


→ Chromophore: Light absorb → colours emit
↳ functional groups.

→ Auxochrome: Change in λ ↑ & ↓
↳ Atoms which alter λ in longer / smaller shift.

→ How to find λ_{max} theoretically?

Conjugated System → Diene

=  → Conjugated diene



(Conjugated diene)

↳ Diene - extend.


Homoannular system - prefer over heteroannular


conjugation & λ_{max}


λ_{max} diene < λ_{max} triene.

Woodward Fieser Rule:

Diene → 3 - Types ↳ λ_{max} - theoretically calculate

1. Ayclic  → 217 (Parent value).

2. Cyclic - Homo Annular  → 2 alternative double bonds in one ring
(Parent value) 253 nm.

Hetero Annular  2 alternative double bonds in 2 rings.

→ Increments Value 214 - 417 nm (P.V).

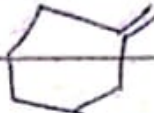
1. parent value


2. Extension of conjugation → +30 (one double bond)

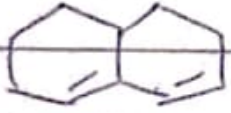
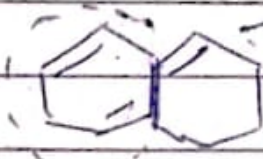
3. Exocyclic double bond = +5

4. Substituent value

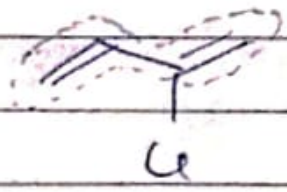
(R) alkyl $\rightarrow +5$ OR $\rightarrow +6$ SR $\rightarrow +30$
 $\text{OCOCH}_3 \rightarrow 0$ X $\rightarrow +5$ $\text{NR}_2 \rightarrow +60$
 Homoannular $\rightarrow 253$

Exocyclic double bond  (outside cycle)

Endocyclic double bond  (inside cycle)

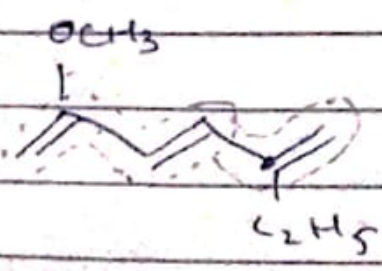
 \downarrow Heteroannular
 \downarrow Diene Exocyclic double bond

\rightarrow Calculate λ_{max}



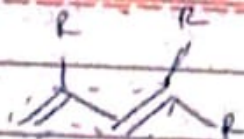
- P-value = 217 nm
- extra conjugation = 0
- exocyclic = 0
- Substituent (X) = +5

$\lambda_{\text{max}} = 222 \text{ nm}$



- P-value = 217 nm
- extra conjugation = +30
- Exocyclic = 0
- Substituent (OCH₃) = +6
- C₂H₅ = +5

$\lambda_{\text{max}} = 258 \text{ nm}$



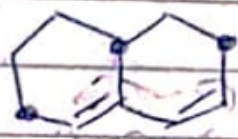
- P-value = 217
- Extension of conjugation = 0
- Exocyclic = 0
- Substituents = $R \times 3 = 15$

$$\lambda_{max} = 232 \text{ nm}$$



- P-value = 217
- Extra conjugation = 0
- Exocyclic = +5
- Substituents = 10

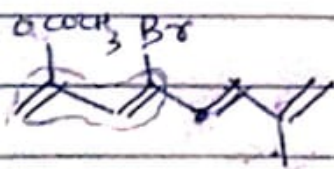
$$\lambda_{max} = 232 \text{ nm}$$



bicyclic
annular
cyclic
diene

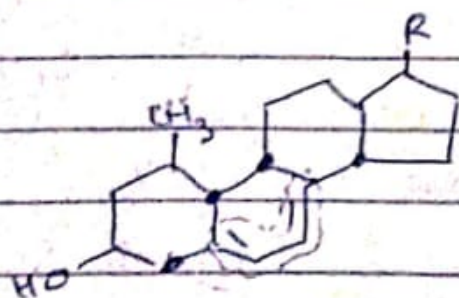
- P-value = 214 nm
- Extra conjugation = 0
- Exocyclic = +5
- Substituents = 15

$$\lambda_{max} = 234 \text{ nm}$$



- P-value = 217
- E-conj = +60
- Exo = 0
- Sub = +5
+5
+0

$$\lambda_{max} = 287 \text{ nm}$$



- P-value = 253 nm
- E-conj = 0
- Exo = +5 + 5
- Sub = 20 + 5 + 5 + 6

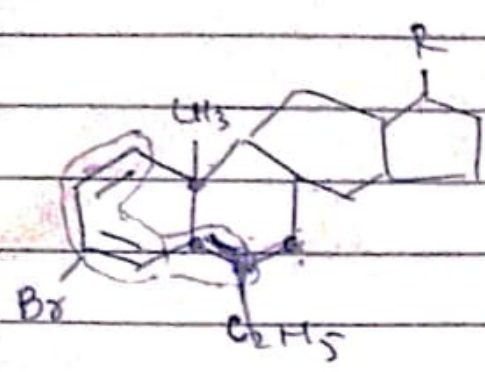
$$\begin{array}{r} 253 \\ 46 \\ \hline \end{array}$$

$$\lambda_{max} = \cancel{294} \text{ nm} \quad 283 \text{ nm}$$



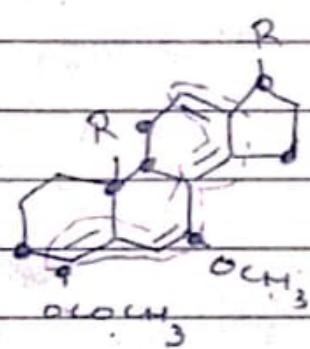
$$\begin{aligned}
 P \text{ value} &= 214 \text{ nm} \\
 \epsilon \text{ conj} &= 0 \\
 \epsilon_{\text{exo}} &= +5 \\
 \text{Sub} &= +15
 \end{aligned}$$

$$\lambda_{\text{max}} = 234 \text{ nm}$$



$$\begin{aligned}
 P &= 253 \\
 \epsilon \text{ C} &= 30 \\
 \epsilon_{\text{exo}} &= 5 \\
 \text{Sub} &= 15 + 5
 \end{aligned}$$

$$\lambda_{\text{max}} = 308 \text{ nm}$$

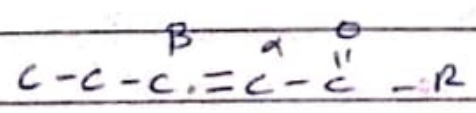


$$\begin{aligned}
 P \text{ value} &= 253 \text{ nm} \\
 \epsilon \text{ conj} &= +60 \\
 \epsilon_{\text{exo}} &= +20 \\
 \text{Sub} &= 30 + 6
 \end{aligned}$$

$$\begin{array}{r}
 253 \\
 + 116 \\
 \hline
 369
 \end{array}$$

$$\lambda_{\text{max}} = 369 \text{ nm}$$

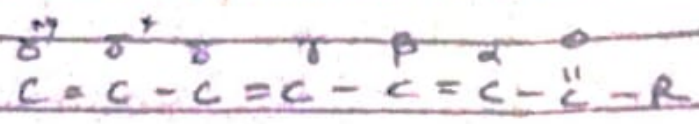
α - β unsaturated carbonyl system.



- 1- Acyclic = 215 nm
- 2- 6 member cyclic ketone = 215 nm
- 3- $-\text{COOH} / \text{COR} = 195 \text{ nm}$
- 4- 5 membered cyclic ketone = 202 nm
- 5- 5 membered cyclic aldehyde 207 nm.

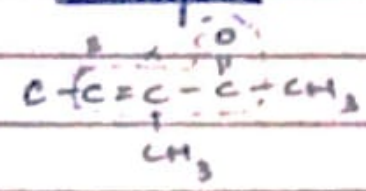
Increase value:

- Hemodienes = +39
- Conjugation = +30
- exocyclic = +5

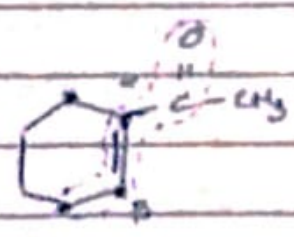


functional group	α	β	γ	δ	δ^{β}	δ^{γ}
OH	35	30	30	20	20	20
OR	35	30	17	31	31	31
OAc	6	6	6	06	06	06
Cl	15	12	12	12	12	12
-R	10	12	18	18	18	18
-Br	25	30	25	25	25	25

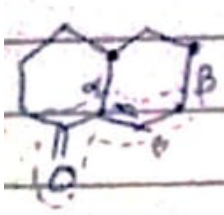
Examples



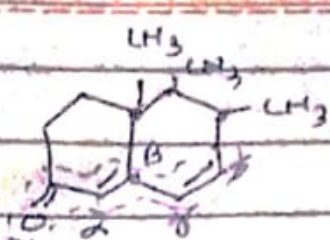
P. value = 215
 Conjugation = 0
 Exocyclic = 0
 Substituents = R - $\alpha = 10$, $\beta = 12$
 $\lambda_{max} = 237 \text{ nm}$



P. value = 215 nm
 Conj = 0
 Exocyclic = 0
 Sub = R - $\alpha = 10$, $\beta = 12$
 $\lambda_{max} = 237 \text{ nm}$



P. value = 215 nm
 Conj = 0
 Exocyclic = +5
 Sub $\alpha = R = 10$, $\beta = 12$
 $\lambda_{max} = 242 \text{ nm}$

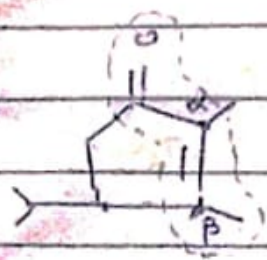


P. value = 215 nm

Exocyclic = +5

Sub $R = \alpha = 0$ $\beta = 12$ $\gamma = 18$

$\lambda_{max} = 280 \text{ nm}$



P. value = 202 nm

Conjugation = 0

Sub $\alpha = R = +10$ $\beta = 2R = 12 \times 2$

$\lambda_{max} = 236 \text{ nm}$



P. value = 202 nm

Conjugation = 0

Exocyclic = +5, +5 = 10

Sub $\beta = R = 24$ $\alpha = 10$

$\lambda_{max} = 246 \text{ nm}$



P. value = 215 nm

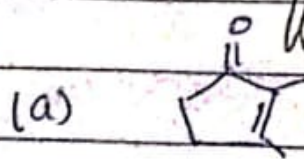
Conjugation = +30, +30

Exocyclic = +5, +5

Sub = 12 + 18 + 18 + 18
= 346 + 139

$\lambda_{max} = 385 \text{ nm}$ \rightarrow Homodiene

Q: A cyclic ketone has $\lambda_{max} = 235 \text{ nm}$
Identify its structure.



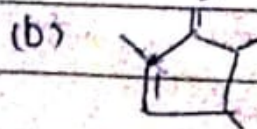
P. V = 202

Conj = 0

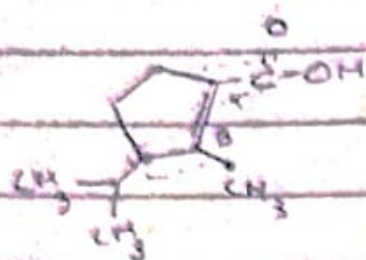
Exo = 0

Sub $\alpha = 10$ $\beta = 24$

$\lambda_{max} = 236 \text{ nm}$



-COOH



P. value = 105 nm

Sub = $\alpha = 10$ $\beta = 24$

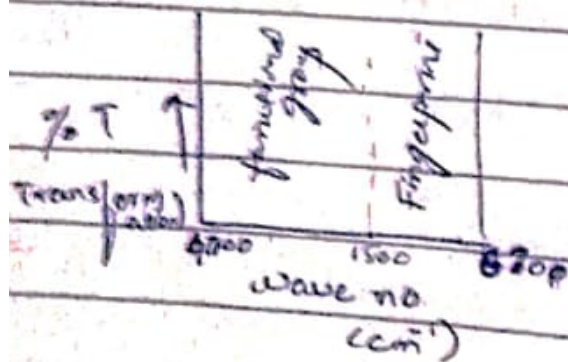
$\lambda_{max} = 229$ nm

λ_{max}

- acetaldehyde = 290 nm
- acetic acid = 208 nm
- amide $-\overset{\overset{O}{\parallel}}{C}-NH_2 = 220$ nm
- Acid chloride = 220 nm

Infrared Spectroscopy (FT-IR)

- Fourier Transform Infrared Spectroscopy
- 670-4000 cm^{-1}
- Also called vibrational spectroscopy.
Because in vibrational mode of atom molecules should be observed.



- 2- Regions
- [fingerprint region (670-1500 cm^{-1})
 - [functional group region (1500-4000 cm^{-1})