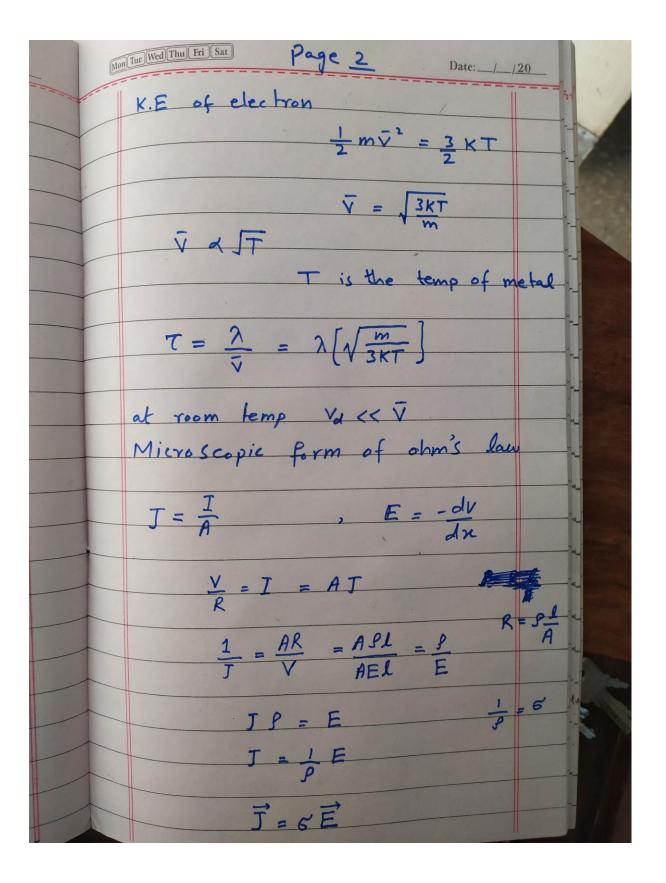


Mon Tue Wed Thu Fri Sat classical free electron theory mean free path = 2 mean collision time = T average speed = V according to drude electron gas model for unit volume of metal  $P = \frac{1}{3}P\bar{v}^2 = \frac{1}{3}mn\bar{v}^2$ here I is the density of electron For molas volume of metal P = 1 9 = 1 m Na v2 Vm = molar volume of metal PVm = 1 mNa + V2 = RT  $m\nabla^2 = \left[\frac{3RT}{N_0}\right] = 3K_8T$ 

Mon Tue Wed Thu Fri Sat properties of metals 1- at low temp g x Ts 2- at room temp and above g x - p = pressure 4- resistivity of an impure Specimen is given by Mathiessen's rule 9 = 90 + 9(T) 5 - Magneto resistance (magnetic field dependent resistivity)



MonTue Wed Thu Fri Sat Page 3 Date: _/_20	
when E field is applied in x-	en de la companya de
direction then all electron accelerate	
in x-direction,	
ain = acceleration of ith at electron	
in n-direction	-1-
main = -eEn	-1
$a_{in} = -\left(\frac{eE_x}{m}\right)$	-%; -%;
$\frac{dV_{in}}{dt} = -\left(\frac{eE_{in}}{m}\right)$	-14-1 -14-1
$\frac{d}{dt} \langle V_{x} \rangle = -\left[\frac{e E_{x}}{m}\right]$ $\langle V_{x} \rangle \text{ is the average}$	
velocity of all n electrons $\langle V_n \rangle = \frac{1}{n} \left[ \sum_{i=1}^n V_{in} \right]$	
current density along n-direction	
$J_n = n(-e) \langle V_n \rangle$	
$\frac{dJ_n}{dt} = -ne \frac{d}{dt} \langle v_n \rangle$	
$-\frac{1}{ne} \left[ \frac{dJ_x}{dt} \right] = \frac{d}{dt} \left( V_x \right)$	

Page 5 Mon Tue Wed Thu Fri Sat Date: \_\_/\_\_/20 It is entermely difficult to solve scattering term. we therefore obtain an enpression for [d (Vn)]el-la by a method which by-passes these dificulties at t = 0, (Va) just switch the field off so (Vn) tends to zero, let assume that this process follows implest law of decay

(Vx) = (Yx), e T = time constant or relaxation time in fact it is the time for the drift velocity to decay /e of its initial value when Ex = 0 Now we can obtain el-la term withoutanyd considring detalis of [d (Vx)] = - (Vx) = - (Vx) Mon Tue Wed Thu Fri Sat Page 6  $\left[\frac{d\langle v_x\rangle}{dt}\right] = -\left[\frac{eEn}{m}\right]$   $\left[\frac{d\langle v_x\rangle}{dt}\right] + \left[\frac{d\langle v_x\rangle}{dt}\right] = \frac{eEn}{m}$  $\left[\frac{d}{dt}\left\langle v_{n}\right\rangle \right]_{d-la}=-\left\langle v_{n}\right\rangle _{T}$ [eEn] - (Vx) = 0  $\langle V_n \rangle = - \left[ e^{\tau} E_n \right]$ so the steaty averge velocity is proportional to the applied field Proportionality Know as mobility "" (Vx) = - M Ex  $u = \langle V_n \rangle = e\tau$ 

- The three integers  $n_x$ ,  $n_y$  and  $n_z$  which are called quantum numbers, are required to specify completely each energy state. No quantum number can be zero, because if any one of them is taken as zero, then  $\psi(x, y, z) = 0$ , which would mean that the particle does not exist in the box.
- 2. It should be noted, however, that the energy E depends only upon the sum of the squares of the quantum numbers  $n_x$ ,  $n_y$  and  $n_z$  and not on their individual values.
- 3 Several combinations of the three integers may give different stationary states or different wave functions, but of the same energy value. Such states and energy levels are said to be degenerate.

### Example:

$$1. n_x = 1, n_y = 1, n_z = 2$$

$$2. n_x = 2, n_y = 1, n_z = 1$$

$$3. n_x = 1, n_y = 2, n_z = 1$$

The corresponding wave functions are then

$$\psi_{112} = A \sin \frac{\pi x}{a} \sin \frac{\pi y}{a} \sin \frac{2\pi z}{a}$$
 where  $A = \sqrt{\frac{8}{a^3}}$ 

$$\psi_{211} = A \sin \frac{2\pi x}{a} \sin \frac{\pi y}{a} \sin \frac{\pi z}{a}$$

$$\psi_{121} = A \sin \frac{\pi x}{a} \sin \frac{2\pi y}{a} \sin \frac{\pi z}{a}$$

and

but in each case the energy value is exactly the same; that is

$$E_{n_x,n_y,n_z} = \frac{6h^2}{8ma^2}$$

We know that if several states have the same energy, they are said to be degenerate. Thus for the level for which the energy is  $\frac{6h^2}{8ma^2}$  there are three independent states having quantum numbers (1,1,2)(2,1,1) and (1, 2, 1). The level is, therefore, said to be three fold degenerate or triply degenerate. It will be noted that the

groundstate (1, 1, 1), as also several other states like (2, 2, 2), are non-degenerate.

It can be shown with an example that the degeneracy breaks on introducing a small modification to the system. Consider, for example, the triply degenerate level, having  $E = \frac{6h^2}{8ma^2}$ . For this energy value, there are

three independent energy states having quantum numbers (2, 1, 1), (1, 2, 1) and (1, 1, 2) for  $n_x$ ,  $n_y$ ,  $n_z$ . Let the length of the cubical box be increased by a small amount, da along the x-axis keeping breadth and thickness

unaltered. The corresponding change in energy of the first state (2, 1, 1), which has an energy  $\frac{4h^2}{8ma^2}$  associated

with the x-direction, will be different from that for the other two states (1, 2, 1) and (1, 1, 2), which have

an energy  $\frac{h^2}{8ma^2}$  associated with the x-direction. The decreases in energy of the first state (2, 1, 1) is

energy  $\frac{3}{2}k_BT$  [refer equation (6.1)]. Thus one kmol of a metal which has  $N_A$  atoms will therefore have  $N_A$ free electrons, assuming that each atom contributes one valence electron to the electron gas. Now the energy associated with one basel of associated with one kmol of such a metal is (6.18)

$$U = \frac{3}{2} N_A k_B T$$
 (6.18)

If heat is supplied to the metal, these free electrons also absorb part of the heat, and the molar elec-

tronic specific heat is obtained as follows: 
$$[C_{\nu}]_{el} = \left[\frac{dU}{dT}\right] = \frac{3}{2}k_{B}N_{A} = 1.5R_{u}$$
 (6.19)

where  $R_u$  is the universal gas constant

$$[C_y]_{el} = 1.5 \times 1.38 \times 10^{-23} \times 6.02 \times 10^{26} = 12.5 \times 10^3 \text{J/kmol/K}$$

That is, molar electronic specific heat = 12.5 kJ/kmol/K. The above value of  $1.5 R_u$  due to free electrons is about hundred times greater than the experimentally predicted value. Since the heat capacity of a solid due to atomic vibrations in 3  $R_u$ , free electrons should make a significant contribution to the total specific heat of a metal. It is, however, that at least at high temperatures, Dulong and Petit law (lattice specific heat) holds good and the total specific heat of a solid is  $3 R_u$ ; this means that the free electrons do not contribute significantly to the heat capacity of a metal. It is, therefore, concluded that the law of equipartition and hence classical Maxwell-Boltzmann statistics must not be applied to evaluate the electronic specific heat in metals.

### (ii) Computation of mean free path

The microscopic expression for the resistivity of a metal is given by the equation (6.12),

$$\rho = \frac{m}{ne^2\tau}$$

The resistivity of the most useful metal copper at 20°C is 1.69×10<sup>-8</sup> ohm-m and the concentration of free electrons in copper,  $n = 8.5 \times 10^{28} / \text{m}^3$ . Thus

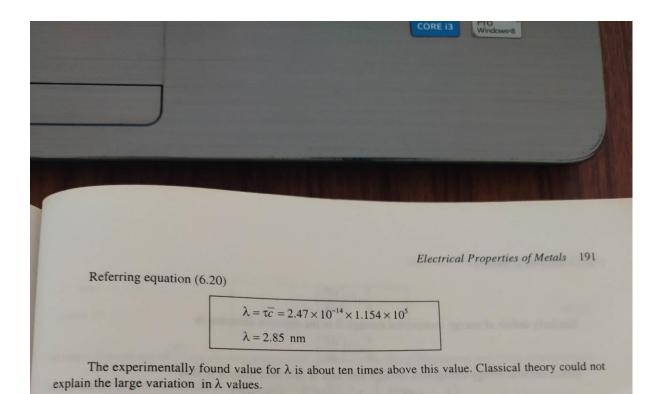
$$\tau = \frac{m}{ne^2 \rho}$$

$$= \frac{9.11 \times 10^{-31}}{8.5 \times 10^{28} \times (1.6 \times 10^{-19})^2 \times 1.69 \times 10^{-19}}$$

$$\tau = 2.47 \times 10^{-14} \text{sec}$$

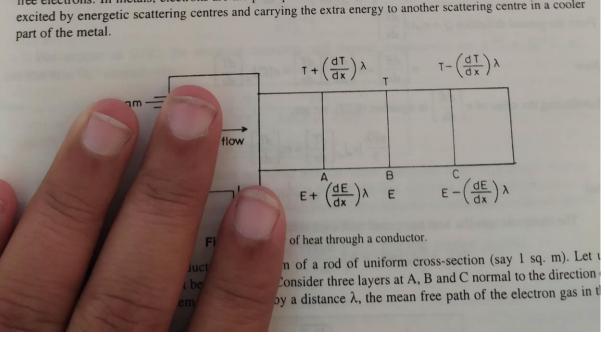
But

$$\tau = \frac{\lambda}{\overline{c}} \tag{6.20}$$



# (iii) Relation between electrical conductivity and thermal conductivity (Wiedemann-Franz law)

Heat conduction in solids may take place either through the mechanism of the atoms or through that of the free electrons. In metals, electrons are the principal carriers of heat energy. Free electrons do this by being excited by energetic scattering centres and carrying the extra energy to another scattering centre in a cooler part of the metal.



Similarly, we can solve equation (6.91) to obtain

$$Z = \left[\sqrt{\frac{2}{c}}\right] \sin \frac{n_z \pi z}{c} \tag{6.96}$$

The complete wave function

$$\psi(x, y, z) = XYZ$$

i.e.,

$$\psi = \left[\sqrt{\frac{2}{a}}\right] \sin\frac{n_x \pi x}{a} \left[\sqrt{\frac{2}{b}}\right] \sin\frac{n_y \pi y}{b} \left[\sqrt{\frac{2}{c}}\right] \sin\frac{n_z \pi z}{c}$$

$$\psi_{n_x, n_y, n_z} = \frac{2\sqrt{2}}{\sqrt{abc}} \sin\frac{n_x \pi x}{a} \sin\frac{n_y \pi y}{b} \sin\frac{n_z \pi z}{c}$$
(6.97)

Equation (6.88) is

$$E = \frac{\hbar^2}{2m} (k_x^2 + k_y^2 + k_z^2)$$

$$E = \frac{\hbar^2}{2m} \left[ \frac{n_x^2 \pi^2}{a^2} + \frac{n_y^2 \pi^2}{b^2} + \frac{n_z^2 \pi^2}{c^2} \right]$$

$$E = \frac{\hbar^2 \pi^2}{2m} \left[ \frac{n_x^2}{a^2} + \frac{n_y^2}{b^2} + \frac{n_z^2}{c^2} \right]$$

$$E_{n_x, n_y, n_z} = \left( \frac{h^2}{8m} \right) \left[ \frac{n_x^2}{a^2} + \frac{n_y^2}{b^2} + \frac{n_z^2}{c^2} \right]$$
(6.98)

i.e.,

where  $n_x$ ,  $n_y$  and  $n_z$  denote any set of three positive integers. Equation (6.97) gives the total normalized wave functions inside the rectangular box for the stationary states. Similarly equation (6.98) gives the eigen values of the energy of the particle. These values are called the energy-levels of the particle.

If the particle is confined in a cubical box in which a = b = c, the eigen values of energy are given by

$$E_{n_x,n_y,n_z} = \frac{h^2}{8ma^2} [n_x^2 + n_y^2 + n_z^2]$$

$$= \frac{h^2 n^2}{8ma^2}, \text{ where } n^2 = n_x^2 + n_y^2 + n_z^2$$
(6.99)

and the normalized wave-functions are

$$\psi_{n_x,n_y,n_z} = \left[ \sqrt{\frac{8}{a^3}} \right] \sin \frac{n_x \pi x}{a} \sin \frac{n_y \pi y}{a} \sin \frac{n_z \pi z}{a}$$
(6.100)

From equations (6.99) and (6.100), we derive the following conclusions:

$$\frac{4h^{2}}{8ma^{2}} - \frac{4h^{2}}{8m(a+da)^{2}} = \frac{4h^{2}}{8m} \left[ \frac{1}{a^{2}} - \frac{1}{(a+da)^{2}} \right]$$

$$\frac{4h^{2}}{8m} \left[ \frac{a^{2} + 2da(a) + (da)^{2} - a^{2}}{a^{4}} \right] \cong \left( \frac{4h^{2}}{8m} \right) \left[ \frac{2da}{a^{3}} \right]$$

$$= \frac{h^{2}}{8ma^{2}} \left( \frac{8da}{a} \right) \tag{6.101}$$

But for the other two states (1, 2, 1) and (1, 1, 2), the energy will decrease only by  $\left(\frac{h^2}{8ma^2}\right)\left(\frac{2da}{a}\right)$  which

is shown below:

$$= \frac{h^2}{8m} \left[ \frac{1}{a^2} - \frac{1}{(a+da)^2} \right]$$

$$= \frac{h^2}{8m} \left[ \frac{a^2 + (da)^2 + 2a(da) - a^2}{a^4} \right]$$

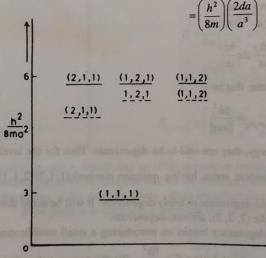
$$= \left( \frac{h^2}{8m} \right) \left( \frac{2da}{a^3} \right)$$

$$= \left( \frac{h^2}{8ma^2} \right) \left( \frac{2da}{a} \right)$$

$$= \left( \frac{h^2}{8ma^2} \right) \left( \frac{2da}{a} \right)$$
(6.102)
This breakdown of degeneracy is shown in Fig. 6.15 by comparing equation (6.102)

Fig. 6.15 by comparing equation (6.102) with (6.101).

Breakdown of degeneracy, which thus occurs due to some modification applied to the system, has relevance to what is often observed in practice, such as the splitting of spectral lines in a magnetic field or electric field. As a result of the application of a magnetic or electric field the degenerate level breaks up into separate levels, and consequently, spectral lines which were single become multiples.



Breakdown of degeneracy of the level (2, 1, 1).

#### FERMI-DIRAC STATISTICS AND ELECTRONIC DISTRIBUTION XIV. IN SOLIDS

The most important assumption of classical statistics is that any number of particles may have identical energies. Though the classical approach gives an understanding of the conductivity of metals, it fails to give a proper explanation for the contribution by electrons to the specific heat and magnetic susceptibility. As the

electrons are free, the total molar specific heat of a metal at room temperature and above should be  $3R_u + \frac{3R_u}{2}$ ,  $3R_u$ 

being the contribution from the lattices and  $\frac{3R_u}{2}$  from the electrons. But experimentally the specific heat of a metal near room temperature is only  $3R_u$ . Thus the specific heat associated with the electron must be very small and

In order to account for the unexpectedly low specific heat of metals, we must look more closely into the statistical behaviour of the electron gas. Such an electron gas behaves like a system of Fermi particles and hence obeys Fermi-Dirac statistics. Fermi-Dirac statistics is applied to indistinguishable particles (called Fermions) which are governed by Pauli's exclusion principle. Sommerfeld investigated the behaviour of a free electron gas taking into account the quantum theory and the Pauli's exclusion principle. He considered the distribution of a large number of electrons of the order of 10<sup>28</sup>/m<sup>3</sup> in thermal equilibrium among the various states in a three dimensional box.

Assuming that the metal is in the form of a cube of side a, we have immediately the wave function of the stationary states as

$$\Psi_{n_x,n_y,n_z} = \left[ \sqrt{\frac{8}{a^3}} \right] \sin \frac{n_x \pi x}{a} \sin \frac{n_y \pi y}{a} \sin \frac{n_x \pi z}{a}$$

within the metal and zero outside, and the energy levels are

$$E_{n_x,n_y,n_z} = \frac{h^2}{8ma^2} (n_x^2 + n_y^2 + n_z^2)$$

where  $n_x$ ,  $n_y$  and  $n_z$  are integers.

$$E_n = \frac{h^2 n^2}{8ma^2}$$
 with  $n^2 = n_x^2 + n_y^2 + n_z^2$ 

According to the Pauli's principle, not more than two electrons may occupy any orbital state, so that, at the absolute zero of temperature, two electrons will go into the ground state, two into each state of next higher energy, and so on, until all the electrons are allocated to states of lowest possible energy. Since the number of electrons is very large, it is thus understandable that, even at the absolute zero of temperature, some electrons have kinetic energies of several electron volt. For a piece of metal of macroscopic dimensions, say centimetre cube, the energy of the ground state  $(n_x = n_y = n_z = 1)$  is of the order of  $10^{-15}$  eV and hence may be taken to be zero for all practical purposes. Also the maximum spacing between consecutive energy level is less than 10<sup>-6</sup> eV, so that the distribution of energy may be regarded as almost continuous or sometimes 'quasi-continuous'.

Now we shall consider how the many electrons of a real material distribute among these available energy states. Since we are talking about an almost continuous range of energies, we can represent the probability of occupying a given state by a continuous distribution function. The probability F(E) of an electron occupying a given energy level is given by

$$F(E) = \frac{1}{1 + e^{\frac{E - E_F}{k_B T}}}$$
(6.103)

Here F(E) is called the Fermi function, E is the energy of the level whose occupancy is being considered,  $E_F$ is the Fermi level and is a constant for the particular system. At absolute zero F(E) = 0 for  $E > E_F$  and F(E)= 1 for  $E < E_F$ . Thus at absolute zero, the Fermi level divides the occupied states from the unoccupied states; i.e., it is the highest energy state for the electrons to occupy at absolute zero.

near the termi level take part in thermal and electrical conductions.

## IV. RELAXATION TIME, COLLISION TIME AND MEAN FREE PATH

(i) Relaxation time

Suppose at a given time t = 0, the average velocity of the electron is  $\langle v_x \rangle_0$ . Just at that instant we switch the field off, so that  $\langle v_x \rangle$  will subsequently tend to zero. In other words, the velocity will be randomized by electron-lattice collisions. Now let us assume that this process follows the simplest law of decay.

$$\langle v_x \rangle = \langle v_x \rangle_0 e^{-t\hbar} \tag{6.27}$$

The factor  $\tau$  is a constant called *relaxation time*, because it gives a measure of the time that the system takes to relax when a constraint (the electric field) is removed. When  $t = \tau$ , the equation (6.27) becomes

$$\langle v_x \rangle = \frac{\langle v_x \rangle_0}{e}$$

Thus relaxation time may be stated as the time taken for the drift velocity to decay to  $\frac{1}{e}$  of its initial value.

Differentiating equation (6.27)

ssp page 1

$$\left[\frac{d}{dt} < v_x > \right]_{elec-latt} = -\frac{\langle v_x >_0 e^{-t/\tau}}{\tau}$$

$$\left[\frac{d}{dt} < v_x > \right]_{elec-latt} = -\frac{\langle v_x >_0 e^{-t/\tau}}{\tau}$$
(6.28)

#### (ii) Collision time

In the last section we introduced the concept of relaxation time in a highly mathematical way. However, we shall now show that it also has a very simple physical meaning. Suppose that the probability of an electron making a collision in time dt is  $\frac{dt}{\tau_c}$ , so that on an average there are  $\frac{1}{\tau_c}$ , collisions per second. It is now our

intention in what follows is to show that  $\tau_c = \tau$ .

Let us assume that all electron collisions are elastic; that is, energy is conserved, so that the speed of the electron is the same before and after collision. This assumption is not strictly correct, because a conductor warms up when current is passing; it implies that energy is transferred to the lattice during electronlattice collisions. However, the actual energy change at each collision is minute, and the fractional change in velocity in a metal is typically 1 in 10<sup>6</sup>, so that the assumption is entirely reasonable. If we assume that after collision the velocities are completely random, we have

$$\langle v_x \rangle_{\text{after}} = 0$$
. Since  $\langle v_x \rangle_{\text{before}} = \langle v_x \rangle$ ,

it follows that the change of  $\langle v_x \rangle$  on collision =  $-\langle v_x \rangle$ .

There are  $\frac{1}{\tau}$  collisions per second, so that

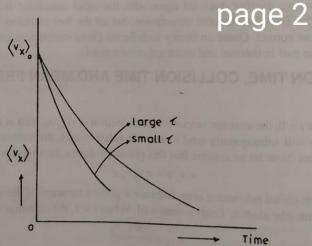


Fig. 6.3 Relaxation of a system of electrons to equilibrium after removing the electric field as given by equation (6.27).

the rate of change of

$$\frac{d}{dt} < v_x > = -\frac{\langle v_x \rangle}{\tau_c} \tag{6.29}$$

However, we could equally write the left-hand side of this equation as

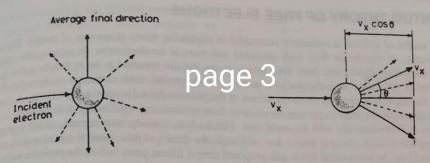


Fig. 6.4(a) Scattering equally probable in all directions, so that the average angle is 90°

Fig. 6.4(b) Scattering predominantly in the forward direction at average angle  $\theta$ . Change in any  $v_x$  is therefore, on an average,  $v_{\star}(1-\cos\theta)$ .

$$\left[\frac{d < v_x >}{dt}\right]_{\text{elect-latt}},$$

<sub>50</sub> that on comparing equations (6.29) and (6.28) we see that  $\tau = \tau_c$  and the relaxation time, therefore, equals the mean time between collisions.

In deriving this result, we assumed that the velocities took up random directions after each collision [Fig. 6.4(a)]; this is equivalent to stating that after each collision the electron had no memory of what went before. However, it is possible that the scattering is very weak, in which case such an assumption is unrealistic. If, on an average, the velocity direction changes by  $\theta$  on collision [see Fig. 6.4 (b)], then the change of  $\langle v_x \rangle$  is on an average,  $-\langle v_x \rangle$  (1 – cos  $\theta$ ). Consequently

$$\tau = \frac{\tau_c}{[1 - \langle \cos \theta \rangle]} \tag{6.30}$$

where  $\tau$  is the relaxation time and  $\tau_c$  is the mean time between collisions.

In other words, the relaxation time is then much longer than the mean time between collisions; these electrons have a memory.

### (iii) Mean free path

The mean free path  $\lambda$  of an electron is defined as below:

$$\lambda = c\tau_c$$

In case, the obstacles are hard spheres, the mean free path  $\lambda$  is determined by the concentration of these for a given concentration of obstacles, the collision time  $\tau_c$  becomes inversely proportional

# Temperature dependence of electrical resistivity

In the absence of an electric field, the free electrons in a metal will be moving about at random in all directions and will be in temperature equilibrium with it. The kinetic energy associated with the electron is

$$\frac{1}{2}m(\overline{c})^2 = \frac{3}{2}k_BT$$

When an electric field is applied, the electron will acquire a drift velocity and the resulting accelera-

tion is,  $a = \frac{eE}{m}$ . The drift velocity is small compared to the random velocity  $\overline{c}$ . Further the drift velocity is not

retained after a collision with an atom because of the relatively large mass of the atom. Hence just after a collision the drift velocity is zero. If the mean free path is  $\lambda$ , then the time that elapses before the next colli-

sion takes place is  $\frac{\lambda}{c}$ . Hence the drift velocity acquired before the next collision takes place is

 $u = acceleration \times time interval$ 

$$= \left(\frac{eE}{m}\right) \left(\frac{\lambda}{c}\right)$$

Thus the average drift velocity is

$$\frac{u}{2} = \frac{eE\lambda}{2m\bar{c}}$$

If n is the number of electrons per unit volume, then the current flowing through unit area for unit time is

$$J_x = \frac{neu}{2} = \frac{ne^2 E \lambda}{2m\overline{c}}$$

or

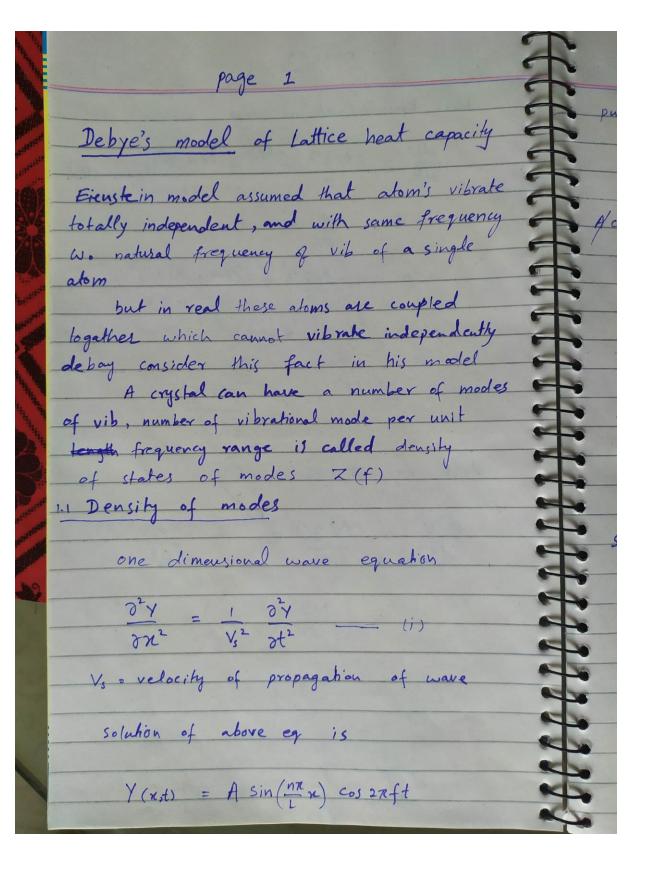
$$\sigma = \frac{J_x}{E} = \frac{ne^2\lambda}{2m\overline{c}}, \text{ or } \rho = \frac{2m\overline{c}}{ne^2\lambda}$$
 (6.16)

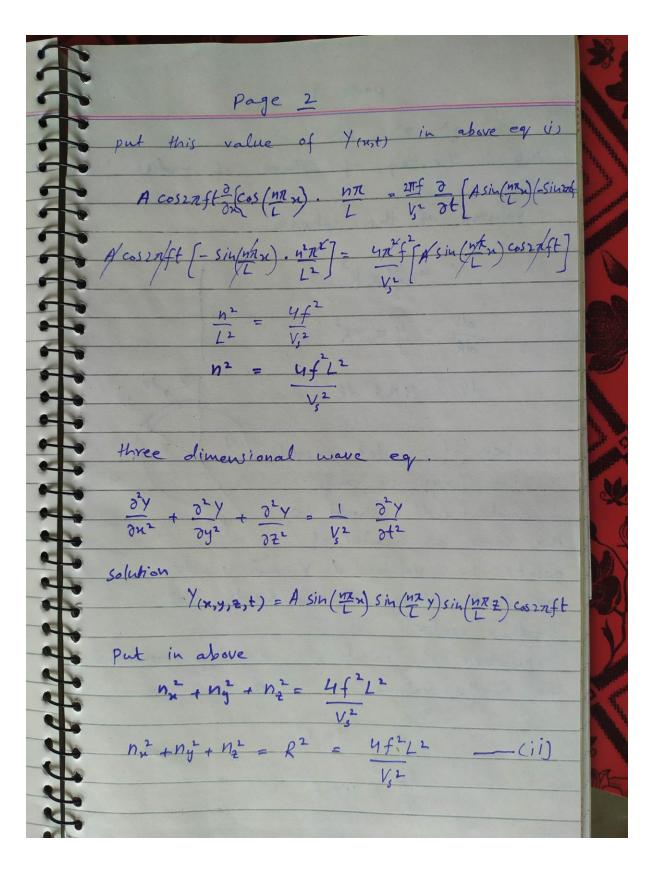
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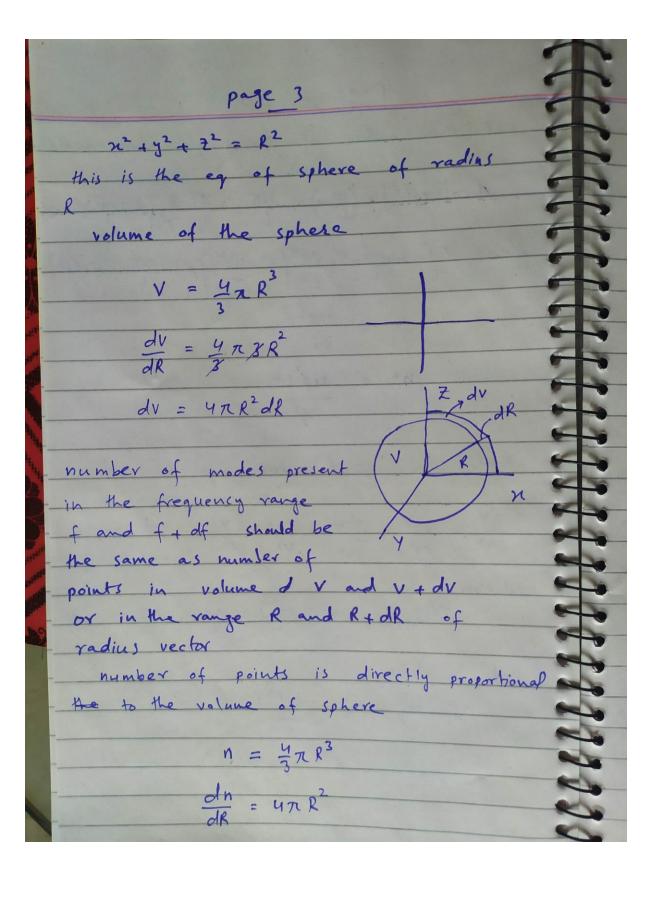
$$\rho = 2 \times \sqrt{\frac{3mk_BT}{ne^2\lambda}}; \sigma = \frac{ne^2\lambda}{\sqrt{12mk_BT}}$$
(6.17)

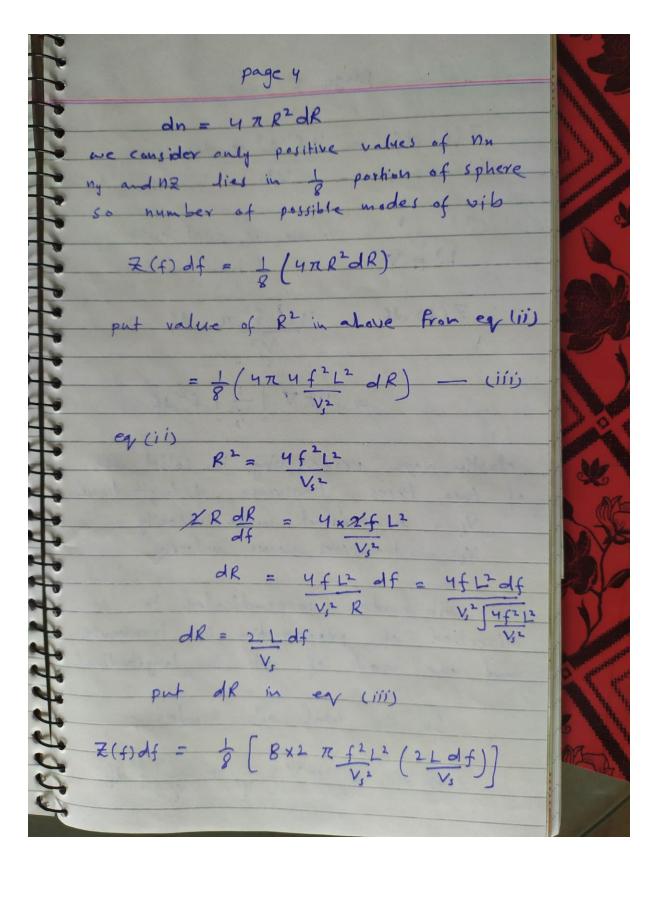
It was assumed by Drude and Lorentz that  $\lambda$  is independent of temperature and that is of the order of interatomic distance. Hence  $\rho \propto \sqrt{T}$ . This means that the specific resistance of an electric conductor is interatomic distance. Hence  $\rho \propto \sqrt{T}$ . This means that the specific resistance of an electric conductor is interatomic distance. Hence  $\rho \propto T$ . Apart from this discrepancy, it is also not correct to assume that the experimental observation that  $\rho \propto T$ . Apart from this discrepancy, it is also not correct to assume that the experimental observation that  $\rho \propto T$ . Apart from this classical theory is almost an unacceptable one. The mean free path is independent of temperature and hence this classical theory is almost an unacceptable one. The mean free path is independent of the field. However, the Ohm's law is derived, since the conductivity in equation (6.16) is independent of the field.

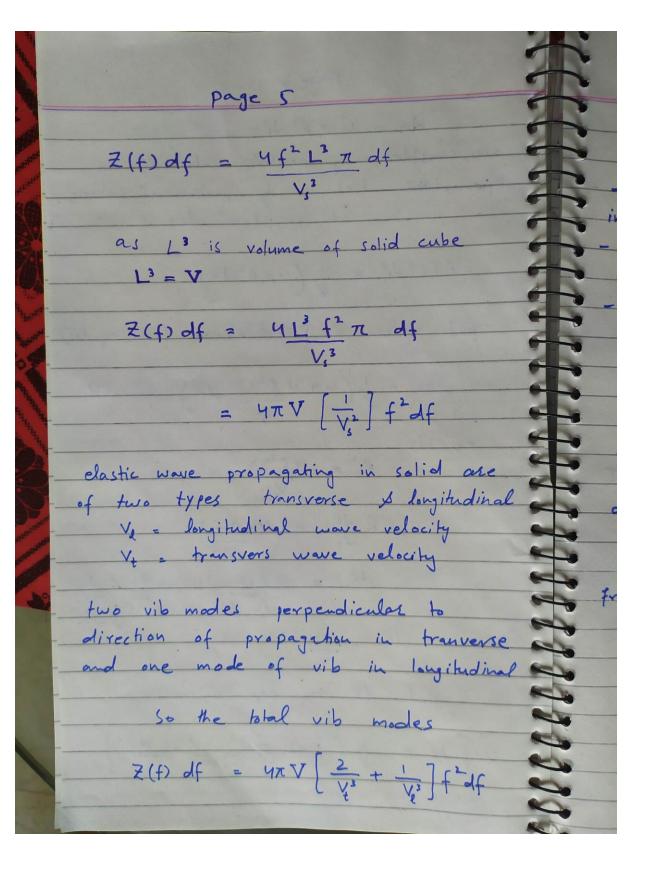
III. DRAWBACKS OF CLASSICAL THEOP

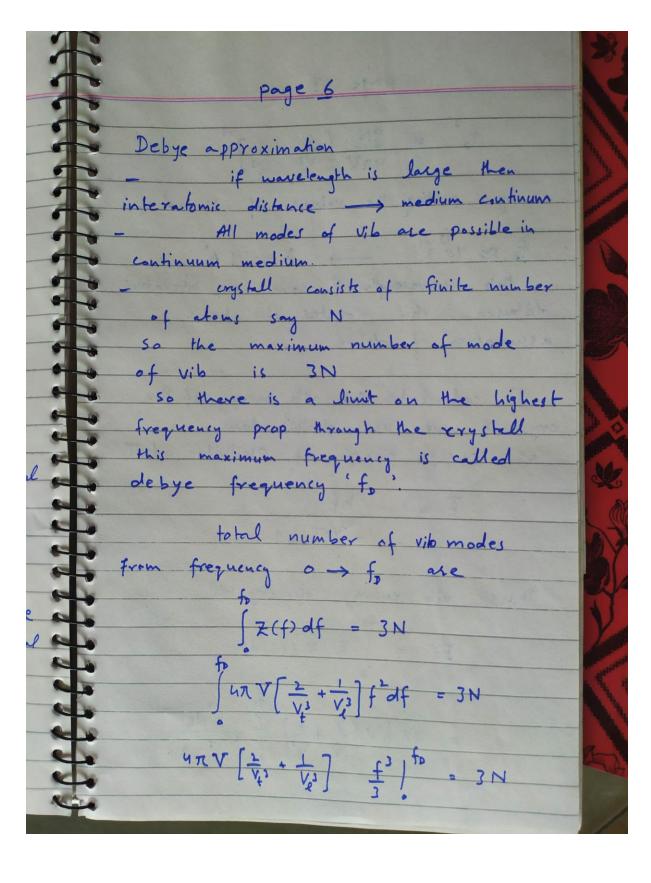


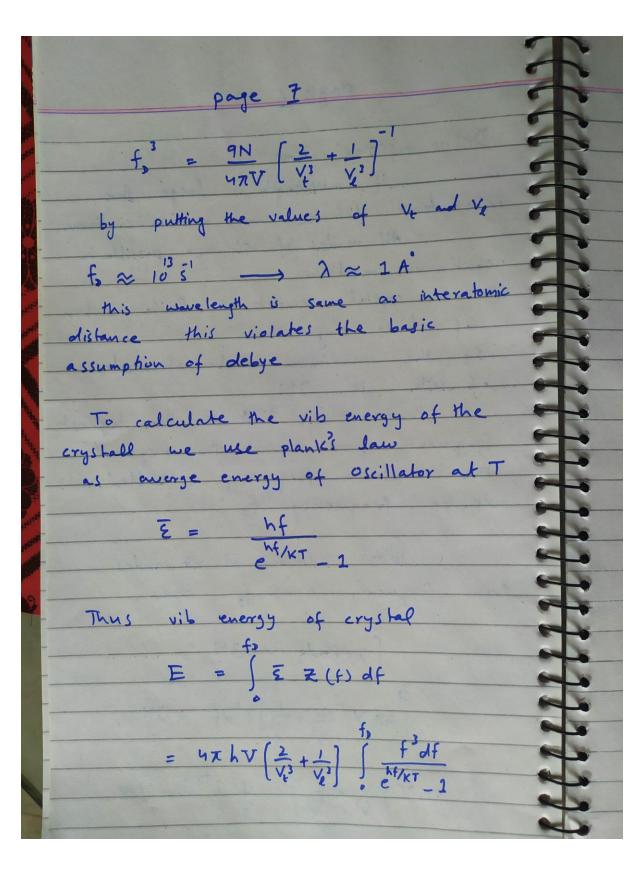












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as 
$$4\pi V \left(\frac{2}{V_{s}^{2}} + \frac{1}{V_{s}^{2}}\right) = \frac{9N}{f_{s}^{3}}$$

$$E = \frac{9Nh}{f_{s}^{3}} = \frac{e^{hf/kT} - 1}{e^{hf/kT} - 1}$$

$$\frac{hf}{kT} = \frac{x}{kT} = \frac{x}{kT}$$

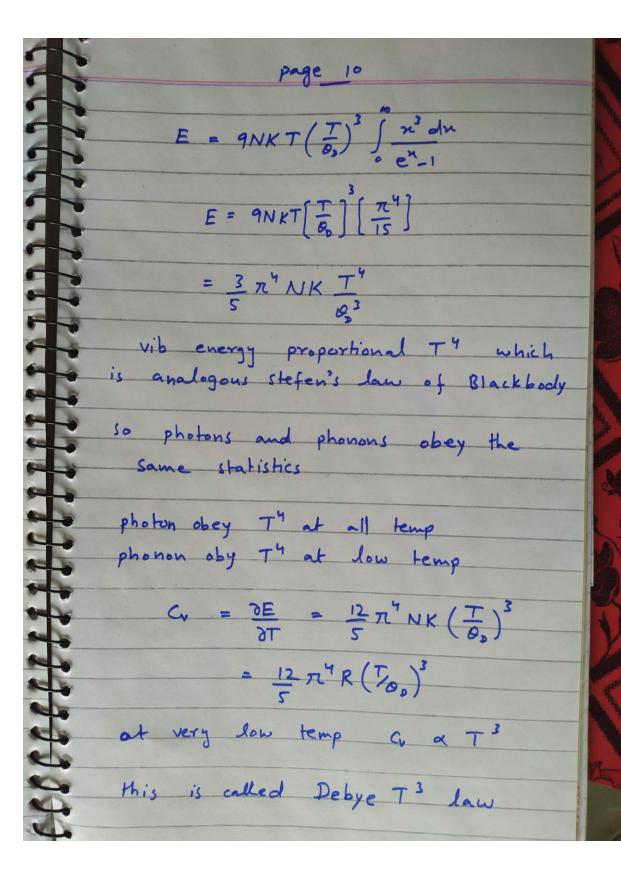
$$\frac{df}{df} = \frac{x}{kT} = \frac{x}{h}$$

$$\frac{df}{f_{s}^{3}} = \frac{x}{h} = \frac{hf_{s}}{h}$$

$$\frac{x}{h} = \frac{\theta_{s}}{h} = \frac{hf_{s}}{h}$$

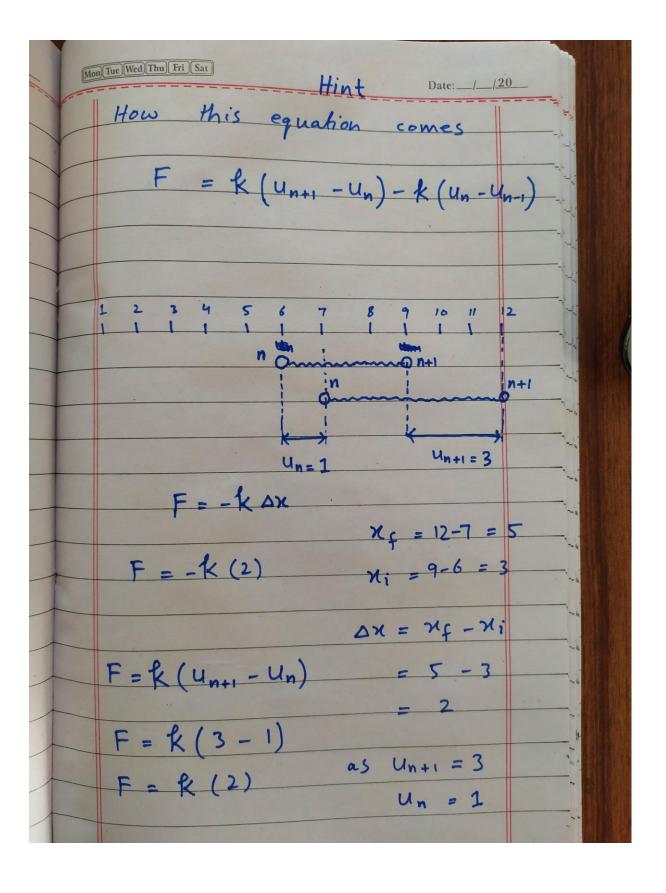
$$\frac{x}{h} = \frac{hf_{s}}{h} = \frac{hf_{s}}{h}$$

heat capacity 3R (0)/T) Fo : N=NA high temp casee T >> 00 E = 3NKT Cv = 3R (Dulong and petit's law) ii) low temp T << 0,  $\frac{\theta_0}{T} = \chi_m \to \infty$ 



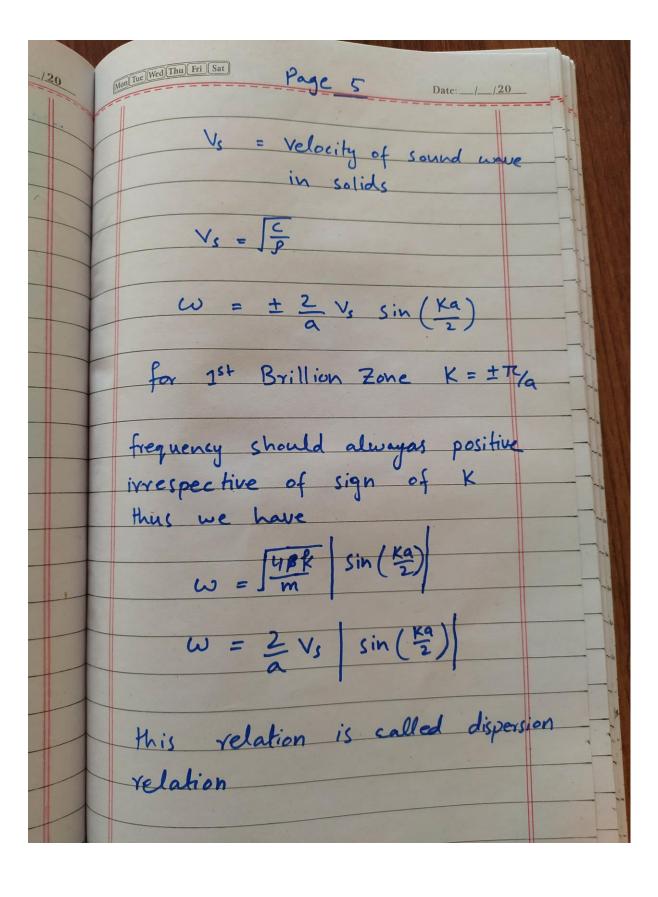
Mon Tue SSP-II Mon Tue Wed Thu Fri Sat Lattice vibrations - motion of atom is coupled Lattice may vib freely due to internal energy Lattic also vib under the effect of dynamical enternal forces which my be mechanical or electromagnetic in nature these vib describes the thermal properties such as specific heat and thermal conductivity. one dimensional chain of atoms N-2 Un-2

	Non Tue Wed Thu Fri Sar Page 2 Date: _/_/20_
The state of the s	the equilibrium state of alon
	if interatomic distance is 'a'
	then x - coordinates of the
1	atoms are na, (n-1)a, (n+1)a,
	from mean position be un, un-
1	Un+1
-	according to Hook's Jaw
	F = Ku
+	Spring displacement
	Constant
	the net force on nth atom
-	is
	F = K (un+1 - un) - K (un- un-1)
	F = K [ Un+1 + Un-1 - 2Un]
	as $F = m \frac{d^2 u_n}{dt^2}$



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m d'un = k [ un++ + un-, - 2 un] - i)	
Solution of above diffrential	
$U_n = U_0 e^{i(\omega t - Kna)}$ (iv)	
displacement of nth atom x = na	
$K = 2\pi/\lambda$	
Similarly for (n+1)th and (n-1)th	
ii) i [wt - K(n+1)a]	
i [wt - K(n-1)a] iii) Un-1 = Uo e	
put value of Un, Un+1 and Un-1	
in eq (v	
t une e e -2 Moè e	1
	m d'un = k [ un+ + un, -2 un] (i)  Solution of above differential  eq is  Un = Uo e (wt - kna)  (iv)  displacement of nth atom x = na  K = 27/2  Similarly for (n+1)th and (n-1)th  atoms  i(wt - k(n+1)a)  Un+1 = Uo e  put value of Un, Un+1 and Un-1  in eq (v  miw Uo e = k [ uo e e e e

Mon Tue Wed Thu Fri Sat Page 4 120 mw² 40 e = k y e e e ka jeka ika -mw2 = k [ e + e - 2] = k [ e ka/2 - ika/2]2 as  $\sin n = \frac{i^n - i^n}{2i}$   $\sin^2 n = -\frac{i^n - i^n}{4}$ -mw2 = -4k sin2 (Ka)  $\omega = \pm \int \frac{4k}{m} \sin\left(\frac{ka}{2}\right)$ longitudinal stiffness C = Ka mass per unit length & = m/a  $\omega = \pm 2 \frac{C/a}{ga} \sin(\frac{\kappa a}{2})$  $\omega = \pm \frac{2}{2} \left[ \frac{C}{f} \sin\left(\frac{Ka}{2}\right) \right]$ 



Mon Tue Wed Thu Fri Sat Page 6 Date: 120	(Mon) Tue
low frequency -> large >	e
$K = \frac{2\pi}{\lambda} = \frac{2\pi}{\omega} \approx 0$	
So K -> O	5
As for small angle 0 ≈ 0 Sin 0 ≈ 0	i
$\frac{\sin\left(\frac{Ka}{2}\right) \to \frac{Ka}{2}}{2}$	
$\omega = \frac{z}{y}, \frac{k\alpha}{z}$	5
$\omega = V_{\varsigma} K$ $V_{\varsigma} = \frac{\omega}{K}$	
we know that	
phase velocity = $V_p = \frac{\omega}{K}$	
group velocity = Vg = dw  As dK	
$V_s = \frac{\omega}{k} = V_p$	

