Nuclear Reactions Applications

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DEFINITION

- Nuclear physics is the field of physics that studies atomic nuclei and their constituents and interactions.
- ≻It includes the study of,
- 1. The general properties of nucleus.
- 2. The particles contained in the nucleus.
- 3. The interaction between these particles.
- 4. Radio activity and nuclear reactions.
- 5. Practical applications of nuclear phenomenon.

NUCLEAR REACTIONS

- ¬A nuclear reaction is a rearrangement of nuclear components induced by particle bombardment.
- 1. Fission
- 2. Fusion
- ¬Nuclear reactions are subject to the following conservation laws:
- 1. Charge.
- 2. Momentum and angular momentum.
- 3. Energy.

4. Total number of nucleons.

FISSION

- Splitting of large nucleus to smaller ones results in the liberation of energy.
- $\neg n + 235U \rightarrow 141Ba + 92Kr + 3n$
- In this reaction more than
 200MeV energy liberated.
- Chain reaction is controlled and is applied in fission reactors.
- Chain reaction left uncontrolled and is applied in atom bomb.



FISSION CHAIN REACTION



At each step energy is released !

NUCLEAR FISSION REACTOR

- \neg Place where controlled fission chain reaction occurs.
- Liberated energy is used for the production of electricity.
- ¬ <u>PARTS OF THE REACTOR;</u>
- 1. Fuel-Uranium, Plutonium, Thorium
- 2. Moderator-Water, Graphite- slow down neutron.
- 3. Control rods-Cadmium-
- 4. Concrete wall-

absorb neutrons.

block radiations.

NUCLEAR FISSION REACTOR



FUSION

- Two light nuclei fuse together to form a heavier nucleus results in the liberation of energy.
- 2H + 3H → 4He + n + 17.6 MeV
- It is the primary source of the sun's energy.
- Fusion reaction is applied in hydrogen bomb.



FISSION & FUSION FISSION FUSION



FUSION REACTOR

work in progress: ITER International Thermonuclear **Experimental Reactor** (500MW power in 1000second through fusion reaction)



FISSION & FUSION

FISSION

- Splitting a large mass nucleus into two medium mass nucleus.
- Energy released per nucleon is less.
- \neg Can be controlled.
- \neg Heat is not needed.
- Lot radioactive

FUSION

- Joining two low mass nuclei into a larger mass nucleus.
- Energy produced per nucleon is large.
- Can not be controlled.
- Extreme heat is needed.
- No radioactive byproducts.

APPLICATIONS ARE

Medical radio-isotopes (imaging & therapy).
 Magnetic Resonance Imaging (MRI).
 Identification of materials.
 Dating of materials.
 Power generation (fusion and fission).
 Weapons of mass destruction (WMD).

REFERENCE

- NCERT physics text book.
- ✤ <u>www.elsevier.com.</u>
- ✤ <u>www.sciencedirect.com.</u>
- * <u>www.wikipedia.org.</u>