# POWER PLANT ENGINEERING

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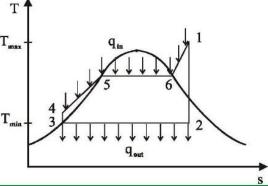
# Topics

- The Ideal Regeneration Cycle
- Feed Water Heater

3. Steam Power Plants: Cycles and Materials

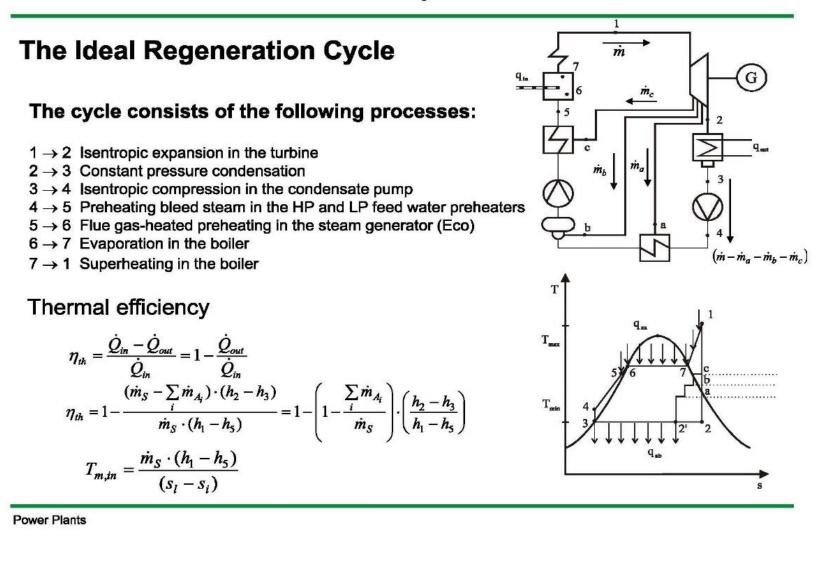
## 3.8 The Ideal Regeneration Cycle

- A careful examination of the *T-s* diagram of the Rankine cycle redrawn below reveals that heat is transferred to the working fluid during process 4-5 at a relatively low temperature.
  - This lowers the average heat addition temperature and thus the cycle efficiency.
- To remedy this, we can raise the temperature of the liquid leaving the pump (called the *feedwater*) before it enters the boiler.
- Thus, heat can be transferred from steam in a counterflow heat exchanger built into the turbine, that is, to use regeneration.
- A practical regeneration process in steam power plants is accomplished by extracting, or "bleeding," steam from the turbine at various points.

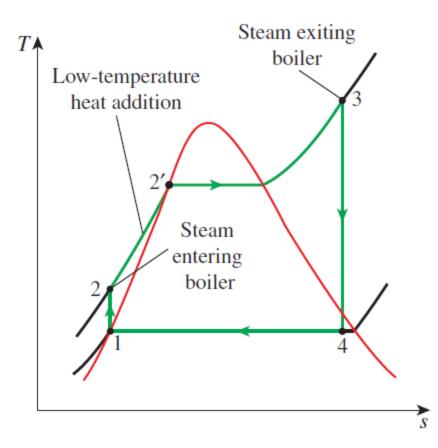


**Power Plants** 

3. Steam Power Plants: Cycles and Materials



# THE IDEAL REGENERATIVE RANKINE CYCLE



## **FIGURE 10–14**

The first part of the heat-addition process in the boiler takes place at relatively low temperatures. Heat is transferred to the working fluid during process 2-2 at a relatively low temperature. This lowers the average heat-addition temperature and thus the cycle efficiency.

In steam power plants, steam is extracted from the turbine at various points. This steam, which could have produced more work by expanding further in the turbine, is used to heat the feedwater instead. The device where the feedwater is heated by regeneration is called a **regenerator**, or a **feedwater heater** (FWH).

A feedwater heater is basically a heat exchanger where heat is transferred from the steam to the feedwater either by mixing the two fluid streams (**open feedwater heaters**) or without mixing them (**closed feedwater heaters**). Regeneration not only improves cycle efficiency, but also provides a convenient means of deaerating the feedwater (removing the air that leaks in at the condenser) to prevent corrosion in the boiler. It also helps control the large volume flow rate of the steam at the final stages of the turbine (due to the large specific volumes at low pressures). Therefore, regeneration has been used in all modern steam power plants since its introduction in the early 1920s.

# **Open Feedwater Heaters**

An open (or direct-contact) feedwater heater is basically a *mixing chamber*, where the steam extracted from the turbine mixes with the feedwater exiting the pump. Ideally, the mixture leaves the heater as a saturated liquid at the heater pressure.

$$q_{in} = h_5 - h_4$$

$$q_{out} = (1 - y)(h_7 - h_1)$$

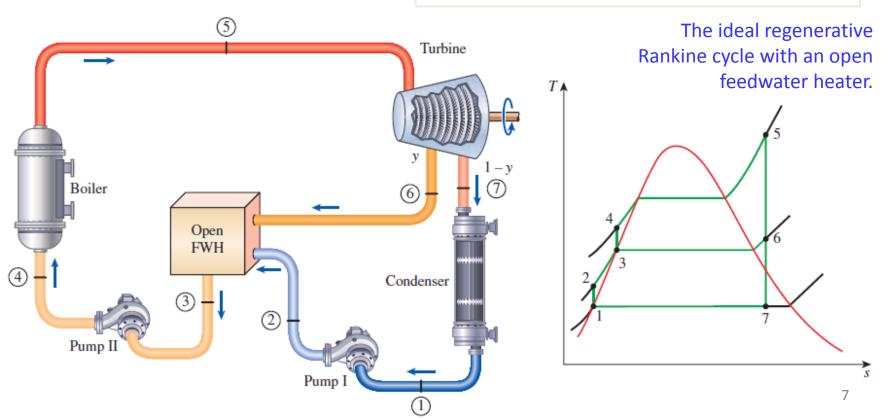
$$w_{turb,out} = (h_5 - h_6) + (1 - y)(h_6 - h_7)$$

$$w_{pump,in} = (1 - y)w_{pump I,in} + w_{pump II,in}$$

$$y = \dot{m}_6/\dot{m}_5 \qquad \text{(fraction of steam extracted)}$$

$$w_{pump I,in} = v_1(P_2 - P_1)$$

$$w_{pump II,in} = v_3(P_4 - P_3)$$



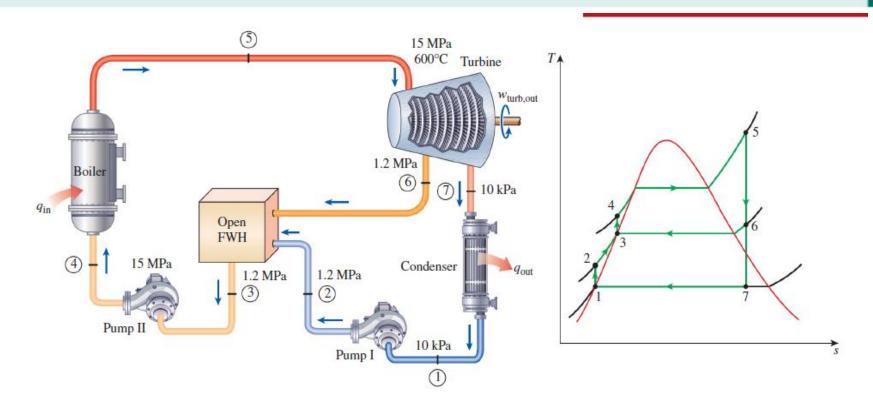
#### **EXAMPLE 10–5** The Ideal Regenerative Rankine Cycle

Consider a steam power plant operating on the ideal regenerative Rankine cycle with one open feedwater heater. Steam enters the turbine at 15 MPa and 600°C and is condensed in the condenser at a pressure of 10 kPa. Some steam leaves the turbine at a pressure of 1.2 MPa and enters the open feedwater heater. Determine the fraction of steam extracted from the turbine and the thermal efficiency of the cycle.

**SOLUTION** A steam power plant operates on the ideal regenerative Rankine cycle with one open feedwater heater. The fraction of steam extracted from the turbine and the thermal efficiency are to be determined.

**Assumptions** 1 Steady operating conditions exist. 2 Kinetic and potential energy changes are negligible.

**Analysis** The schematic of the power plant and the *T-s* diagram of the cycle are shown in Fig. 10–18. We note that the power plant operates on the ideal regenerative Rankine cycle. Therefore, the pumps and the turbines



are isentropic; there are no pressure drops in the boiler, condenser, and feedwater heater; and steam leaves the condenser and the feedwater heater as saturated liquid. First, we determine the enthalpies at various states:

State 1: 
$$P_1 = 10 \text{ kPa}$$
  
Sat. liquid  $\begin{cases} h_1 = h_{f \oplus 10 \text{ kPa}} = 191.81 \text{ kJ/kg} \\ v_1 = v_{f \oplus 10 \text{ kPa}} = 0.00101 \text{ m}^3/\text{kg} \end{cases}$   
State 2:  $P_2 = 1.2 \text{ MPa}$   
 $s_2 = s_1$   
 $w_{\text{pump I,in}} = v_1(P_2 - P_1) = (0.00101 \text{ m}^3/\text{kg})[(1200 - 10) \text{ kPa}] \left(\frac{1 \text{ kJ}}{1 \text{ kPa} \cdot \text{m}^3}\right)$   
 $= 1.20 \text{ kJ/kg}$   
 $h_2 = h_1 + w_{\text{pump I,in}} = (191.81 + 1.20) \text{ kJ/kg} = 193.01 \text{ kJ/kg}$   
State 3:  $P_3 = 1.2 \text{ MPa}$   
 $s_4 = s_3$   
 $w_{\text{pump II,in}} = v_3(P_4 - P_3)$   
 $= (0.001138 \text{ m}^3/\text{kg})[(15,000 - 1200) \text{ kPa}] \left(\frac{1 \text{ kJ}}{1 \text{ kPa} \cdot \text{m}^3}\right)$   
 $= 15.70 \text{ kJ/kg}$   
 $h_4 = h_3 + w_{\text{pump II,in}} = (798.33 + 15.70) \text{ kJ/kg} = 814.03 \text{ kJ/kg}$ 

State 5:  $P_5 = 15 \text{ MPa}$  $T_5 = 600^{\circ}\text{C}$   $h_5 = 3583.1 \text{ kJ/kg}$  $s_5 = 6.6796 \text{ kJ/kg-K}$ 

State 6:  $P_6 = 1.2 \text{ MPa}$  $s_6 = s_5$   $h_6 = 2860.2 \text{ kJ/kg}$  $(T_6 = 218.4^{\circ}\text{C})$ 

*State 7:*  $P_7 = 10 \text{ kPa}$ 

$$s_7 = s_5 \quad x_7 = \frac{s_7 - s_f}{s_{fg}} = \frac{6.6796 - 0.6492}{7.4996} = 0.8041$$
$$h_7 = h_f + x_7 h_{fg} = 191.81 + 0.8041(2392.1) = 2115.3 \text{ kJ/kg}$$

The energy analysis of open feedwater heaters is identical to the energy analysis of mixing chambers. The feedwater heaters are generally well insulated ( $\dot{Q} = 0$ ), and they do not involve any work interactions ( $\dot{W} = 0$ ). By neglecting the kinetic and potential energies of the streams, the energy balance reduces for a feedwater heater to

$$\dot{E}_{\rm in} = \dot{E}_{\rm out} \longrightarrow \sum_{\rm in} \dot{m}h = \sum_{\rm out} \dot{m}h$$

or

$$yh_6 + (1 - y)h_2 = 1(h_3)$$

where y is the fraction of steam extracted from the turbine  $(-\dot{m}_6/\dot{m}_5)$ . Solving for y and substituting the enthalpy values, we find

$$v = \frac{h_3 - h_2}{h_6 - h_2} = \frac{798.33 - 193.01}{2860.2 - 193.01} = 0.2270$$

Thus,

$$q_{in} = h_5 - h_4 = (3583.1 - 814.03) \text{ kJ/kg} = 2769.1 \text{ kJ/kg}$$
  
 $q_{out} = (1 - y)(h_7 - h_1) = (1 - 0.2270)(2115.3 - 191.81) \text{ kJ/kg}$   
 $= 1486.9 \text{ kJ/kg}$ 

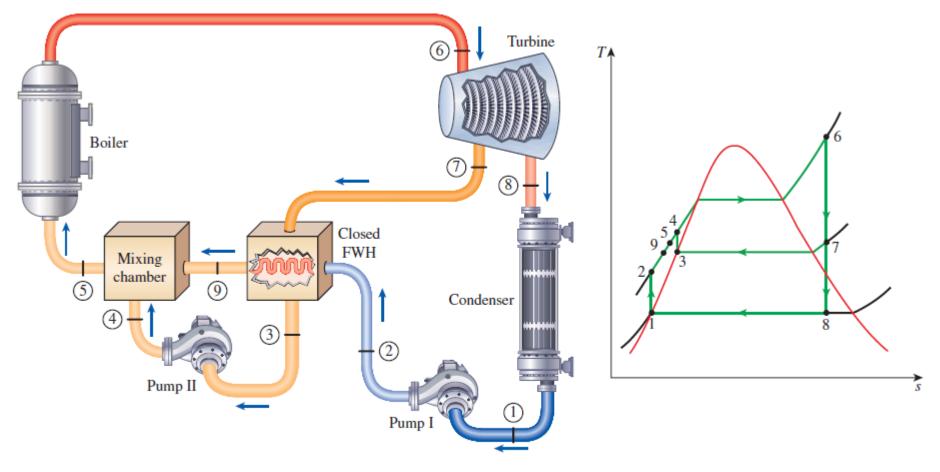
and

$$q_{\rm th} = 1 - \frac{q_{\rm out}}{q_{\rm in}} = 1 - \frac{1486.9 \,\text{kJ/kg}}{2769.1 \,\text{kJ/kg}} = 0.463 \text{ or } 46.3\%$$

**Discussion** This problem was worked out in Example 10–3*c* for the same pressure and temperature limits but without the regeneration process. A comparison of the two results reveals that the thermal efficiency of the cycle has increased from 43.0 to 46.3 percent as a result of regeneration. The net work output decreased by 171 kJ/kg, but the heat input decreased by 607 kJ/kg, which results in a net increase in the thermal efficiency.

## **Closed Feedwater Heaters**

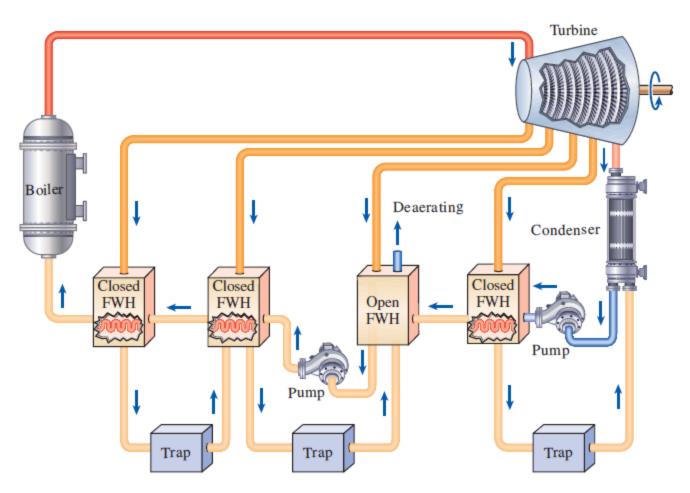
Another type of feedwater heater frequently used in steam power plants is the **closed feedwater heater**, in which heat is transferred from the extracted steam to the feedwater without any mixing taking place. The two streams now can be at different pressures, since they do not mix.



### FIGURE 10–16

The ideal regenerative Rankine cycle with a closed feedwater heater.

The closed feedwater heaters are more complex because of the internal tubing network, and thus they are more expensive. Heat transfer in closed feedwater heaters is less effective since the two streams are not allowed to be in direct contact. However, closed feedwater heaters do not require a separate pump for each heater since the extracted steam and the feedwater can be at different pressures.



Open feedwater heaters are simple and inexpensive and have good heat transfer characteristics. For each heater, however, a pump is required to handle the feedwater.

Most steam power plants use a combination of open and closed feedwater heaters.

## **FIGURE 10–17**

A steam power plant with one open and three closed feedwater heaters.

### EXAMPLE 10-6 The Ideal Reheat-Regenerative Rankine Cycle

Consider a steam power plant that operates on an ideal reheat-regenerative Rankine cycle with one open feedwater heater, one closed feedwater heater, and one reheater. Steam enters the turbine at 15 MPa and 600°C and is condensed in the condenser at a pressure of 10 kPa. Some steam is extracted from the turbine at 4 MPa for the closed feedwater heater, and the remaining steam is reheated at the same pressure to 600°C. The extracted steam is completely condensed in the heater and is pumped to 15 MPa before it mixes with the feedwater at the same pressure. Steam for the open feedwater heater is extracted from the low-pressure turbine at a pressure of 0.5 MPa. Determine the fractions of steam extracted from the turbine as well as the thermal efficiency of the cycle.

**SOLUTION** A steam power plant operates on the ideal reheat-regenerative Rankine cycle with one open feedwater heater, one closed feedwater heater, and one reheater. The fractions of steam extracted from the turbine and the thermal efficiency are to be determined.

**Assumptions** 1 Steady operating conditions exist. 2 Kinetic and potential energy changes are negligible. 3 In both open and closed feedwater heaters, feedwater is heated to the saturation temperature at the feedwater heater pressure. (Note that this is a conservative assumption since extracted steam enters the closed feedwater heater at 376°C and the saturation temperature at the closed feedwater pressure of 4 MPa is 250°C).

**Analysis** The schematic of the power plant and the *T*-s diagram of the cycle are shown in Fig. 10–19. The power plant operates on the ideal reheat–regenerative Rankine cycle and thus the pumps and the turbines are isentropic; there are no pressure drops in the boiler, reheater, condenser,

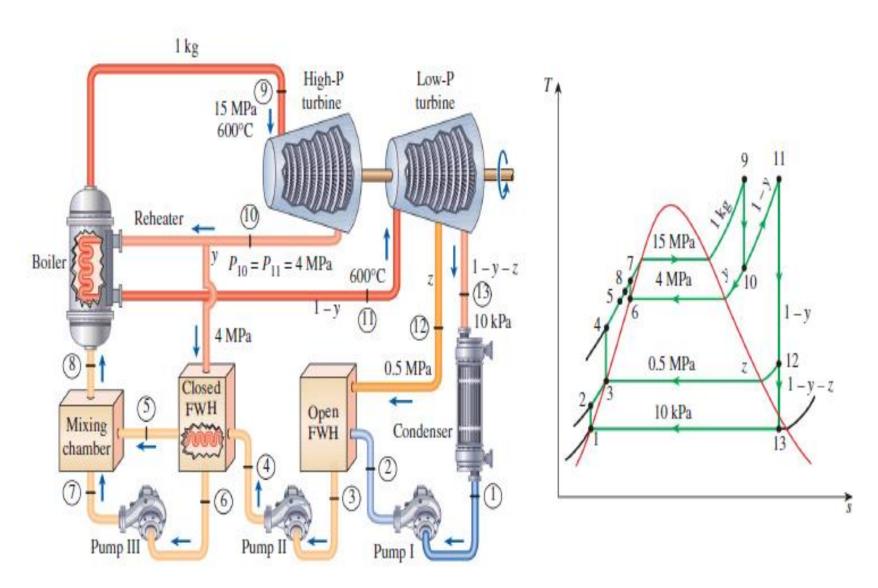


FIGURE 10–19 Schematic and *T-s* diagram for Example 10–6.

and feedwater heaters; and steam leaves the condenser and the feedwater heaters as saturated liquid.

The enthalpies at the various states and the pump work per unit mass of fluid flowing through them are

$h_1 = 191.81 \text{ kJ/kg}$	$h_9 = 3155.0 \text{ kJ/kg}$
$h_2 = 192.30 \text{ kJ/kg}$	$h_{10} = 3155.0 \text{ kJ/kg}$
$h_3 = 640.09 \text{ kJ/kg}$	$h_{11} = 3674.9 \text{ kJ/kg}$
$h_4 = 643.92 \text{ kJ/kg}$	$h_{12} = 3014.8 \text{ kJ/kg}$
$h_5 = 1087.4 \text{ kJ/kg}$	$h_{13} = 2335.7 \text{ kJ/kg}$
$h_6 = 1087.4 \text{ kJ/kg}$	$w_{\text{pump I,in}} = 0.49 \text{ kJ/kg}$
$h_7 = 1101.2 \text{ kJ/kg}$	$w_{\text{pump II,in}} = 3.83 \text{ kJ/kg}$
$h_8 = 1089.8 \text{ kJ/kg}$	$w_{\text{pump III,in}} = 13.77 \text{ kJ/kg}$

The fractions of steam extracted are determined from the mass and energy balances of the feedwater heaters:

Closed feedwater heater:

$$\dot{E}_{in} = \dot{E}_{out}$$

$$yh_{10} + (1 - y)h_4 = (1 - y)h_5 + yh_6$$

$$y = \frac{h_5 - h_4}{(h_{10} - h_6) + (h_5 - h_4)} = \frac{1087.4 - 643.92}{(3155.0 - 1087.4) + (1087.4 - 643.92)} = 0.1766$$

Open feedwater heater:

$$\dot{E}_{in} = \dot{E}_{out}$$

$$zh_{12} + (1 - y - z)h_2 = (1 - y)h_3$$

$$z = \frac{(1 - y)(h_3 - h_2)}{h_{12} - h_2} = \frac{(1 - 0.1766)(640.09 - 192.30)}{3014.8 - 192.30} = 0.1306$$

The enthalpy at state 8 is determined by applying the mass and energy equations to the mixing chamber, which is assumed to be insulated:

$$\dot{E}_{in} = \dot{E}_{out}$$
  
(1) $h_8 = (1 - y)h_5 + yh_7$   
 $h_8 = (1 - 0.1766)(1087.4) \text{ kJ/kg} + 0.1766(1101.2) \text{ kJ/kg}$   
= 1089.8 kJ/kg

Thus,

$$\begin{aligned} q_{\text{in}} &= (h_9 - h_8) + (1 - y)(h_{11} - h_{10}) \\ &= (3583.1 - 1089.8) \text{ kJ/kg} + (1 - 0.1766)(3674.9 - 3155.0) \text{ kJ/kg} \\ &= 2921.4 \text{ kJ/kg} \\ q_{\text{out}} &= (1 - y - z)(h_{13} - h_1) \\ &= (1 - 0.1766 - 0.1306)(2335.7 - 191.81) \text{ kJ/kg} \\ &= 1485.3 \text{ kJ/kg} \end{aligned}$$

and

$$\eta_{\text{th}} = 1 - \frac{q_{\text{out}}}{q_{\text{in}}} = 1 - \frac{1485.3 \text{ kJ/kg}}{2921.4 \text{ kJ/kg}} = 0.492 \text{ or } 49.2\%$$

**Discussion** This problem was worked out in Example 10–4 for the same pressure and temperature limits with reheat but without the regeneration process. A comparison of the two results reveals that the thermal efficiency of the cycle has increased from 45.0 to 49.2 percent as a result of regeneration.

The thermal efficiency of this cycle could also be determined from

$$\eta_{\rm th} = \frac{w_{\rm net}}{q_{\rm in}} = \frac{w_{\rm turb,out} - w_{\rm pump,in}}{q_{\rm in}}$$

where

$$w_{\text{turb,out}} = (h_9 - h_{10}) + (1 - y)(h_{11} - h_{12}) + (1 - y - z)(h_{12} - h_{13})$$

 $w_{\text{pump,in}} = (1 - y - z)w_{\text{pump I,in}} + (1 - y)w_{\text{pump II,in}} + (y)w_{\text{pump III,in}}$ 

Also, if we assume that the feedwater leaves the closed FWH as a saturated liquid at 15 MPa (and thus at  $T_5 = 342^{\circ}$ C and  $h_5 = 1610.3$  kJ/kg), it can be shown that the thermal efficiency would be 50.6 percent.