## **CHEM – 750**

# **Advanced Organic synthesis**

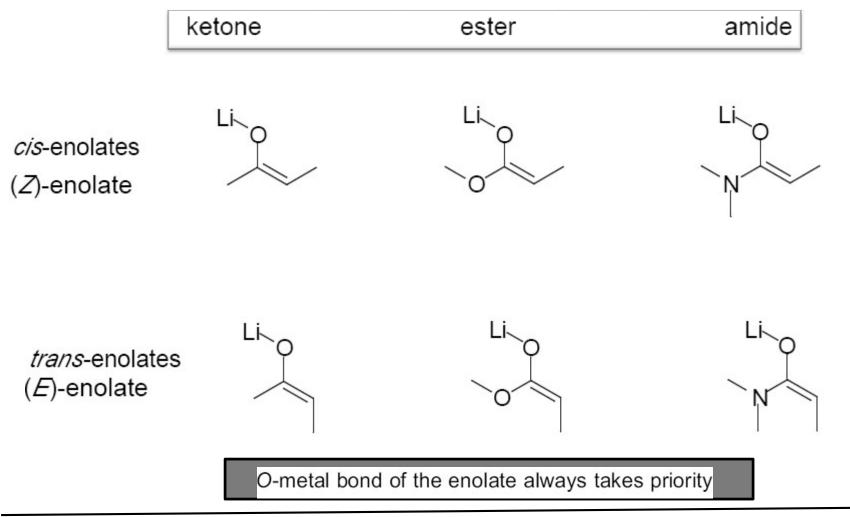


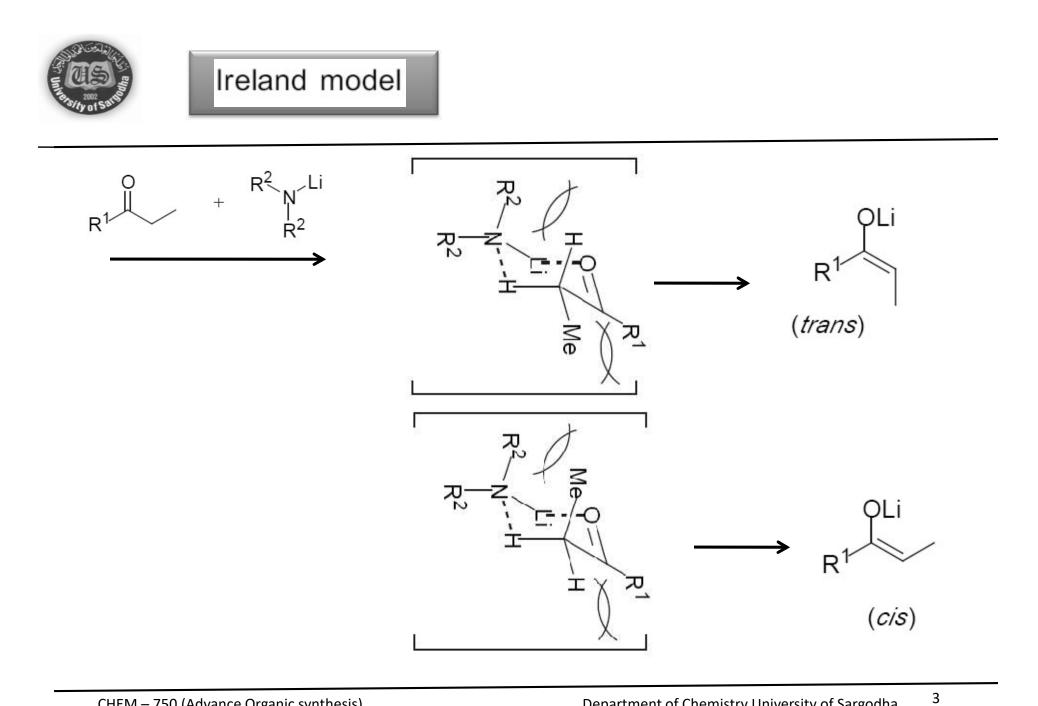
#### Dr. Humaira Yasmeen Gondal

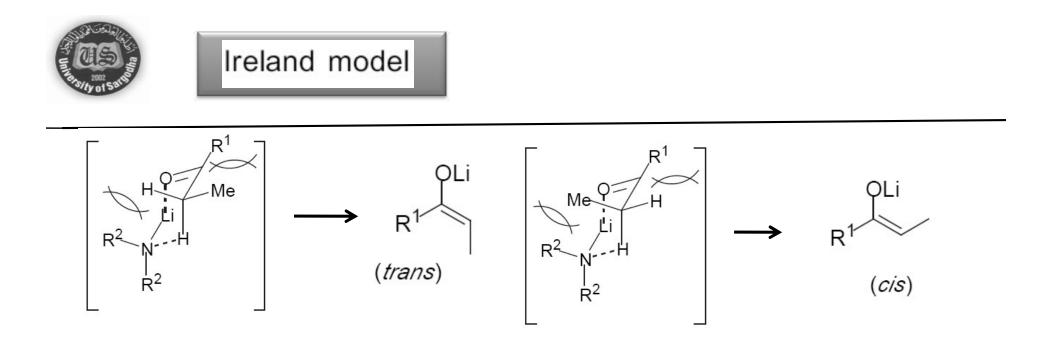
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#### **Stereoselective Enolate Formation**



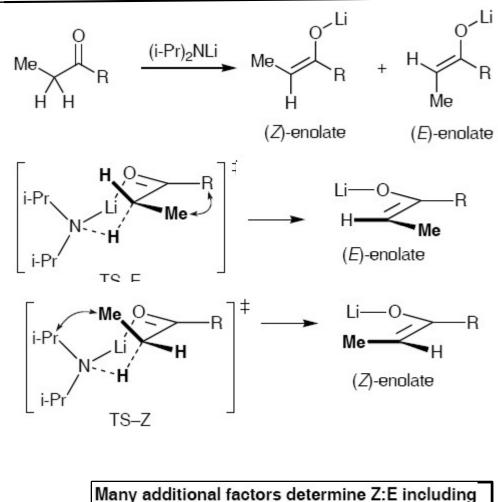




- when R<sup>1</sup> is NOT sterically demanding and when H ↔ R<sup>2</sup> << Me ↔ R<sup>2</sup> then the *trans*-stereoisomer is favoured.
- 2) when  $R^1$  is large (<sup>t</sup>Bu is especially good) then Me  $\leftrightarrow R^1 \gg H \leftrightarrow R^1$  and this overrides the Me  $\leftrightarrow R^2$  interaction favouring the formation of the *cis*-stereoisomer.



### Rationalisation: the Ireland model



As the size of R increases, the preference for (Z) increases

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R	Z: E
Et i-Pr t-Bu Ph	1:3 3:2 98:2 98:2

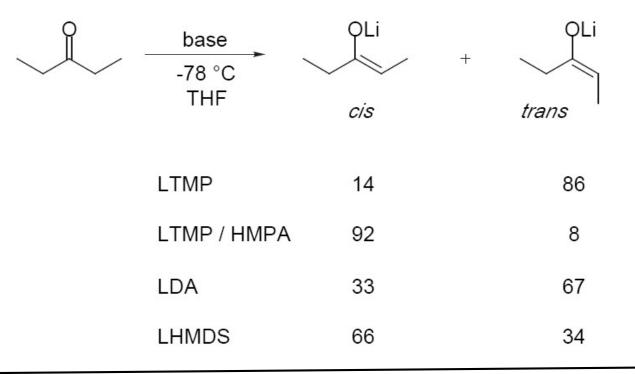
1. When R is small, the i-Pr/Me interaction in TS-Z dominates forcing the reaction to proceed via TS-E.

2. When R is large, the R/Me interaction in TS-E dominates forcing the reaction to proceed via TS-Z.

- 1. solvent
- 2. temperature
- 3. metal cation (Li, Na, K)
- 4. size of groups on the nitrogen base
- 5. additives (e.g. dipolar aprotic solvents)

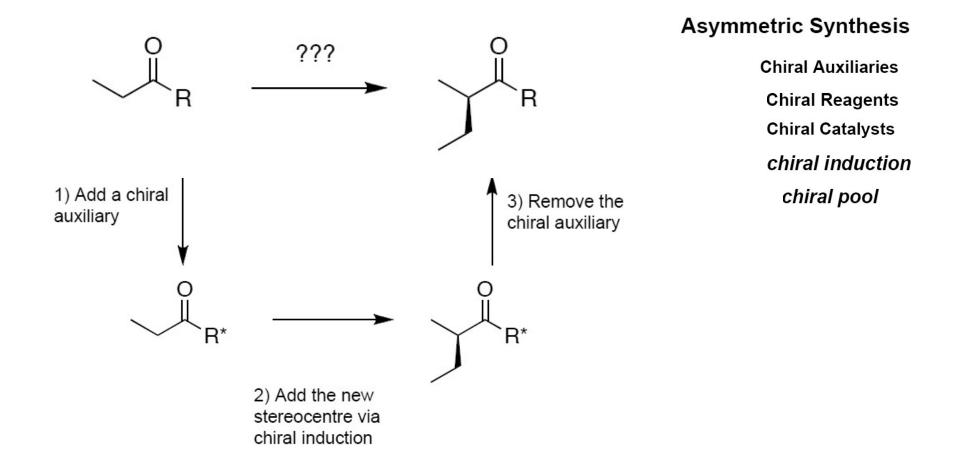


- 1) LHMDS generally provides the *cis*-enolate as the major product
- 2) LTMP (very bulky) affords the *trans*-enolate as the major product
- 3) LDA gives intermediate results.
- 4) use of HMPA as a strongly Lewis basic donor-co-solvent can reverse selectivity.





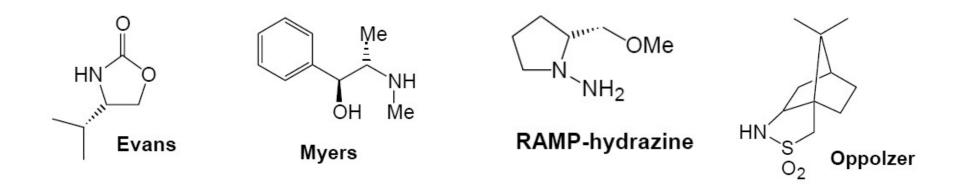
#### Asymmetric Alkylation of Enolates





#### Asymmetric Alkylation of Enolates

Chiral Auxiliary: A chiral control element temporarily incorporated into the structure of the substrate in order to direct the stereochemistry at new stereogenic centre(s) formed in a reaction. The auxiliary is removed (either immediately during work up or in a separate subsequent step) and may be recovered for re-use. Some examples are given below.





### Racemic alkylation - no chiral auxiliary present.

