

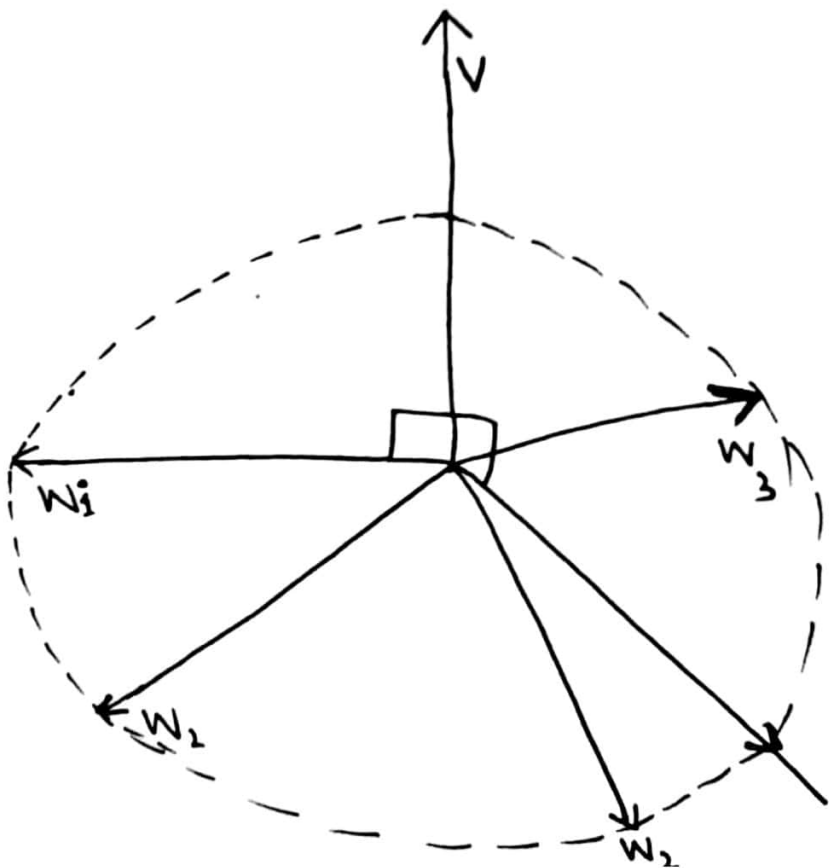
(Q.1)

Difference between orthogonal and Normalization wave function with example?

⇒ Orthogonal Wave function:

Defination:-

Means perpendicular. For perpendicular vectors their dot product is zero wave function can be regarded vectors in infinite dimensional linear vector space to represent such vectors we need an infinite dimensional space with orthogonal basic set of vectors.



Orthogonal Wave Function :-

Let us consider wave function ψ and ϕ and $\psi^* \phi^*$

$$\int \psi^* \phi \, dv = 0 \quad E_n \rightarrow \text{no degenerate} \\ E_n \neq E_n$$

Proof

$$\hat{H}\psi_n = E_n \psi_n \rightarrow (1)$$

$$\hat{H}\psi_m = E_m \psi_m \rightarrow (2)$$

$$\int_{-\infty}^{\infty} \psi_m^* \hat{H}\psi_n \, dv = \int \psi_m^* E_n \psi_n \, dv$$

$$= E_n \int \psi_m^* \psi_n \, dv$$

$$= E_n \int \psi_m^* \psi_n \, dv \rightarrow (3)$$

$$\int_{-\infty}^{\infty} (\hat{H}\psi_m)^* \psi_n \, dv = E_m \int \psi_m^* \psi_n \, dv \rightarrow (4)$$

$$\int (A\psi)^* \psi \, dv = \int \psi^* (A\psi) \, dv$$

$$E_m \int \psi_m^* \psi_n \, dv = E_n \int \psi_m^* \psi_n \, dv$$

$$(E_m - E_n) \int \psi_m^* \psi_n \, dv = 0$$

$$E_m \neq E_n$$

$$\int \Psi_m^* \Psi_n dv = 0$$

If two wave function $\Psi_1 + \Psi_2$

$$\int \Psi_1^* \Psi_2 dv = \int \Psi_1 \Psi_2^* dv = 0$$

For Example :-

$$\Psi_1 = \sin x + \Psi_2 = \cos$$

$$\int_0^{2\pi} \sin x \cdot \cos x dx = \frac{1}{2} \int_0^{2\pi} 2 \sin x \cdot \cos x$$

$$\frac{1}{2} [\cos 2x]_0^{2\pi}$$

$$\frac{1}{2} \int_0^{2\pi} \sin 2x dx = \frac{1}{2} [\cos 2x]_0^{2\pi}$$

$$\cos 2\pi = 1 = \frac{1}{4} [\cos 2x]_0^{2\pi}$$

$$\frac{1}{4} [\cos 4\pi - \cos 0]$$

$$-\frac{1}{4} \cdot 1$$

⇒ Orthogonal wave function

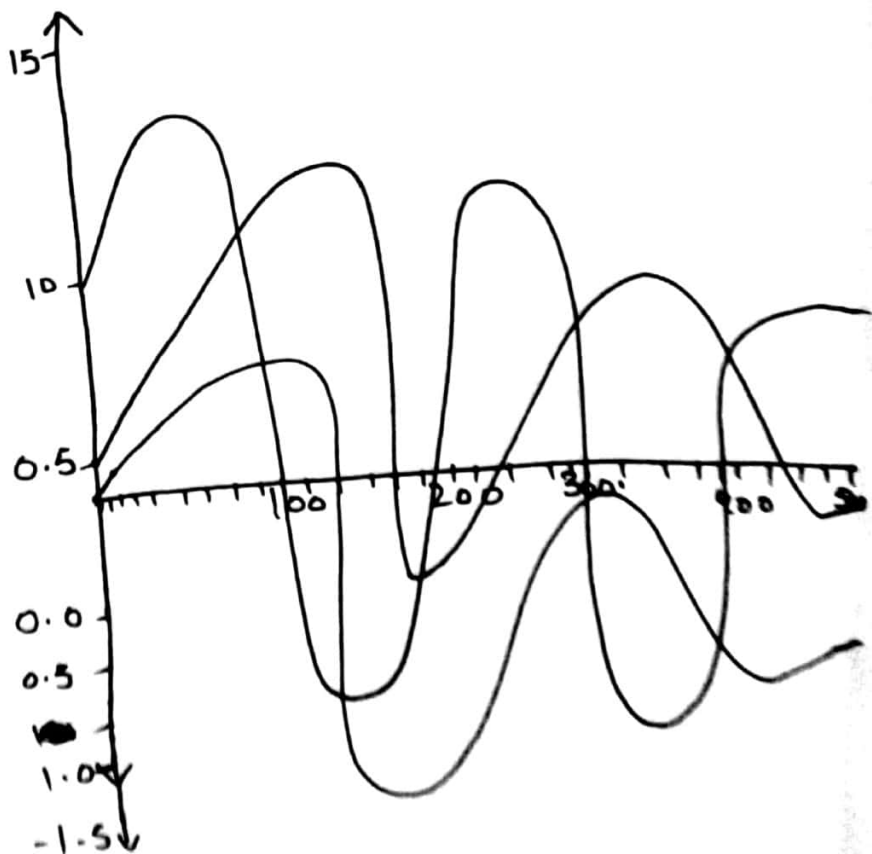
* For Example

$$\int \psi_i^* \psi_j d\tau = 0$$

Both $\sin x$ and $\sin 2x$ are eigenfunction of d/dx show if $\sin x$ and $\sin 2x$ are orthogonal.

$$\int \sin ax \sin bx dx = \frac{\sin(a-b)x}{2(a-b)} - \frac{\sin(a+b)x}{2(a+b)} + c$$

$$\int_0^{2\pi} \sin x \sin 2x dx = \frac{\sin(1-2)x}{2(1-2)} - \frac{\sin(1+2)x}{2(1+2)} + c = 0$$



Expectation value-

$$\langle x \rangle = \int \psi_i^* x \psi_j d\tau$$

* Calculate the average value of the distance of e^- from the h of H-atom

$$\Psi_{H-e} = \left(\frac{1}{\pi a_0^3} \right)^{1/2} e^{-r/a_0}$$

$$\langle r \rangle = \int \Psi^* r \Psi d\tau = \frac{1}{\pi a_0^3} \int_0^\infty r^3 e^{-2r/a_0} dr \int_0^\pi \sin\theta d\theta \int_0^{2\pi} d\phi$$

$$\langle r \rangle = \int \Psi^* r \Psi d\tau = \frac{1}{\pi a_0^3} \int_0^\infty r^3 e^{-2r/a_0} dr \int_0^\pi \sin\theta d\theta \int_0^{2\pi} d\phi$$

$$\int_0^\infty x^n e^{-ax} dx = \frac{n!}{a^{n+1}}$$

$$\frac{1}{\pi a_0^3} \frac{3! a_0^4}{2^4} \times 2 \times 2\pi = \frac{3}{2} a_0 = 79.4 \text{ pm}$$

Uncertainty Principle:-

Uncertainty in position along an axis

uncertainty in linear momentum

uncertainty the calculate the $\Delta p \Delta q \geq \frac{1}{2} \hbar$
 minimum uncertainty in the position of
 mass 1.0 g and the speed is known with
 1 um s^{-1}

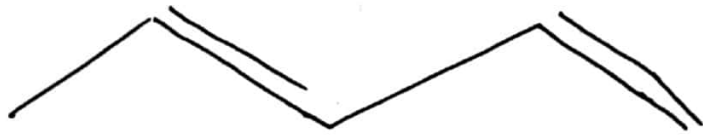
$$\Delta q \geq \frac{\hbar}{2\Delta p} = \frac{\hbar}{2m\Delta v}$$

$$\geq \frac{1.055 \times 10^{-34} \text{ Js}}{2(1.0 \times 10^{-3} \text{ kg})(1 \times 10^{-6} \text{ ms}^{-1})}$$

$$\geq 5 \times 10^{-26} \text{ m}$$

Probability (particle in box)

Wave function of conjugated electron of Polyene can be approximated by PAB. Find probability of locating electron between $x = 0$ and $x = 0.2 \text{ nm}$ in the lowest state in conjugated molecule of length 1.0 nm



$$\int_0^L \Psi_n^2 dx = \frac{2}{L} \int_0^L \frac{\sin^2 n\pi x}{L} dx$$

$$= \frac{1}{L} - \frac{1}{2n\pi} \frac{\sin 2\pi n x}{L}$$

$$0.05 = \text{when } n = 1, L = 1.0 \text{ and } x = 0.2$$

⇒ Normalization of the wave function

Normalization is the scaling of wave functions so that all the probability add to 1. A normalized wave function would be said to be normalized if it is not 1 and is instead equal to some other constant we incorporate that constant into the wave function to normalize.

$$\int_0^a \Psi^2 dx = 1$$

Putting the value of Ψ .

$$\int_0^a \left(A \sin \frac{n\pi x}{a} \right)^2 dx = 1$$

$$A^2 \int_0^a \sin^2 \left(\frac{n\pi x}{a} \right) dx = 1$$

$$\text{Since, } \sin^2 \theta = \left(\frac{1 - \cos 2\theta}{2} \right)$$

$$\text{where, } \theta = \left(\frac{n\pi x}{a} \right)$$

$$\frac{1}{2} A^2 \int_0^a \left(1 - \cos \frac{2n\pi x}{a} \right) dx = 1$$

$$\Psi^*(x) \Psi(x)$$

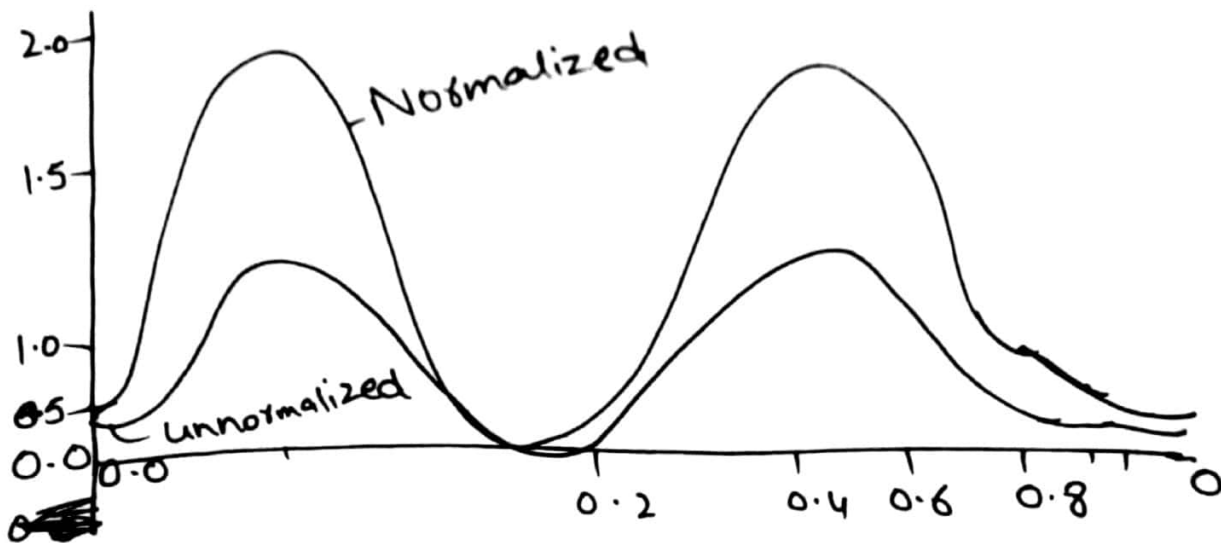
$$\int_{-\infty}^{\infty} \Psi^*(x) \Psi(x) dx = 1$$

⇒ Example:-

$n = 2$ state for a particle in a box of length $L = 1$

Unnormalized = $\Psi(x) = \sin(2\pi x)$

$\Psi(x) = 2\sqrt{\sin(2\pi x)}$



* Unnormalized - probability sum to 0.5 un-physical.

* Normalized - probability sum to 1.0 physical

$$\int_{-\infty}^{\infty} \Psi^*(x) \Psi(x) dx = \int_0^1 \sin(2\pi x) [\sin(2\pi x)] dx$$

$\frac{1}{2}A^2$ is constant is taken outside the integration sign.

$$\frac{A^2}{2} \left[x - \frac{a}{2n\pi} \sin \frac{2n\pi x}{a} \right]_0^a = 1$$

Putting the value of limits,

$$\frac{A^2}{2} \left[a - \frac{a}{2n\pi} \sin 2n\pi \right] = 1$$

$$\frac{A^2 a}{2} = 1$$

$$A = \sqrt{\frac{2}{a}}$$

The expression for Eigen function is

$$\psi = \sqrt{\frac{2}{a}} \cdot \sin \left(\frac{n\pi x}{a} \right)$$

Similarly probability for finding a particle for a particular of "x" is given by taking the square ψ^2 :

$$\psi^2 = \frac{2}{a} \cdot \sin^2 \left(\frac{n\pi x}{a} \right)$$

$$\int_0^1 \sin^2(2\pi x) dx$$

$$\frac{1}{2\pi} \int_0^{2\pi} \sin^2 u du$$

$$\frac{1}{2\pi} (\pi) = \frac{1}{2} \text{ Not normalized}$$

=> Example:-

Normalize a wave function for a particle in a box of length L given by.

$$\Psi(x) = x(L-x)$$

$$\int \Psi^*(x) \Psi(x) dx = \int_0^L [A x(L-x)] [A x(L-x)] dx$$

$$A^2 \int_0^L x^2 (L-x)^2 dx = A^2 \int_0^L x^4 - 2Lx^3 + L^2 x^2 dx$$

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$$A^2 \left[\frac{x^5}{5} - \frac{2Lx^4}{4} + \frac{L^2 x^3}{3} \right]_{x=0}^{x=L}$$

$$\frac{A^2 L^2}{30} = 1 \text{ if normalized}$$

$$\Rightarrow A \frac{\sqrt{30}}{L^2} = \Psi(x) = \frac{\sqrt{30}}{L^2 x(L-x)}$$

Our final normalized wave function