decarboxylase in *C. acetobutylicum* which enhanced selective production of butanol to 0.82 g butanol/g ABE in a medium supplemented with methyl viologen.

Harris et al. metabolically engineered *C. acetobutylicum* PJC4BK and claimed a butanol concentration of up to 16.7 g/L compared with 11.7 g/L for wildtype *C. acetobutylicum* [80]. The PJC4BK was made by disruption of the *buk* gene encoding butyrate kinase, the enzyme involved in butyrate formation pathway [81]. Final butanol titer can also be increased by using other methods such as enhancing the product tolerance in a microorganism through construction of mutants [82]. For instance, Tomas et al. used overexpression of the molecular chaperone GroESL to enhance the activity of the solventogenic enzymes, increase butanol tolerance and eventually boost the final butanol titer [83]. Major achievements in clostridial 1-butanol production have been recently reviewed by Moon et al. [72].

2.1.2.2. *E. coli*. Solventogenic clostridia are strict anaerobes and grow slowly compared with other commonly known aerobic bacteria [12]. This adversely impacts the economics of butanol production by the conventional ABE fermentation. Furthermore, genetic manipulation of *Clostridium* sp. has proven to be difficult and this has shifted the focus towards engineering the clostridial fermentative pathway in bacteria such as *E. coli* [74,84] and *B. subtilis* [11,85,86] for butanol production. In an effort by Atsumi et al., a functional butanol biosynthesis pathway was reconstructed in *E. coli* by introducing the *thl*, *hbd*, *crt*, *bcd*, *efAB* and *adhE2* genes from *C. acetobutylicum* [73]. This allowed a butanol titer of 139 mg/L to be attained under anaerobic conditions [73]. In another study, a metabolically-engineered *E. coli* harboring the *thl*, *crt*, *bcd*, *efAB*, *hbd*, and *adhE2* genes of *C. acetobutylicum* produced 1.2 g/L of butanol from 40 g/L glucose [87].

Despite these achievements, a major bottleneck to further improvements in butanol production remains the lack of sufficient

understanding of how the metabolic shift from acid to solvent production is regulated at the molecular level. The relevant inducing signals, the regulators and their interactions, and the connections among the regulatory networks are poorly understood [88–90].

2.1.2.3. Other bacteria. Other bacteria including *Corynebacterium glutamicum* [91–94] and *Ralstonia eutropha* [95] have also been used to produce higher alcohols from 2-ketoacids, the metabolic intermediates of amino acid biosynthetic pathway. A *C. glutamicum* made by the deletion of the *ilvE*, *aceE*, and *pqo* genes encoding transaminase B, pyruvate dehydrogenase subunit E1, and pyruvate:quinone oxidoreductase, respectively, and by the overexpression of the *ilvBNCD* genes was shown to possess an elevated ability to produce isobutanol from glucose [92].

An engineered *R. eutropha* strain was confirmed to convert carbon dioxide to higher alcohols in an electromicrobial process [96]. This process involved the use of electrical current in a specially designed bioreactor. The engineered bacterium could produce nearly 90 mg/L of isobutanol and 50 mg/L of 3-methyl-1-butanol using CO₂ as the sole carbon source. Autolithotrophic bacteria such as *R. eutropha* can use electricity instead of hydrogen as source of electrons [96] and, therefore, may be promising for biofuel production processes if there is a net energy benefit. Table 2 summarizes the studies reported on production of biobutanol and higher alcohols by the use of metabolically-engineered microorganisms. The metabolic pathways involved in the production of higher alcohols are summarized in Fig. 2.

2.1.3. Biohydrogen

Hydrogen burns cleanly leaving only water. It is therefore, believed to be a promising future fuel [119–122]. Metabolic engineering of bacteria for production of biohydrogen has received much attention, but commercial production does not seem to be viable at present. In bacteria, hydrogen may be produced by photofermentation and dark fermentation. Photoheterotrophic hydrogen production requires an organic carbon source as well as sunlight and is carried out by anoxygenic photosynthetic bacteria such as the purple nonsulfur bacteria [123]. Nitrogenase is the key enzymes involved in the process and anaerobic conditions are required for hydrogen production [124,125]. The hydrogen production activity of nitrogenases tends to be low and the expression of these enzymes is repressed by ammonium ions [126,127]. Moreover, various other reactions competitively scavenge electrons from nitrogenase leading to reduced hydrogen yield and photochemical efficiency [124,128,129].

Biohydrogen production by dark fermentation of various substrates is mainly carried out by facultative and obligate anaerobic bacteria [130]. Dark fermentation results in higher rates of hydrogen production relative to photofermentation [123,131]. Both mesophilic and thermophilic dark fermentation processes exist depending on the bacterial species involved. Both wildtype and metabolically-engineered bacteria have been studied for hydrogen production [125,126]. Mesophilic bacteria such as *Clostridium butyricum*, *Enterobacter aerogenes*, and *E. coli* have attracted particular attention [132–134]. Table 3 summarizes the studies conducted on hydrogen production using metabolically engineered bacteria.

2.1.3.1. *E. coli*. Yoshida et al. reported on a genetically modified *E. coli* for hydrogen production by dark fermentation [135]. The bacterium was engineered by inactivating the formate-hydrogen lyase (FHL) repressor (*hydA*) and the negative regulator for FHL. In addition, the bacterium was modified to overexpress FHL and its activator genes (*fhlA*). The resulting organism showed a high volumetric hydrogen productivity of 300 L H₂/L/h in a high cell-density fermentation (93 g biomass dry weight/L) with formate as the substrate. Subsequently, the bacterium was further modified through disruption of *ldhA* encoding lactate dehydrogenase and *frdBC* encoding fumarate dehydrogenase in attempts to improve hydrogen yield [136]. Using this strain, up to

Table 2
Engineered microorganisms used for production of biobutanol and higher alcohols.

Product	Host	Genotype (knockout; overexpression)	Metabolic engineering approach	Substrate	Medium	Titer (g/L)	Cultivation	Comment	Reference
1-Propanol	<i>E. coli</i> BW25113	<i>ΔibA ΔibB; LiliVd ScADH2M MJcmA^{mut} leuABCD</i>	Directed evolution of <i>Methanococcus jannaschii</i> citramalate synthase (CimA) in <i>E. coli</i>	Glucose	Defined	2.78	Shake flask	1-Butanol 0.39 g/L	[98]
1-Propanol	<i>E. coli</i> W3110	<i>ΔlacI ΔysA ΔmetA ΔadhA ΔicdR ΔiblJH ΔibvBN AmpS thrA^{C1034T} lysC^{C1005T} phr::P tac PppC:Prc iibA^{C139T,G134T,C135G,T1352C} thrABC MJcmA ackA adhE^{mut}</i>	Introduction of a feedback resistant <i>iblA</i> , overexpressing the <i>cimA</i> gene encoding citramalate synthase and the <i>ackA</i> gene encoding acetate kinase A/propanate kinase I; introduction of a modified <i>adhE</i> gene encoding an aerobically functional <i>AdhE</i> , and deletion the <i>tpoS</i> gene encoding the stationary phase sigma factor	Glucose, glycerol	Semi defined	10.8	Bioreactor	Aerotolerant <i>AdhE</i> ; 20 g/L initial glucose; no feeding	[99]
Propanol	<i>B. subtilis</i> BSUDL-03	–	Introduction of PalSD-controlled auto-inducible 2-ketoisovalerate biosynthetic pathway and a <i>Pspac</i> -controlled IPTG-inducible Ehrlich pathways in <i>Bacillus subtilis</i>	Glucose	Defined	3.83 ± 0.46	fed-batch Pro fermentation system (DASGIP)	–	[93]
Isopropanol	<i>E. coli</i> ATCC 11303	None; <i>lacIQ CatIII atoDA Caaddc adhB-593</i>	Production of synthetic pathway by combination of different genes from <i>C. acetobutylicum</i> thi (acetyl-coenzyme A [CoA] acetyltransferase), <i>E. coli</i> atoAD (acetoacetyl-CoA transferase), <i>C. acetobutylicum</i> ade (acetoacetate decarboxylase), and <i>C. beijerinckii</i> adh (secondary alcohol dehydrogenase)	Glucose	Semi defined	5	Baffled shake flask	Yield of 0.15 g/g; acetone accumulated on depletion of glucose	[100]
Isopropanol	<i>E. coli</i> ATCC 11303	None; <i>lacIQ CatIII atoDA Caaddc adhB-593</i>	Use of <i>E. coli</i> TA76 and optimization of fermentation conditions and isopropanol removal by gas stripping	Glucose	Semi defined	143	Stirred flask	Yield of 0.23 g/g; gas stripping; 240 h	[101]
Isopropanol	<i>E. coli</i> ATCC 11303	None; <i>CatIII atoDA Caaddc adhB-593 Tfbgl-blc</i> (fused)	Firstly, use of genetically-engineered <i>E. coli</i> to produce isopropanol directly from cellobiose via cellobiose degradation by β -glucosidase (BGL) on the cell surface. Then, introduction of the synthetic pathway for isopropanol production into the strains and comparison of their isopropanol production in the presence of glucose	Cellobiose	Semi defined	4.1	Shake flask	Yield of 0.08 g/g	[102]
Isopropanol	<i>E. coli</i> JM109	None; <i>CatIIFAB Caaddc adhB-593</i>	<i>E. coli</i> JM109 Harboring isopropanol-producing pathway consisting of five genes encoding four enzymes, thiolase, coenzyme A (CoA) transferase, and acetoacetate decarboxylase from <i>C. acetobutylicum</i> ATCC 824, and primary-secondary alcohol dehydrogenase from <i>C. beijerinckii</i> NRRL B593	Glucose	Complex	13.6	Baffled shake flask	Yield of 0.17 g/g; acetone yield of 0.03 g/g	[103]

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Table 2 (continued)

Product	Host	Genotype (knockout; overexpression)	Metabolic engineering approach	Substrate	Medium	Titer (g/L)	Cultivation	Comment	Reference
Mixed alcohols	<i>C. acetobutylicum</i> ATCC 824	$\Delta buk:ermC; adc\ ctFA\ adh_{B-593}$	Primary/secondary alcohol dehydrogenase gene from <i>C. beijerinckii</i> NRRL B-593 (i.e., <i>adhB-593</i>) introduced in <i>C. acetobutylicum</i> ATCC 824	Glucose	Semi defined	20.4	Bioreactor	Yield of 0.30 g/g; isopropanol 4.4 g/L, butanol 14.1 g/L. With gas stripping	[111]
Mixed alcohols	<i>C. acetobutylicum</i> ATCC 824 $\Delta buk:ermC; adc\ ctFA\ adh_{B-593}$	$\Delta buk:ermC; adc\ ctFA\ adh_{B-593}$	A synthetic acetone operon (act operon; <i>adc-ctFA-ctFB</i>) used to increase flux to isopropanol	Glucose	Semi defined	35	Bioreactor	Yield of 0.26 g/g; isopropanol 4.1 g/L, butanol 25.1 g/L. With gas stripping	[111]
Mixed alcohols	<i>C. acetobutylicum</i> ATCC 824	$\Delta buk\ \Delta CA_C1502; adc\ ctFA\ adh_{B-593}$	Different synthetic isopropanol operons were constructed and introduced on plasmids in a butyrate minus mutant strain (<i>C. acetobutylicum</i> ATCC 824 $\Delta act15\Delta ppA\Delta buk$)	Glucose	Semi defined	20.4	Bioreactor	Yield of 0.33 g/g; with gas stripping	[104]
Mixed alcohols	<i>C. acetobutylicum</i> Rh8	None; <i>adh_{B-593}</i>	Overexpression of a single secondary alcohol dehydrogenase from <i>C. beijerinckii</i> NRRL B-593 in strain Rh8 under the control of thl promoter	Glucose	Semi defined	23.9	Bioreactor	Random mutagenized strain; yield of 0.31 g/g; isopropanol, 7.6 g/L; butanol, 15 g/L	[105]
Mixed alcohols	<i>C. acetobutylicum</i> BKM19	$\Delta buk:ermC; adh_{B-593} hydG_B-593$	Developed a hyper ABE producing BKM19 strain which converted acetone to isopropanol.	Glucose	Semi defined	28.5	Bioreactor	Pilot-scale fermentation; random mutagenized strain; yield of 0.37 g/g; isopropanol, 3.5 g/L; butanol, 15.4 g/L; ethanol, 9.6 g/L	[70]
1-Butanol	<i>C. acetobutylicum</i> ATCC 824	$\Delta pta\ \Delta buk; adhE^{D485G}$	Transformation of BKM19 strain with the plasmid pPA100 containing the <i>sadh</i> (primary/secondary alcohol dehydrogenase) and <i>hydC</i> (putative electron transfer protein) genes from the <i>C. beijerinckii</i> NRRL B-593 cloned under the control of the thiolase promoter	Glucose	Semi defined	18.9	Bioreactor	Simultaneous disruption of the <i>pta</i> and <i>buk</i> genes, encoding phosphotransacetylase and butyrate kinase; overexpression of the <i>adhE/D485G</i> gene, encoding a mutated aldehyde/alcohol dehydrogenase	[106]
1-Butanol	<i>C. acetobutylicum</i> ATCC 824	$\Delta pta\ \Delta buk; adhE^{D485G}$	Simultaneous disruption of the <i>pta</i> and <i>buk</i> genes, encoding phosphotransacetylase and butyrate kinase, overexpression of the <i>adhE/D485G</i> gene, encoding a mutated aldehyde/alcohol dehydrogenase	Glucose	Semi defined	130	Bioreactor	Volumetric productivity of 1.32 g/L h; with <i>in situ</i> recovery; yield of 0.31 g/g	[106]
1-Butanol	<i>C. acetobutylicum</i> ATCC 824	$\Delta adc; Ecgsh\ adhE\ ctFA\ tbh\ bcd\ ort\ bcd$	Inactivation of <i>adc</i> gene to eliminate the production of acetone; expression of <i>gshAB</i> genes in <i>adc</i> locus to improve the strain's robustness	Glucose	Semi defined	14.9	Bioreactor	Yield of 0.34 g/g; 3.3 g/L ethanol	[107]
1-Butanol	<i>C. tyrobutyricum</i> ATCC 25755	$\Delta ack; CaadhE2$	Overexpression of aldehyde/alcohol dehydrogenase 2 (<i>adhE2</i> , Genebank no. AF321779) from <i>C. acetobutylicum</i> ATCC 824, which converted butyryl-CoA to butanol, under the control of native thiolase (<i>tbh</i>) promoter	Glucose	Semi defined	10	Anaerobic bottle	Yield of 0.27 g/g; 5.8 g/L butyrate; manual control of pH with NaOH	[108]
1-Butanol	<i>C. tyrobutyricum</i> ATCC 25755	$\Delta ack; CaadhE2$	As above	Mannitol	Semi defined	16	Anaerobic bottle	Yield of 0.31 g/g; 1.0 g/L butyrate; manual control of pH with NaOH	[109]

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Table 2 (continued)

Product	Host	Genotype (knockout; overexpression)	Metabolic engineering approach	Substrate	Medium	Titer (g/L)	Cultivation	Comment	Reference
1-Butanol	<i>C. tyrobutyricum</i> ATCC 25755	None; <i>CadAhd22</i>	Expression of <i>adhE22</i> gene in <i>C. tyrobutyricum</i> under the control of a native thiolase promoter using four different conjugative plasmids (pMT182151, 83151, 84151, and 85151) each with a different replicon (<i>pBP1</i> from <i>C. butylicum</i> NCTC2916, <i>pCB102</i> from <i>C. butyricum</i> , <i>pCD6</i> from <i>C. difficile</i> , and <i>pM13</i> from <i>Bacillus subtilis</i>)	Mannitol	Complex	20.5	Anaerobic bioreactor	Yield of 0.332 g/g; productivity of 0.32 g/(L h); 1 g/L butyrate; manual control of pH with NaOH	[109]
1-Butanol	<i>E. coli</i> DH1	None; <i>RephaAB AcaphaJ Tdtter CaadH22 aceEF lpd</i>	Construction of two plasmid system as first, synthetic genes encoding PhaA, PhaB and Crt as a single operon driven by the relatively weak arabinose promoter, and second, plasmid containing an operon comprising <i>ccr</i> and <i>adhE2</i> using a strong T7lac promoter (<i>pET-ccr.adhE2</i>)	Glucose	Complex	3.4	Shake flask	Anaerobic after induction	[110]
1-Butanol	<i>E. coli</i> DH1	None; <i>RephaA Cahbd Caert Tdtter CaadH22 aceEF lpd</i>	As above	Glucose	Complex	4.7	Shake flask	Yield of 0.28 g/g; shift to anaerobic conditions after induction	[110]
1-Butanol	<i>E. coli</i> BW25113/F'	<i>ΔaldhA ΔadhE ΔfrdBC Δpta; atoB Caahbd Caert CaadH22 Cbfhd Tdtter</i>	Construction of a modified clostridial 1-butanol pathway in <i>E. coli</i> to provide an irreversible reaction catalyzed by trans-enoyl-coenzyme A (CoA) reductase (Ter) and created NADH and acetyl-CoA driving forces to direct the flux	Glucose	Complex	15	Anaerobic bioreactor	Yield of 0.36 g/g; no gas stripping	[111]
1-Butanol	<i>E. coli</i> BW25113/F'	<i>ΔaldhA ΔadhE ΔfrdBC Δpta; atoBCaahbd Caert CaadH22 Cbfhd Tdtter</i>	As above	Glucose	Complex	30	Anaerobic bioreactor without gas stripping	Yield of 0.36 g/g; without gas stripping	[111]
1-Butanol	<i>E. coli</i> MG1655	<i>fadR_crp* ΔarcA ΔadhE ΔptaΔfrdA ΔyqfHD; atoC_yqeF fucO</i>	Introduction of <i>fadR</i> and <i>atoC(c)</i> mutations in <i>E. coli</i> ; deletion of <i>arcA</i> gene to prevent ArcA-mediated repression of most operons encoding the β-oxidation cycle	Glucose	Defined	14	Baffled shake flask		[112]
Isobutanol	<i>E. coli</i> BW25113	<i>ΔadhE ΔaldA ΔfrdBC Δpta Δpta ApfB; BsalsS ilvCD LlkivD ScADH2</i>	Overexpression of five KDCs gene (<i>Pde6 Aro10 Thi3</i>) from <i>S. cerevisiae</i> , <i>Kld</i> from <i>Lactococcus lactis</i> 18, and <i>Pde</i> from <i>C. acetobutylicum</i>) with alcohol dehydrogenase 2 (Adh2) of <i>S. cerevisiae</i> 19	Glucose	Semi defined	22	Shake flask	Yield of 0.35 g/g	[74]
Isobutanol	<i>B. subtilis</i>	<i>Δldh; CglbVCD alsS LlkivDScADH2</i>	Introduction of an efficient heterologous Ehrlich pathway controlled by the promoter P43 into <i>B. subtilis</i> for the isobutanol biosynthesis	Glucose	Complex	2.62	Bioreactor		[113]
Isobutanol	<i>B. subtilis</i>	<i>Δldh; CglbVCD alsS LlkivDScADH2</i>	Construction of a P alsSD-controlled auto-inducible 2-ketosovalerate biosynthetic pathway and a P spac-controlled IPTG-inducible Ehrlich pathway in <i>B. subtilis</i> to modulate gene expression	Glucose	Complex	3.83	Bioreactor	Auto-inducible 2-ketovaleate synthetic operon	[93]

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Table 2 (continued)

Product	Host	Genotype (knockout; overexpression)	Metabolic engineering approach	Substrate	Medium	Titer (g/L)	Cultivation	Comment	Reference
Isobutanol	<i>C. cellulolyticum</i> ATCC 35319	None; <i>LikvD EcyqHD BsalS EcIvcD</i>	Cloning of the genes encoding <i>B. subtilis</i> -acetoacetate synthase, <i>E. coli</i> acetohydroxyacid isomerase/ductase, <i>E. coli</i> dihydroxy acid dehydratase, <i>L. lactis</i> ketoacid decarboxylase, and <i>E. coli</i> and <i>L. lactis</i> alcohol dehydrogenases into a pAT187 derivative plasmid under the control of constitutive ferredoxin (Fd) promoter from <i>C. pasteurianum</i>	Cellulose (Sigmacell type 50)	Defined	0.66	Not specified	7–9 d; strong expression of <i>alsS</i> may have been deleterious	[114]
Isobutanol	<i>C. glutamicum</i> ATCC 13032	$\Delta pyc\ \Delta ldhA; BsalS\ LikvD\ iivCD\ adhA$	Overexpression of <i>alsS</i> of <i>B. subtilis</i> , <i>iibC</i> and <i>iibD</i> of <i>C. glutamicum</i> , kind of <i>L. lactis</i> , and a native alcohol dehydrogenase (<i>adhA</i>)	Glucose	Complex	4.9	Shake flask	Yield of 0.09 g/g	[92]
Isobutanol	<i>C. glutamicum</i> ATCC 13032	$\Delta aceE\ \Delta pgo\ \Delta iibE\ \Delta ldhA\ \Delta mdh; iivBNCD\ EcprtAB\ LikvD\ adhA$	Inactivation of the pyruvate dehydrogenase (<i>adhA</i>) overexpression of the pyruvate transaminase B, and additional overexpression of the <i>iibBNCD</i> genes, encoding acetohydroxyacid synthase, acetohydroxyacid isomerase/ductase, and dihydroxyacid dehydratase in <i>C. glutamicum</i>	Glucose	Semi defined	13	Bioreactor	Yield of 0.20 g/g; productivity of 0.32 g/(L h); shifted to anaerobic conditions	[91]
Isobutanol	<i>R. eutropha</i> H16	$\Delta phaCAB\ \Delta iibE\ \Delta bkdAB\ \Delta aceE; adh(con)\ iibBHCD\ LikvD$	Overexpression of plasmid-born, native branched-chain amino acid biosynthesis pathway genes in combination with various mutant strains of <i>R. eutropha</i> with isobutyraldehyde dehydrogenase activity. Overexpression of heterologous ketosovalerate decarboxylase gene for biosynthesis of isobutanol and 3-methyl-1-butanol	Fructose	Defined	0.27	Shake flask	Coproduced 40 mg/L 3-methyl-1-butanol	[109]
Isobutanol	<i>E. coli</i> DH1	<i>fadE; tesA fadD AcaR1</i>	Elimination of the first two competing enzymes associated with β -oxidation, <i>FadD</i> and <i>FadE</i>	Glucose	Defined	0.06	Baffled shake flask		[115]
Isobutanol	<i>E. coli</i> MG1655	$\Delta araBAD\ \Delta fadAB\ \Delta ackA\ AtpA; UcfatB\ fadD\ Maacr2$ (Maqu_2507)	Overexpression of an acyl-ACP thioesterase (BTE), an acyl-CoA ligase (FadD), and an acyl-CoA/ aldehyde reductase (MAACR) genes in an engineered strain of <i>E. coli</i> to produce a high level of 1-dodecanol and 1-tetradecanol	Glucose	Defined	1.65	Bioreactor	Yield of 0.134 g/g mainly C12 and C14 alcohols	[116]
Isobutanol	<i>E. coli</i> BL21 (DE3)	$\Delta fadE; fadD\ Maacr2$ (Maqu_2220) <i>'tesA</i>	Introduction of two fatty acyl-CoA reductases encoded by <i>Maqu_2220</i> and <i>Maqu_2507</i> genes from <i>Marinobacter aquaeolei</i> VT8 into <i>E. coli</i>	Glucose	Defined	1.73	Bioreactor	Yield of 28.3 mg/g	[117]
Isobutanol	<i>E. coli</i> BL21 (DE3)	$\Delta fadE; Searr$	Introduction of fatty acyl-ACP reductase gene of <i>Synechococcus elongatus</i> to an <i>E. coli</i> strain	Glycerol	Defined	0.75	Bioreactor	Yield of 0.022 g/g	[118]

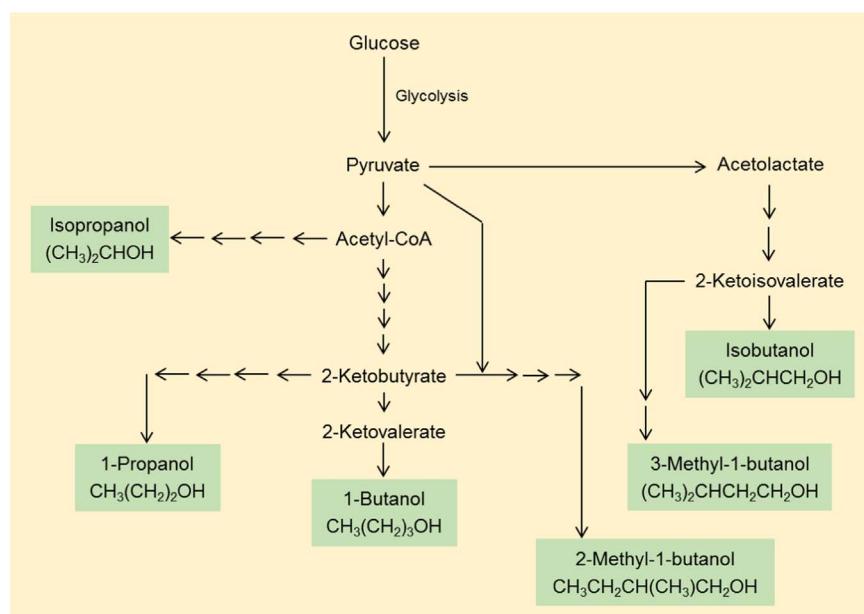


Fig. 2. Metabolic pathways for production of some higher alcohols. Based on Choi et al. (2014) [97].

1.82 mol of hydrogen could be produced from 1.08 mol of glucose.

In another study, a combinatorial engineering of the FHL complex and carbon metabolism was implemented in *E. coli* to obtain improved hydrogen yield [137]. Disruption of the *hycA* gene enabled a 2-fold improvement in FHL activity. Subsequent deletion of two uptake hydrogenases (1, *hya*; and 2, *hyb*) improved hydrogen yield from 1.2 to 1.48 mol/mol glucose confirming the critical impact of uptake hydrogenases on anaerobic fermentation of glucose. Disruption of multiple genes (*ldhA*, *frdAB*, *hycA*, *hya*, and *hyb*) further improved hydrogen yield on glucose to 1.8 mol/mol despite a high partial pressure of hydrogen in the culture system. With a low partial pressure of hydrogen, the same multiple mutated strain produced 2.11 mol H₂/mol glucose [137]. This was a little higher than the theoretical maximum yield attainable with *E. coli* strains relying on the formate-dependent pathway.

Following a similar approach, hydrogen yield in *E. coli* BW25113 *ΔhyaB ΔhybC ΔhycA ΔfdoG/pCA24 N-FhlA* was improved from 0.47 to 0.70 mol H₂/mol of glucose through overexpression of the FHL complex and inactivation of the uptake hydrogenases [138]. Seol et al. (2011) further produced genetically engineered recombinant *E. coli* SH5 with multiple deletion mutations in uptake hydrogenases (*hya* and *hyb*), *hycA*, and the carbon metabolic pathways (*ldhA* and *frdAB*) and claimed a hydrogen productivity of up to 2.4 L H₂/L/h [138].

A low yield of hydrogen on the carbon substrate is a general problem of fermentative hydrogen production. Co-production of ethanol and hydrogen has been proposed to overcome this. Pursuing this approach, Seol et al. developed *E. coli* BW25113 *ΔhycA ΔhyaAB ΔhybBC ΔldhA ΔfrdAB Δpta-ackA ΔpfkA* (designated as SH*_{ZG}) overexpressing *zwf* and *gnd* encoding the key enzymes in the pentose-phosphate (PP)

Table 3
Examples of metabolic engineering for biohydrogen production in bacteria.

Strain	Metabolic engineering approach	Yield (mol H ₂ /mol glucose)	Rate (mmol H ₂ /(g cell h))	Reference
<i>E. coli</i> W3110	Wildtype	ND	~ 90	[135]
<i>E. coli</i> W3110 <i>ΔhycA/fhlA</i>	Inactivated FHL repressor and overexpressed FHL	ND	~ 260	[135]
<i>E. coli</i> W3110	Wildtype	1.08	9.5	[135]
<i>E. coli</i> W3110 <i>ΔldhA ΔfrdBC</i>	Inactivated lactate dehydrogenase and fumarate dehydrogenase	1.82	13.4	[135]
<i>E. coli</i> BW25113	Wildtype	1.20	30.6	[136,137]
<i>E. coli</i> SH1 (BW25113 <i>ΔhycA</i>)	Inactivated negative regulator for FHL	1.17	58.2 ^a	[137]
<i>E. coli</i> SH2 (SH1 <i>ΔhyaAB</i>)	Inactivated uptake hydrogenase 1	1.37	46.2 ^a	[137]
<i>E. coli</i> SH3 (SH2 <i>ΔhybBC</i>)	Inactivated uptake hydrogenase 2	1.48	49.2 ^a	[137]
<i>E. coli</i> SH4 (SH3 <i>ΔldhA</i>)	Inactivated lactate dehydrogenase	1.61	45.6 ^a	[137]
<i>E. coli</i> SH5 (SH4 <i>ΔfrdAB</i>)	Inactivated fumarate reductase	1.80/2.11 ^b	52.2 ^a	[137]
<i>E. coli</i> BW25113/pCA24N	Wildtype containing empty vector (pCA24N)	0.47	3.7 ^c	[138]
<i>E. coli</i> BW25113 <i>ΔhyaB ΔhybC ΔhycA ΔfdoG/pCA24NFhlA</i>	Inactivated uptake hydrogenases and overexpress FHL complex	0.70	12 ^c	[138]
<i>E. coli</i> BL21(DE3) pA pYdbK	Wildtype harboring <i>E. coli</i> ydbK encoding a putative pyruvate: flavodoxin/ferredoxin oxidoreductase	0.67	ND	[141]
<i>E. coli</i> BL21(DE3) ΔiscR pAF pYdbK	Inactivated negative transcriptional regulator of the isc operon and expressed Fd-dependent hydrogenase (hydA) of <i>C. acetobutylicum</i> , maturation factors of hydA, and [4Fe-4 S]-ferredoxin of <i>C. pasteurianum</i>	1.88	ND	[141]
<i>Clostridium paraputreficum</i> M-21	Overexpressed [FeFe]-hydrogenase (HydA)	2.4	ND	[142]
<i>Clostridium tyrobutyricum</i> JM1	Overexpressed hydrogenase (HydA)	1.8	ND	[143]
<i>Thermococcus kadaikaraensis</i> KOD1	Gene cloning and knockout gene	3.3	ND	[144]

ND, Not determined.

^a Measured using formate as a substrate.

^b Measured under low H₂ partial pressure.

^c Measurement based on protein (mmol H₂/(g protein h)).

pathway [139]. Although the resulting strain successfully co-produced ethanol (1.38 mol/mol) and hydrogen (1.32 mol/mol) from glucose, without acetate, the generation of pyruvate at a high rate of 0.18 mol/mol apparently reduced the efficiency of the co-production [139]. Subsequently, Sekar et al. (2016) attempted to enhance the co-production rate of the strain SH^{*}_ZG by eliminating pyruvate accumulation and overexpressing the *zwf* and *gnd* genes. They designated the new strain as SH9^{*}_ZG. It co-produced H₂ and ethanol at 1.88 and 1.40 mol/mol, respectively, under anaerobic conditions [140].

2.1.3.2. Thermophilic bacteria. Success with metabolically-engineered mesophiles has led to renewed interest in engineering of thermophilic bacteria for hydrogen production [145–147]. The most common thermophilic hydrogen producers include *Pyrococcus furiosus*, *Thermococcus kodakaraensis*, and species of the genera Clostridium and Thermoanaerobacterium and the families Thermotoga and Caldicellulosiruptor [146,148–154].

For example, overexpression of [Fe]-hydrogenase (HydA) in *Clostridium paraputrificum* M-21 improved hydrogen yield on glucose from 1.4 to 2.4 mol/mol [151]. Similarly in a different investigation, overexpression of HydA in *C. tyrobutyricum* JM1 increased hydrogen yield on glucose from 1.2 to 1.8 mol/mol [141]. Liu et al. inactivated the *ack* gene encoding acetate kinase of *C. tyrobutyricum* and achieved a nearly 1.5-fold increase in hydrogen yield compared to the wildtype strain [143].

The above achievements notwithstanding, use of metabolic engineering requires an in-depth understanding of the often complex metabolic pathways and regulatory circuits involved in hydrogen production [123].

2.2. Yeasts

Yeasts are promising hosts for production of industrial products and have been widely used for commercial production of recombinant enzymes and proteins [155]. The relatively large spheroidal cells of yeasts often simplify bioprocessing relative to many bacteria and mycelial fungi. Moreover, they can be grown at ambient temperature and have simple nutritional requirements. On the other hand, the ability of yeasts to thrive at low pH values helps in minimizing contamination at large scale industrial operations where rigorous sterilization of the media and bioreactor may not be feasible [156]. Unlike many bacteria, yeasts are relatively resistant to potential phage contaminations [156]. Capacity of yeasts for sexual reproduction provides opportunities for improving them through cellular engineering [157].

In view of the abovementioned characteristics, yeasts have attracted much attention as cell factories for production of biofuels, in particular bioethanol. The yeast *S. cerevisiae* has attracted special interest as it is perhaps the best known yeast with a long history of industrial use [156,157]. *S. cerevisiae* is generally recognized as safe (GRAS) and many recombinant technologies have been developed for it [158]. Much of the sugarcane derived bioethanol in Brazil [51] and the corn starch hydrolysate-derived product in the United States [52] depends on yeasts for fermentation.

Several recent reviews have provided an overview of the many metabolic engineering schemes that have been used in *S. cerevisiae* [159–165]. Some of the key developments relating to production of biofuels in *S. cerevisiae* are summarized in Table 4. Surface-engineered yeasts used for biofuels production are discussed in a separate section.

2.2.1. Bioethanol

2.2.1.1. *S. cerevisiae*. The well-known yeast *S. cerevisiae* has been the mainstay of bioethanol production from sugars from the earliest times [174]. Compared to wildtype bacteria such as *E. coli*, yeasts better tolerate relatively high levels of alcohols although certain bacteria are also highly alcohol-tolerant. In view of its potential, much information exists on the biology, metabolism, and genetics of *S. cerevisiae*. This

information has been used to improve strains for production of fuel ethanol. As a result, ethanol yield on sugars now often approaches the theoretical limit of 0.51 g ethanol/g of glucose. Up to 93% of the glucose used may be converted to ethanol while the remainder is mostly converted to cell biomass [175].

Native *S. cerevisiae* can effectively use sugars but cannot use sugar polymers such as starch and cellulose to produce ethanol. Therefore, a prior hydrolysis of the sugar polymers is essential prior to use in yeast fermentation. However, this prior hydrolysis step adds to the cost of processing while a competitive fuel must essentially be extremely cheap. This has led to successful efforts to engineer a hydrolytic capability in *S. cerevisiae* so that polymeric substrates could be used directly [176]. *S. cerevisiae* engineered to surface display minicellulosomes has allowed simultaneous saccharification and fermentation of crystalline cellulose leading to an ethanol titer of 1.4 g/L [177]. In other developments, *S. cerevisiae* was engineered to ferment sugars that it previously could not. For instance, Ha et al. reported an engineered strain capable of cofermenting mixtures of xylose and cellobiose with an ethanol productivity of nearly 0.65 g/L/h [178]. Engineering of the abilities to ferment new sugars and hydrolyze extracellular sugar polymers have not remained confined to *S. cerevisiae* and other microorganisms with these capabilities have also been developed.

A major by-product of ethanol fermentation is glycerol. Glycerol is required to maintain an osmotic balance in yeast cells. Deletion of *gpd1* and *gpd2*, encoding cytosolic NADH-dependent glycerol-3-phosphate dehydrogenases, has reduced or completely eliminated the production of glycerol and allowed more carbon to be channeled to ethanol synthesis; however, such double deletion strains do not grow well anaerobically [175]. Much efforts has focussed on overcoming this problem [179].

2.2.1.2. Native xylose-fermenting yeasts. Genetic and metabolic engineering of native xylose fermenting yeasts such as *Pichia stipites* [180] and *Spathaspora passalidarum* [181] hold much promise.

2.2.1.3. Thermotolerant yeasts. Use of thermotolerant yeasts offers important advantages in production of bioethanol [163,182–184] and these yeasts too can be further enhanced by metabolic engineering. The thermotolerant yeast *Kluyveromyces marxianus* ferments glucose to ethanol and can grow at nearly 50 °C. In addition, *K. marxianus* can use a variety of substrates such as cellobiose, xylose, xylitol, arabinose, glycerol, lactose, and inulin. Hu et al. claimed that in ethanol production from inulin hydrolysates, *K. marxianus* provided a significantly higher ethanol yield than native *S. cerevisiae* [185].

The thermotolerant yeast *P. kudriavzevii* was reported to ferment cassava hydrolysate to ethanol at 40 °C to achieve an ethanol concentration of 78.6 g/L in about 24 h [186]. Ethanol productivity of 3.28 g/L/h with a yield of 85.4% of theoretical was reported for this process [186]. As the knowledge available on some of the lesser-known yeasts improves, they could be metabolically-engineered for much improved ethanol production processes than available now.

2.2.1.4. Halophilic yeasts. Certain hydrolysates of sugar polymers are naturally high in salt and require suitably tolerant yeasts to ferment. For instance, hydrolysates of seaweed biomass contain high salt concentrations. A marine yeast (*Candida* sp.) has been shown to ferment polysaccharide-based seaweed hydrolysate to ethanol in the presence of 6.25–11.25% salt [187].

2.2.1.5. Other yeasts. The sugar alcohol glycerol is a byproduct of the processes used in biodiesel production. The yeast *Pachysolen tannophilus* was reportedly able to ferment glycerol [188]. A final ethanol titer of 17.5 g/L was reported in a medium supplemented with 5% (v/v) crude glycerol as the substrate. This corresponded to 56% of the theoretical yield [188].

Table 4
Some biofuels produced in engineered yeasts.

Strain	Product	Key enzymes expressed and pathway	Titer (g/L)	Substrate	References
<i>S. cerevisiae</i> TMB3057	Ethanol	XR-XDH strain overexpressing XKS1 and PPP Δgre3	50	Xylose	[166]
<i>S. cerevisiae</i> TMB3066	Ethanol	<i>Piromyces</i> XI strain overexpressing XKS1 and PPP Δgre3	50	Xylose	[167]
<i>S. cerevisiae</i> RWB202-AFX	Ethanol	<i>Piromyces</i> XI evolved isolate	20	Xylose	[168]
<i>S. cerevisiae</i> INVSc1/pRS406XKS/ pILSUT1/pWOXYLA	Ethanol	Orpinomyces XI strain overexpressing XKS1 and SUT1	50	Xylose	[169,170]
<i>S. cerevisiae</i> ADAP8	Ethanol	Evolved Orpinomyces XI strain overexpressing XKS1 and SUT1	20	Xylose	[170]
<i>S. cerevisiae</i> CEN.PK 2-1C	Isobutanol	Overexpression of genes in valine metabolism as ILV2, ILV3, ILV5, and BAT2	4.12×10^{-3}	Glucose	[171]
<i>S. cerevisiae</i> BY4741	Isobutanol	Elimination of genes encoding members of the pyruvate dehydrogenase complex such as LPD1, overexpression of enzymes responsible for transhydrogenase-like shunts such as pyruvate carboxylase, malate dehydrogenase, and malic enzyme	1.62	Glucose	[172]
<i>S. cerevisiae</i> S288C(MAT α)	Isobutanol	Mitochondrial expression of ILV, KIVD, ADH genes	635×10^{-3}	Glucose	[173]
<i>S. cerevisiae</i> BY4742	n-Butanol	n-Butanol biosynthetic pathway in which isozymes from a number of different organisms (<i>S. cerevisiae</i> , <i>E. coli</i> , <i>Clostridium beijerinckii</i> , and <i>Ralstonia eutropha</i>) were substituted for the clostridial enzymes	2.5×10^{-3}	Galactose	[173]

2.2.2. Higher alcohols

2.2.2.1. *S. cerevisiae*. As pointed out earlier, higher alcohols are more attractive as transport fuels compared with ethanol [189]. In this context biobutanol has generated the highest level of interest and microorganisms are being engineered for its production. A *S. cerevisiae* engineered to produce biobutanol from galactose achieved a titer of 2.5 mg/L [117]. In a study by Chen et al., overexpression of genes in the valine pathway of *S. cerevisiae* enabled isobutanol production [171]. Brat et al. further improved the ethanol titer to 0.63 g/L by relocating the valine biosynthetic enzymes into the cytosol [190]. In a different study, Kondo et al. genetically enhanced the Ehrlich pathway and altered the carbon flux in *S. cerevisiae* to achieve isobutanol titers as high as 0.143 g/L [191]. Compartmentalization of the synthesis pathway in yeast mitochondria enabled titers of 0.64 g/L for isobutanol together with 0.1 g/L of isopentanol and 0.12 g/L of 2-methyl-1-butanol [192]. All this suggests that yeasts can be successfully engineered to produce certain higher alcohols. Biobutanol production using metabolically engineered *S. cerevisiae* is already being commercialized by Gevo, Inc., in the United States. Sugars produced from starch or woody biomass by enzymatic hydrolysis are used as feedstock.

2.3. Cyanobacteria

Cyanobacteria are photosynthetic prokaryotic bacteria previously known as blue-green algae, but are actually not algae at all as microalgae are microscopic photosynthetic eukaryotes. Cyanobacteria and microalgae are sunlight-driven cell factories that convert carbon dioxide and water to sugars and eventually to diverse biochemicals [193]. Many types of fuels and fuel precursors can be produced from microbial biomass grown via photosynthesis (Fig. 3). As sunlight, carbon dioxide and water are often readily available, cyanobacteria and microalgae have attracted much attention for producing fuels and other products [194–200]. Continuing developments notwithstanding, there are many hurdles to commercialization of fuels from microalgae and cyanobacteria [201].

Cyanobacteria may have a higher photosynthetic efficiency compared to land plants [202]. Furthermore, cyanobacteria do not compete for land with terrestrial crops [203]. Being prokaryotes, cyanobacteria are structurally simpler than microalgae and unlike algae, can be genetically engineered relatively easily. In view of this, cyanobacteria are attracting attention as subjects for metabolic and genetic engineering [193]. More than 126 cyanobacterial genomes have been sequenced [204] and the necessary platforms for genetic manipulation of cyanobacteria have been established. Overall, the major foci of metabolic engineering of cyanobacteria include direct production, product

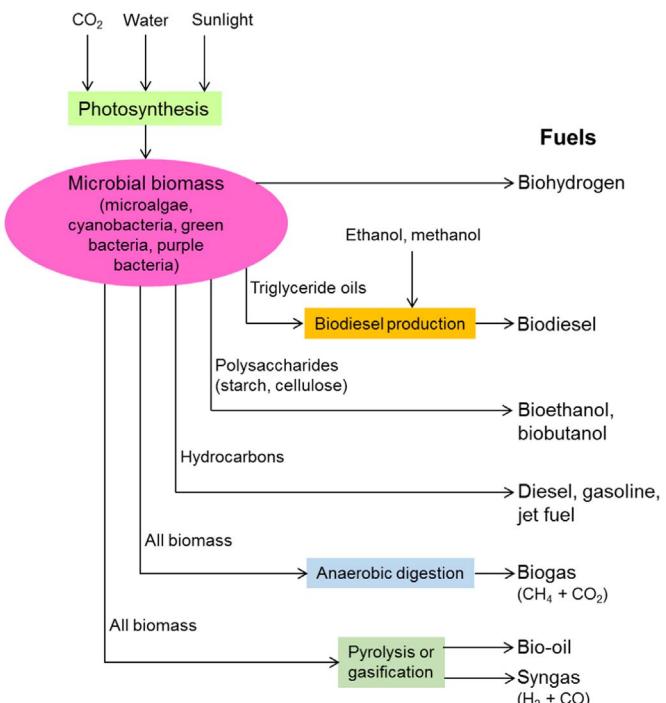


Fig. 3. Potential fuels and fuel precursors obtainable via microbial photosynthesis.

secretion, and process optimization [200]. Metabolic engineering aspects of cyanobacteria have been extensively reviewed [195,205,206]. Studies have spanned production of biohydrogen [169,171,207–209], ethanol [207,210–213], isobutyraldehyde [214], isoprene [213,215], β-caryophyllene [226], sucrose [217], butanol [218], fatty alcohols [219,220], and fatty acids [221,222]. Biofuels and other chemicals produced in metabolically-engineered cyanobacteria are summarized in Table 5.

2.3.1. Biohydrogen

Many studies have reported biohydrogen production using cyanobacteria [126,233–237]. The electrons released in the primary reactions of oxygenic photosynthesis can be directly channeled to hydrogen production [237]. More specifically, hydrogen production in cyanobacteria occurs via two pathways: 1) as a byproduct of nitrogen fixation by cyanobacterial nitrogenases; and 2) through the action of bidirectional (or reversible) hydrogenases [205]. Hydrogen production via nitrogenases requires ATP whereas bidirectional hydrogenases do not

Table 5

Titers of some biochemicals produced in engineered cyanobacteria.

Compound	Organism	Titer or productivity ^a	Manipulated gene(s)	Reference
Acetone	<i>Synechocystis</i> sp. PCC6803	36 mg/L	–	[223]
2,3-Butanediol	<i>Synechococcus elongatus</i> sp. PCC7942	2.4 g/L	–	[224]
1-Butanol	<i>S. elongatus</i> sp. PCC7942	30 mg/L	<i>ter, nphT7, bldh, yghD, phaJ, phaB</i>	[218]
1-Butanol	<i>Synechococcus</i> sp.	14.5 mg/L	<i>hbd, crt, adhE2, ter, atoB</i>	[225]
Ethanol	<i>Synechococcus</i> 7942	230 mg/L	<i>pdc, adh</i>	[210]
Ethanol	<i>Synechocystis</i> 6803	552 mg/L	<i>pdc, adh</i>	[211]
Ethanol	<i>Synechocystis</i> sp. PCC6803	5.5 g/L	<i>pdc, adh</i>	[226]
Ethylene	<i>Synechocystis</i> sp. PCC6803	171 mg/L/d	–	[227]
Fatty acids	<i>Synechocystis</i> sp. PCC6803	197 mg/L	<i>tesA, accBCDA, fatB1, fatB2, tesA137</i>	[221]
Fatty alcohol	<i>Synechocystis</i> 6803	200 ± 8 µg/L	<i>far</i>	[213]
Fatty alcohol	<i>Synechocystis</i> 6803	20 ± 2 µg/L/OD	<i>far, aas</i>	[226]
Fatty alcohol	<i>Synechocystis</i>	2.87 mg/g DCW	$\Delta sll0208, \Delta sll0209$	[220]
Isobutanol	<i>Synechococcus</i> 7942	18 mg/L	<i>kivd, yghD</i>	[214]
Isobutanol	<i>S. elongatus</i> sp. PCC7942	450 mg/L	<i>alsS, ilvC, ilvD, kivd, yghD</i>	[214]
Isobutyraldehyde	<i>S. elongatus</i> sp. PCC7942	1.1 g/L	<i>alsS, ilvC, ilvD, kivd, rbcls</i>	[214]
Isoprene	<i>Synechocystis</i> sp. PCC6803	50 mg/g DCW/d	–	[216]
2-Methyl-1-butanol	<i>S. elongatus</i> sp. PCC7942	2 mg/L	<i>kivd, yghD, cims</i>	[228]
Alkanes	<i>Synechocystis</i> 6803	162 ± 10 µg/L/OD	<i>accBCDA</i>	[219]
Alkanes	<i>Synechocystis</i>	2.3 mg/L/OD	<i>sll0208, sll0209</i>	[229]
Hydrogen	<i>Synechococcus</i> 7942	2.8 µmol/h/mg Chl-a	<i>hydEF, hydG, hydA</i>	[230]
Hydrogen	<i>Arthrosphaera maxima</i>	400 µmol/L/h		[231]
Methane	<i>A. maxima</i>	0.4 L/d		[232]

^a Chl-a, chlorophyll a; DCW, dry cell weight; OD, optical density.

require ATP and therefore, they are more efficient for hydrogen production [238]. Nevertheless, both these routes are challenging in practice for commercially-viable production despite the claimed advantages of cyanobacterial hydrogen production relative to the other microbial producers [237]. Although diverse cyanobacteria are reportedly able to produce hydrogen under various culture conditions, *Anabaena* spp. are particularly good producers (e.g. 68 µmol H₂/mg chl-a/h) [234].

Novel hydrogen producing cyanobacteria are being sought. Culture strategies as well as genetic and metabolic engineering of specific cyanobacteria to improve hydrogen production are receiving much attention [239–242]. The main challenges faced by cyanobacterial hydrogen production are the oxygen sensitivity of the hydrogenase enzymes involved in the process and the generally low hydrogen productivity as consequence of the competition for electrons by other pathways [208,242]. Oxygen-free operation can be difficult as photosynthesis naturally evolves oxygen. Thus, there is an inherent incompatibility between oxygenic photosynthesis and hydrogen production. Protein engineering has been attempted to reduce oxygen sensitivity of hydrogenases as well as to increase the photosynthetic efficiency by truncating the light harvesting antenna size [208].

In addition to the above-mentioned challenges, the availability of the reducing agents such as ferredoxin and NADPH is another bottleneck as these are needed also for other metabolic activities such as respiration [243]. Therefore, for enhancing hydrogen production, a part of the available electrons must be redirected to hydrogen producing enzymes and oxygen-tolerant hydrogenases must be engineered [205].

Both wildtype and engineered cyanobacteria are being extensively researched for hydrogen production. For instance, attempts are being made to eliminate pathways that consume reducing agents. Mutants of *Synechococcus* 7002 lacking lactate dehydrogenase have demonstrated 5-fold greater hydrogen production compared with the wildtype [242]. In a different investigation, Bandyopadhyay et al. described *Cyanothece* sp. ATCC 51142, a unicellular nitrogen-fixing cyanobacterium with the ability to generate high levels of hydrogen (465 µmol H₂/mg of chlorophyll h) in the presence of glycerol under aerobic conditions [244].

In addition to microbial engineering, novel bioprocesses involving temporal or spatial separation of aerobic photosynthesis and anoxic hydrogen production are being developed [245]. Notwithstanding

these advances, cyanobacterial hydrogen production requires further improvement to become economically viable.

2.3.2. Bioethanol

Starch produced by cyanobacteria may be hydrolyzed and fermented into bioethanol or longer chain alcohols using the same technology as employed with corn and cassava starch. Some cyanobacteria are naturally capable of fermenting the accumulated starch into ethanol and excreting the ethanol in the extracellular environment [243]. Ethanol production pathway in cyanobacteria involves an initial conversion of pyruvate to acetaldehyde through decarboxylation and this is followed by the reduction of the produced acetaldehyde to ethanol [193]. Ethanol can also be produced directly from carbon dioxide, water, and sunlight in metabolically-engineered cyanobacteria.

In general, dark fermentation (i.e. an absence of photosynthesis) is not a primary route of energy production in most cyanobacteria [243,246] and occurs at a limited rate [243]. Nevertheless, attempts have been made to increase the rate of natural dark fermentation through genetic modification. For instance, *Synechococcus* sp. PCC 7942 was modified by inclusion of the coding sequences for pyruvate decarboxylase and alcohol dehydrogenase II from *Z. mobilis* [210]. These genes were reportedly expressed under the control of the cyanobacterial rbcLS operon promoter alone and also in combination with the *E. coli* lac promoter and the ethanol productivity of the transformed strain reached 54 nmol per OD730 unit/L/d [209]. Ethanol productivity could be further increased/(mg chl h) when the CI-PL temperature inducible promoter was used [212]. In another study, the model cyanobacterium *Synechocystis* sp. PCC 6803 was transformed via a double homologous recombination system to integrate the pyruvate decarboxylase (*pdc*) and alcohol dehydrogenase II (*adh*) genes from *Z. mobilis* into the cyanobacterial chromosome under the control of the strong light-driven psbAII promoter [211]. The ethanol productivity achieved using this engineered cyanobacterium was 5.2 mmo per OD730 unit/L/d [211].

Increasing NADPH production was hypothesized to increase production of biomass and ethanol in cyanobacteria [247]. This hypothesis was proved by over expressing the endogenous *zwf* gene encoding glucose-6-phosphate dehydrogenase of pentose phosphate pathway in the model cyanobacterium *Synechocystis* sp. PCC 6803. Overexpression of *zwf* led to increased NADPH production and a consequent increase in production of biomass and ethanol compared with the wild type under both autotrophic and mixotrophic conditions [247]. In one study, the

freshwater cyanobacterium *Synechococcus elongatus* PCC 7942 was subjected to transformation by introducing the pyruvate decarboxylase (*pdc*) and alcohol dehydrogenase B (*adhB*) genes of *Z. mobilis* into its genome, but only one construct produced bioethanol at detectable levels [248].

2.3.3. Higher alcohols

Keto acid pathway offers a route to propanol and butanol. 2-Ketobutyrate is an intermediate in the biosynthesis of citramalate and citramalate synthase has been found in cyanobacteria [249,250]. Therefore, there is the possibility of producing propanol and butanol from 2-ketobutyrate in cyanobacteria. For instance, engineering of isobutanol biosynthetic pathway and overexpression of Rubisco in *Synechococcus elongatus* PCC 7942 enabled an isobutyratdehyde productivity of 6230 µg/L/h and an isobutanol productivity of 3000 µg/L/h [214]. In a different study, *Synechocystis* PCC 6803 was genetically modified with the *ispS* gene of *Pueraria montana* leading to the production of isoprene [216]. Isoprene, a potential biofuel, is a hydrocarbon with higher energy content than ethanol and is easier to separate from an aqueous broth compared with ethanol.

Some cyanobacteria naturally produce alkanes as well [250]. As pure hydrocarbons, alkanes are energy-rich potential fuels. This could open up new options for producing fuels in engineered cyanobacteria in the future.

2.4. Microalgae

Microalgae are photosynthetic cell factories capable of producing starch, oils, and other biochemicals [22]. Compared with higher plants, microalgae grow more rapidly as they utilize sunlight better. Therefore, microalgae have the potential to provide huge quantities of starch and oils for conversion to various types of fuels [251,252]. Microalgae with a modified keto acid pathway may be a promising platform for producing isobutanol as well [253,254]. Long chain alcohols (C5–C8) have also been successfully produced in microalgae by modifying the cells for overproduction of 2-keto acids, the intermediates of the amino acid biosynthesis [254,255]. These intermediates were later converted to butanol derivatives by the heterologous expression of 2-keto acid decarboxylase and alcohol dehydrogenase [71,254].

In general, microalgae have attracted the most attention as producers of oils for fuels [8,197]. Oil productivity of microalgae is far greater than that of oil palm, one of the most productive commercial oil crops [256–260]. Moreover, microalgae can be grown on land which is not suitable for agriculture using seawater and brackish water. Their numerous advantages notwithstanding, commercialization of oil production from algae faces difficult challenges that are unlikely to be overcome in the short-term [201]. The costs of production of the biomass and its downstream processing are high [259]. Hence, economic viability of large-scale production of microalgae-based biofuels is questionable. Metabolic and genetic engineering are important to a future commercialization of algal fuels. Extensive effort is underway in these areas. The genomes of more than 30 microalgae have already been sequenced [259].

2.4.1. Biodiesel

Microalgae often produce some of the same triglyceride oils found in oil crops such as canola and sunflower. As such, the triglyceride component of the algal oil may be readily converted to biodiesel. As mentioned earlier, under suitable conditions, the oil content of microalgae can be as high as 30–70% of their dry biomass and this has been the source of interest in biodiesel derived from microalgae [8,261–265].

Much of the existing effort on large-scale production of algal oils has focused on native algae. Nonetheless, attempts are being made to metabolically engineer algae for an enhanced production capability. The green alga *Chlamydomonas reinhardtii* has been the main focus of

genetic and metabolic engineering work simply because its genome is known and it has a long history of use in research as a model alga. Genes associated with biosynthesis and accumulation of triglyceride oil in *C. reinhardtii* have been isolated and characterized to some level [266,267]. Because triacylglycerols (TAGs) are derived from either acylation of diacylglycerol (DAG) via a de novo pathway (acyl CoA-dependent pathway) or recycling of membrane lipids (acyl CoA-independent pathway), most of the metabolic engineering strategies have focused on manipulating the rate-limiting steps of these two pathways hoping to increase the metabolic flux for TAG production. For example, Li et al. claimed a 10-fold increase in the production of TAGs in a starchless mutant of *Chlamydomonas* by the inactivation of ADP-glucose pyrophosphorylase [266]. The resulting mutant BAFJ5 lacked the small subunit of ADP-glucose pyrophosphorylase and accumulated up to 32.5% by weight of neutral lipids and 46.5% by weight of total lipids [268].

Many of the genes involved in lipid synthesis in microalgae have been subjected to both knockout and overexpression in order to understand their role and identify the most efficient strategies for increasing the lipid content of the biomass. The enzymes acetyl CoA carboxylase (ACCase) and type-II fatty acid synthase (FAS) have been reported to be the main rate-limiting enzymes in fatty acid synthesis pathway [259]. ACCase is involved in forming malonyl CoA from carboxylate acetyl CoA and FAS is involved in elongating fatty acid chains. Therefore, it may be possible to enhance lipid production rates by increasing the activity of these enzymes via genetic engineering [264]. In pursuing this approach, expression vectors containing the gene encoding ACCase have been constructed and a transformation protocol has been developed for overexpression of ACCase in *Cyclotella cryptica* and other algal species [269]. However, ACCase overexpression has not increased lipid content [270]. It may be that ACCase activity is a limiting step in lipid biosynthesis mainly in cells that normally do not store large amounts of lipids [254].

In an attempt to increase lipid production without affecting growth, the marine diatom *Thalassiosira pseudonana* was engineered by knockdown of a multifunctional lipase/phospholipase/acyltransferase [271]. The resultant strains had up to ~3-fold lipid content relative to the wildtype during exponential growth. Metabolic engineering has also been used to alter the lipid profile in the marine diatom *Phaeodactylum tricornutum* [272]. The engineered diatom produced TAG with increased levels of shorter chain length fatty acids (lauric acid, C12:0; myristic acid, C14:0) which are more desirable for biodiesel production [271].

The above improvements notwithstanding, economic production of microalgal biodiesel is unlikely in the near future.

2.4.2. Biohydrogen

Like cyanobacteria, purple non-sulfur photosynthetic bacteria, some dark fermenting microorganisms and microalgae can also be used to produce hydrogen [273–281]. Three pathways are known to exist for hydrogen production in microalgae (Fig. 4). Two of these pathways require light while the third functions in the dark.

Biohydrogen is appealing specially for electricity production by fuel cells [281]. In many cases hydrogen is naturally produced in microalgae as a secondary metabolite to balance the redox energetics [282,283]. Hydrogen production has been studied to various depths in *C. reinhardtii* [280,284,285], *Scenedesmus obliquus* [286–288], *Chlorella fusca* [281], *Chlorella* sp. [289,290], *Chlorella vulgaris* [291], *Chlorella protothecoides* [292], and *Platymonas subcordiformis* [293,294]. Although hydrogen is naturally produced in microalgae under suitable culture conditions and hydrogenase activity can be externally induced by adding DCMU (a PSII electron chain uncoupler), hydrogen production cannot be maintained together with active photosynthesis because oxygen inactivates hydrogenase [281]. Under suitable conditions, hydrogen production may be augmented by channeling the electron flux to hydrogenases so that protons are biocatalytically converted to

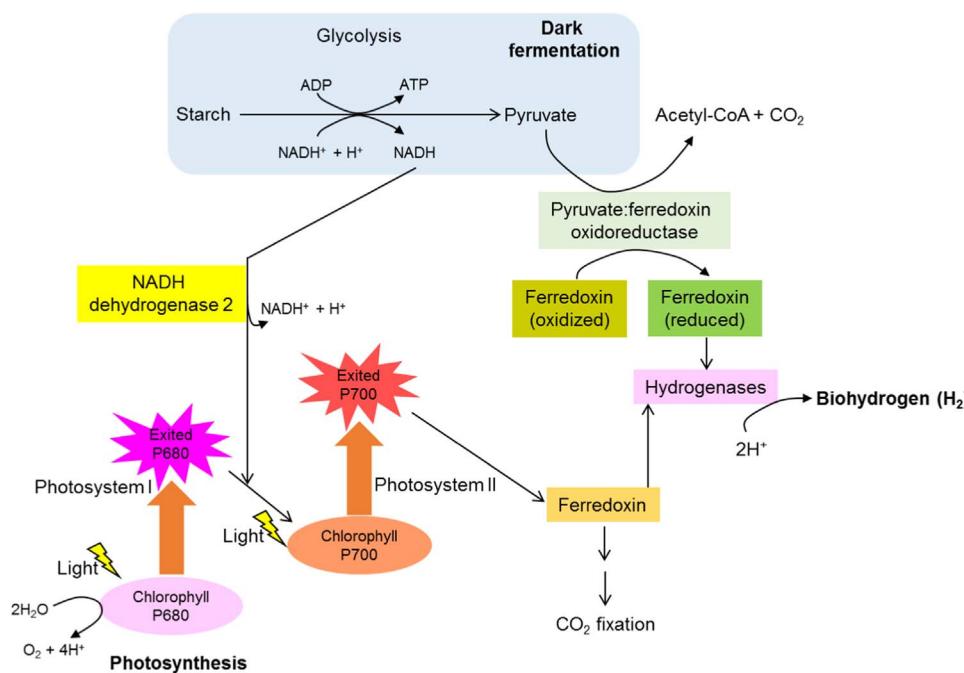


Fig. 4. The production of biohydrogen in microalgae. In the alga *Chlamydomonas reinhardtii* at least three pathways exist for hydrogen production. Two of these pathways involve light: (1) in a direct pathway, activity of the photosystem II supplies electrons to hydrogenases through the photosynthetic electron transport chain to drive production of hydrogen; and (2) in an indirect pathway, hydrogen is produced without the involvement of photosystem II through a process that depends on photosystem I. In the third pathway, the dark fermentation, glycolysis of starch drives hydrogen production. Based on Dubini and Gonzalez-Ballester (2016) [152]. ADP – adenosine diphosphate; ATP – adenosine triphosphate; NADH – nicotinamide adenine dinucleotide reduced form; NADH⁺ – nicotinamide adenine dinucleotide oxidized form.

hydrogen ($2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2$). To achieve this in *C. reinhardtii*, Kruse et al. blocked the cyclic electron transfer around photosystem I to reduce diversion of electrons away from hydrogenases. This improved hydrogen production rate by 5- to 13-fold relative to the control under a range of conditions [278]. In a different investigation, Rumpel et al. enhanced hydrogen production rate by 5-fold by redirecting electrons from photosystem I to hydrogenases [295]. Similarly, expression of an exogenous ferredoxin from *C. acetobutylicum* in addition to the native ferredoxin of *C. reinhardtii*, led to an enhanced electron flow to the hydrogenase HydA via siphoning of electrons from the fermentation of internal reducing equivalents (such as glycogen) [296]. This enhanced hydrogen production by nearly 2-fold under light-dependent anoxic conditions [296]. A ferredoxin-NADP⁺ reductase (FNR)-down-regulated mutant of *C. reinhardtii* had a 2.5-fold higher hydrogen production activity relative to the parent strain under sulfur deprivation [297]. Other similar approaches have also been reported [298].

3. Cell surface engineering for biofuel processes

Surface-engineered cells carry specific non-native proteins on their external surface. These surface proteins may extend from the plasma membrane through the cell wall, or they may be bound to the cell-surface components [16]. Surface displayed proteins are generally confined to particular domains on the surface of the cell.

The yeast *S. cerevisiae* has a rigid cell wall which is about 200 nm thick [299] and mainly consists of mannoproteins and β -linked glucans. The β -glucans rich layer occurs just above the plasma membrane [299]. Some of the proteins displayed on the surface of the yeast cells have secretion signal peptides at the N-terminal and glycosylphosphatidylinositol (GPI) anchors at the C-terminal. Protein synthesis occurs in the cell and the resulting proteins are then translocated into the lumen of the endoplasmic reticulum (ER) and on to the Golgi apparatus. Subsequently, membrane-enclosed vesicles carry the protein from the Golgi to the plasma membrane [300]. GPI-anchored proteins are further moved to the exterior of the plasma membrane via the secretory pathway. They are then released from the plasma membrane by the action of a phosphatidylinositol-specific phospholipase C (PI-PLC) and transferred to the outermost surface of the cell wall [299].

Either a single protein or a cocktail of proteins may be surface-displayed [301]. The displayed protein may be a hydrolytic enzyme, for

example, to breakdown a polymeric substrate such as lignocellulosic biomass into small molecules that can be taken up by the cell as substrate (Fig. 5). Multiple displayed proteins may act synergistically to carry out a certain extracellular transformation before the product of such a transformation is internalized by the cell [299,301].

3.1. Yeasts and other microorganisms

Engineered microbial cells armed with surface-displayed bioactive proteins are potentially powerful biocatalysts for production of biofuels and other products [302]. Surface display technologies were initially established for phages and bacteria [26,301] and have since been applied more broadly. Numerous proteins and peptides have been successfully displayed with retention of their bioactivity [16]. Simultaneous display of multiple types of molecules on the same cell has become possible to greatly expand the usefulness of the surface-engineered microorganisms [15]. Compared to most bacteria, yeasts offer a larger surface for display and are easier to recover from a broth after

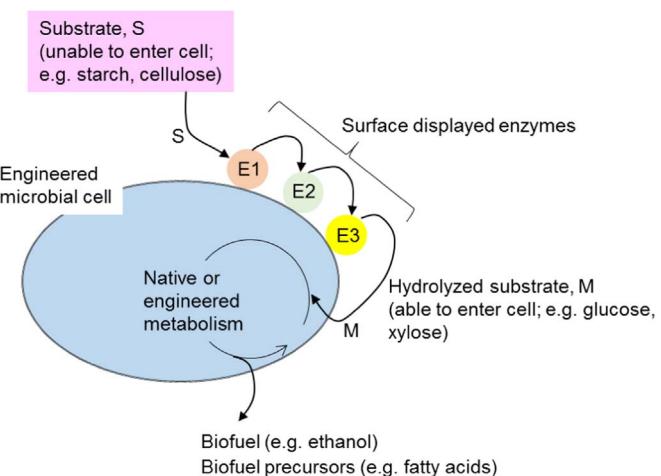


Fig. 5. Metabolically engineered microbial cell with surface displayed enzymes E1–E3. These enzymes extracellularly convert the substrate S to products (i.e., M) that can be taken up by the cell. These products, or modified substrates, then enter the metabolic pathways in the cell to eventually generate a biofuel or fuel precursor.

use.

Among yeasts, surface arming was first developed in *S. cerevisiae*. Subsequently, similar display systems were used to transform other yeasts, including *Pichia pastoris* [303] and *Yarrowia lipolytica* [304]. With suitably displayed enzymes capable of hydrolyzing extracellular sugar polymers to simple sugars, a yeast can use substrates that it could not previously use. Such surface displayed yeasts allow consolidation into a single operation the otherwise separate steps of polymer hydrolysis and sugar fermentation (Fig. 5). This kind of consolidated bioprocessing is certain to greatly improve the economics of conversion of lignocellulosic substrates to ethanol [305].

Direct production of ethanol from starch involving combined saccharification-fermentation steps and surface engineered yeast has been shown to work. The biocatalyst used was *S. cerevisiae* with exo glucoamylase (a starch-degrading enzyme) of *Rhizopus oryzae* displayed on its surface [306,307]. The production efficiency of ethanol from starch was further enhanced by co-displaying of exo glucoamylase and endo α-amylase from *Bacillus stearothermophilus* [308,309]. A surface engineered yeast expressing β-glucosidase and cellulase of *Aspergillus aculeatus* was shown to assimilate cellobiooligosaccharides [307]. Surface display of α-amylase of *Streptococcus bovis* on a glucoamylase displaying yeast was reported by Shigechi et al. (2004) [310]. The engineered yeast was shown to thrive and produce ethanol in a medium with starch as the sole carbon source.

Yeasts capable of directly fermenting cellulose require co-display of multiple enzymes including endoglucanases (EGs), cellobiohydrolases (CBHs), and β-glucosidases (BGLs) [311]. EGs act randomly on the amorphous region of the cellulose chain to produce reducing and nonreducing ends. CBHs are needed to release cellobiose from the exposed ends while BGLs hydrolyze the released oligosaccharides to glucose. A surface engineered yeast co-displaying endoglucanase II (EG II) and cellobiohydrolase II (CBH II) of *Trichoderma reesei* and BGL1 of *A. aculeatus* was shown to produce ethanol directly from cellulose that had been pre-swollen with phosphoric acid [312]. In the sake yeast *S. cerevisiae*, ethanol production from β-glucan was achieved by cell surface display of EG and BGL of *Aspergillus oryzae* [313].

In cells engineered to co-display multiple enzymes, an important consideration is the control of the relative ratios of the various enzyme activities displayed [192,302]. Attempts have been made to optimize the ratios of the surface displayed enzyme activities to improve synergism [314,315]. Although such optimization is based mostly on empiricism, it has been shown to improve ethanol production from cellulosic substrates such as newsprint [316].

Some clostridia produce cellulosome, a complex of scaffolding proteins and the multiple enzymes needed for hydrolysis of cellulose and hemicellulose [317]. This complex is formed through interaction of a cohesin module in the scaffolding proteins and a dockerin module in cellulosomal enzymes [318]. Tsai et al. engineered a yeast to display a mini-cellulosome [319]. The scaffolding proteins of this displayed cellulosome consisted of three cohesin domains derived from *Clostridium thermocellum*, *Clostridium cellulolyticum*, and *Ruminococcus flavefaciens*. The scaffolding proteins were displayed on the surface of *S. cerevisiae* using the α-agglutinin-based display system. The incubation of the scaffolding protein-displaying yeast with three recombinant cellulases (EG, CBH, and BGL) fused with a dockerin domain produced by *E. coli* led to a cellulosome displaying yeast able to synergistically hydrolyze cellulose [319].

A consortium of microorganisms displaying different proteins may also be co-cultured to achieve hydrolysis–fermentation of substrates such as cellulose. Goyal et al. developed a yeast consortium consisting of four engineered yeasts that were co-cultured [320]. In their experiments, one of the yeasts displayed cellulosome scaffolding protein and the other three yeasts secreted three different dockerin-fused enzymes. This resulted in the formation of a mini-cellulosome on the surface of the yeast displaying the scaffolding protein.

Hemicellulose, including xylan, also occurs in cellulosic biomass. In

addition to cellulose, the hemicellulose components of the biomass must also be effectively hydrolyzed to allow full utilization of the biomass for production of ethanol. Xylan is hydrolyzed to xylo-oligosaccharides by endo-β-xylanase and the xylo-oligosaccharides are further hydrolyzed to D-xylose by β-xylosidase. A *S. cerevisiae* co-displaying xylanase II (XYN II) of *Trichoderma reesei* and β-xylosidase (XylA) of *A. oryzae* was engineered using the α-agglutinin-based display system and shown to hydrolyze xylan to xylose [318]. In a further development, xylose reductase (XR) and xylitol dehydrogenase (XDH) of *Pichia stipitis* and xylulokinase (XK) of *S. cerevisiae* were produced in the yeast co-displaying XYN II and XylA. The resulting yeast simultaneously saccharified xylan and fermented xylose to ethanol [321,322].

The enzyme xylose isomerase catalyzes the two-step isomerization of xylose to xylulose. In a study, bioactive xylose isomerase of *C. cellulovorans* was surface-displayed on *S. cerevisiae*. The resulting yeast could grow in a medium containing xylose as the sole carbon source and produced ethanol directly from xylose [323].

Lignin is another constituent of lignocellulosic biomass. Lignin impedes the action of cellulases by masking the cellulose fibers [324]. Therefore, various biomass pretreatments are used to remove the lignin prior to enzymatically hydrolyzing the cellulosic residue in a second step [324,325]. The enzyme laccase from various white-rot fungi can degrade lignin. A *S. cerevisiae* was engineered via the α-agglutinin-based display system to surface display laccase I of *Trametes* sp. [325]. Hydrothermally-processed rice straw was treated with the laccase-displaying yeast and subsequently treated with a yeast co-displaying EG II, CBH II, and BGL1. This two-step treatment improved ethanol yield from rice straw [325].

4. Future outlook of metabolic engineering for biofuel production

Although metabolic engineering is a key enabling technology for transforming microorganisms into efficient cell factories [325–327], identifying suitable enzyme targets and metabolic pathways for modification or replacement are not easy. Successful metabolic engineering relies on methods that aid assembly and optimization of novel pathways in microbes. Diverse factors contribute to the performance of a metabolic pathway and problems linked to abundance of mRNA and enzyme activities may not be identified simply by monitoring product titers. Synthetic biologists and metabolic engineers rely on a variety of methods to identify the limiting parts of a pathway [328–333]. Bio-products not normally made by microorganisms in nature may be synthesized using emerging development in metabolic engineering [334–339]. Such systems metabolic approaches are contributing to development of biofuel producer species as summarized in Table 6. Emerging techniques include high-throughput sequencing for deciphering genomes [340,341]; highly sensitive transcriptomics, proteomics [342–344] and metabolomics methods; [345,346] and advances in fluxomics [347] and computational tools. These technologies are enhancing the potential of metabolic engineering to influence fuel production.

5. Concluding remarks

Metabolic engineering approaches are proving useful for production of biofuels and biofuel feedstocks in diverse microorganisms including bacteria, yeasts, cyanobacteria, and microalgae. Compared to the traditional random mutagenesis for strain improvement, the metabolic engineering is rational, rapid, and extremely powerful. More specifically, entire new pathways can be introduced into suitable microorganisms to enable production of biofuels and precursors of interest. Alternatively, parts of specific biochemical pathways may be engineered to optimize production of a biofuel. Commercialization of biofuels will inevitably require microbial cell factories that are substantially superior to wildtypes and some of the engineered strains now available. The surface display technology will allow microorganisms to

Table 6
System metabolic engineering approaches for the production of biofuels.

Strategy	Target	Strain	Achievement	Reference
Introduction of an engineered n-butanol pathway comprising genes of <i>atob</i> , <i>hbd</i> , <i>ert</i> , <i>bcd</i> , and <i>adhE2</i> along with comparative proteomics analysis	n-butanol	<i>C. acetobutylicum</i>	Yields of 40 and 120 mg/L n-butanol from cellobiose and crystalline cellulose, respectively	[348]
Transcriptome analysis and overexpression of YOR1, SNQ2, PDR5, and PDR15	Alkanes	<i>S. cerevisiae</i>	Intracellular C10 and C11 alkanes amounts were lowered by 33% and 94.4%, respectively	[349]
Heterologous expression of ABC2 and ABC3 transporters	Ethanol and butanol	<i>S. cerevisiae</i>	Tolerance to decane increased 80-fold	[350]
Quantitative proteomic analysis		<i>C. acetobutylicum</i> ATCC 824	Reduction of ethanol (0.47 g/L for cellobiose; 0.27 g/L for cellobiose plus lignin) and butanol production (0.13 g/L for cellobiose; 0.04 g/L for cellobiose plus lignin)	[351]
Heterologous expression of mevalonate pathway by principal component analysis of proteomics (PCAP) approach	Bisabolene	<i>E. coli</i>	The titer of isoprenoid-based Biofuels was increased 40%	[344]
Metabolomics and proteomics methods and engineering of the heterologous isoprenoid pathway	Isoprenoid-based C5 alcohols (3-methyl-3-butene-1-ol, 3-methyl-2-butene-1-ol, and 3-methyl-1-butanol)	<i>E. coli</i>	The titers of 3-methyl-3-butene-1-ol, 3-methyl-2-butene-1-ol and 3-methyl-1-butanol were increased by 2.23 g/L, 150 mg/L, and 300 mg/L, respectively.	[352]
Targeted proteomics, via selected-reaction monitoring (SRM) mass spectrometry	Sesquiterpene, amorpho-4:1:1-diene	<i>E. coli</i>	The final amorpho-4:1:1-diene titer was improved by over threefold (> 500 mg/L)	[353]
Integrated analysis of gene expression and metabolic fluxes	Medium-chain-length poly(3-hydroxyalkanoates) (incl- PHA)	<i>Pseudomonas putida</i> KT2440	A higher PHA content (29.7 wt% of CDW) was achieved at low dilution rate as compared with a high rate (12.8 wt% of CDW)	[354]

make use of substrates that they previously could not use. This capability will improve the economics of production of some biofuels. Genetic and metabolic engineering in combination with synthetic biology and systems biology are the key to generating highly capable cell factories for production of biofuels. Based on the present state of development of metabolically engineered microorganisms discussed in this review, the following can be inferred:

1. Commercial production of cellulases and other hydrolytic enzymes using engineered microorganisms is well established for producing feedstock sugars from polymeric substrates (lignocellulose, starch, xylans) for microbial fermentation to mainly bioethanol and bio-butanol. Production of commercial fuel alcohols relies heavily on these enzymes, but lignocellulosic ethanol remains relatively expensive as a separate process is required for producing the enzymes.
2. Yeasts engineered to ferment simple sugars to isobutanol instead of the conventional ethanol are already being commercialized. Sugars derived from enzymatic hydrolysis of starch, or woody biomass, are used as the fermentation feedstock.
3. Yeasts engineered to surface display hydrolytic enzymes appear to be highly promising and have good prospects for near-term commercial use in production of bioethanol directly from starch and pretreated cellulosic biomass.
4. The bacterium *Zymomonas mobilis* metabolically engineered to ferment the wood sugar xylose is being used in commercial processes for making bioethanol from lignocellulosic biomass.
5. Production of biogas, already well established commercially, will continue to rely on native microbial populations in the near term because of the complexities of engineering methanogenic bacteria.
6. Microalgae, cyanobacteria and other photosynthetic bacteria, whether native or engineered, do not seem to have a near term potential for generating commercializable biofuels.

In brief, genetically and metabolically engineered microorganisms are now having an impact on the commercial biofuels scene and this is likely to increase rapidly in the future.

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