

# CHEM – 647

## Organic synthesis

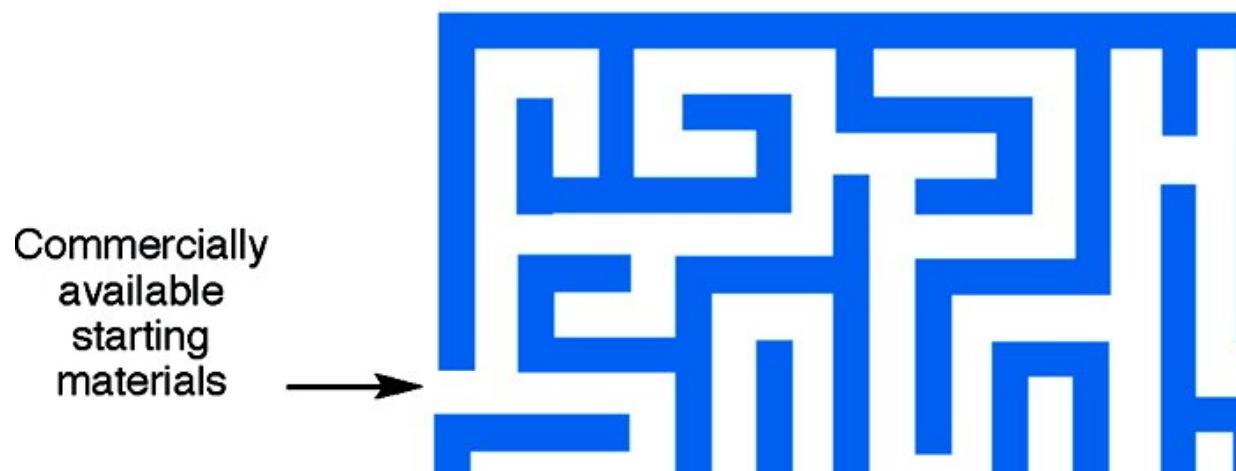


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# RETROSYNTHETIC ANALYSIS

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The ultimate goal of Organic Synthesis is to assemble an organic compound (target) from readily available starting materials and reagents in the most efficient way.

This process usually begins with the design of a synthetic plan (Strategy)



# Ideal Synthesis

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## ***Economically acceptable***

*Cost effective: Less expensive starting materials*

## ***Environmentally acceptable***

*Green route: avoid wastes side products and toxic intermediates*

## ***Safe***

*Avoid risky and hazardous procedures*

## ***Robust***

*Easily scale-up and reproducible*

## ***Highly efficient***

*Simple, high yield, a few steps*



# Requirements for Ideal Synthesis

1. *Knowledge of Reactivity (Structure-Mechanism)*

*Tactics*

2. *Design ability (Retrosynthetic Analysis)*

*Strategy*

## **LEFT BRAIN**

Logic  
Analysis  
Organization  
Knowledge /Facts  
Detail  
Maths & Science



## **RIGHT BRAIN**

Intuition  
Emotion  
Spirituality  
Belief  
Big picture  
Art / Music

***Synthesis is logical as well as artistic activity at the same time. It involves freedom, imagination, creativity, courage, persistence, knowledge and skills***

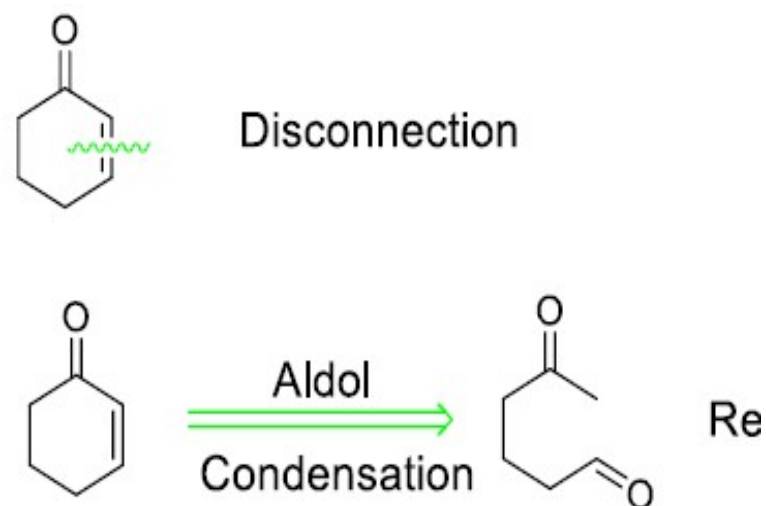


# RETROSYNTHETIC ANALYSIS

Retrosynthetic analysis is a problem solving technique for transforming the structure of a synthetic target molecule (TM) to a sequence of progressively simpler structures along a pathway which ultimately leads to simple or commercially available starting materials for a chemical synthesis.

▪ Reverse of Synthesis is the process of breaking down the TM into available starting materials by FGI and disconnection

▪ Disconnection is reverse operation to a reaction: An imaginary cleavage of a bond to break the molecule into starting materials (↔)





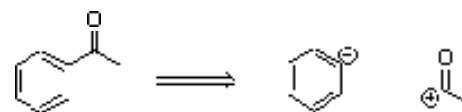
# RETROSYNTHETIC ANALYSIS

## Disconnection

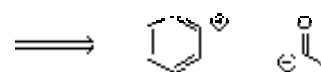


## Synthon

*An idealised fragment resulting from disconnection, usually cation or anion*



Looks possible because it implies electrophilic attack on benzene ring

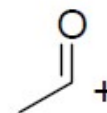


Does not look so good (nucleophilic attack less usual)

## Reagent

*compound used in practice for a synthon.*

Synthon



## Synthetic Equivalent

*Actual substrates used for the forward synthesis*

Synthon

R<sup>+</sup>

F



# RETROSYNTHETIC ANALYSIS

Acceptor synthon :

carbocation (electrophilic)

Common Acceptor Synthon

Synthetic equivalents

$R^+$  (alkyl cation = carbenium ion)

$RCI, RB$

$RC^+=O$  (acylium ion)



$CH_2=C$

$C^+H_2OH$  (oxocarbenium ion)





# RETROSYNTHETIC ANALYSIS

## Donor synthon :

## Carbanion (nucleophilic)

### Common Donor Synthon

### Synthetic equivalents

$R^-$  (alkyl, aryl anion)

$RMgX, RLi, R_2Mg$

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$RC\equiv C^-$  (acetylide)

$RC\equiv CMgX, RC\equiv CLi$        $RC\equiv CH$



enolate

carbonyls





# RETROSYNTHETIC ANALYSIS

## FGI-Functional Group Interconversion

Functional group interconversion (FGI) describes a process of converting one functional group to another: e.g. an alcohol to an aldehyde, alkyne to alkene etc.

It is the process of writing one functional group for another to help synthetic planning and to help disconnection. Note, there must be a good reaction in the reverse (forward!) direction.

