SULFURIC ACID SOLFORIC ACID



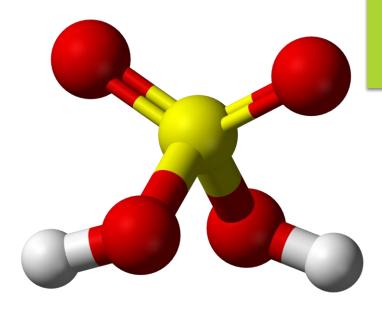
SULFURIC ACID

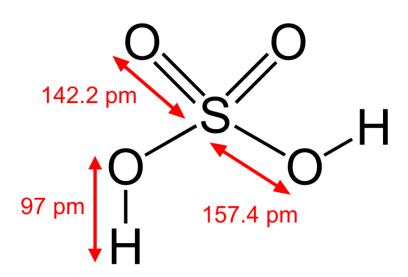
- Sulfuric acid has a yellow oily liquid
- It is soluble in <u>water</u> with release of heat
- It is corrosive to metals
- It is used to make fertilizers
- Sulfuric acid is a very strong acid
- Its chemical formula is <u>H2SO4</u>
- lts average mass is 98.078 amu



Sulfuric Acid 2D Structure

Sulfuric Acid 3D Structure



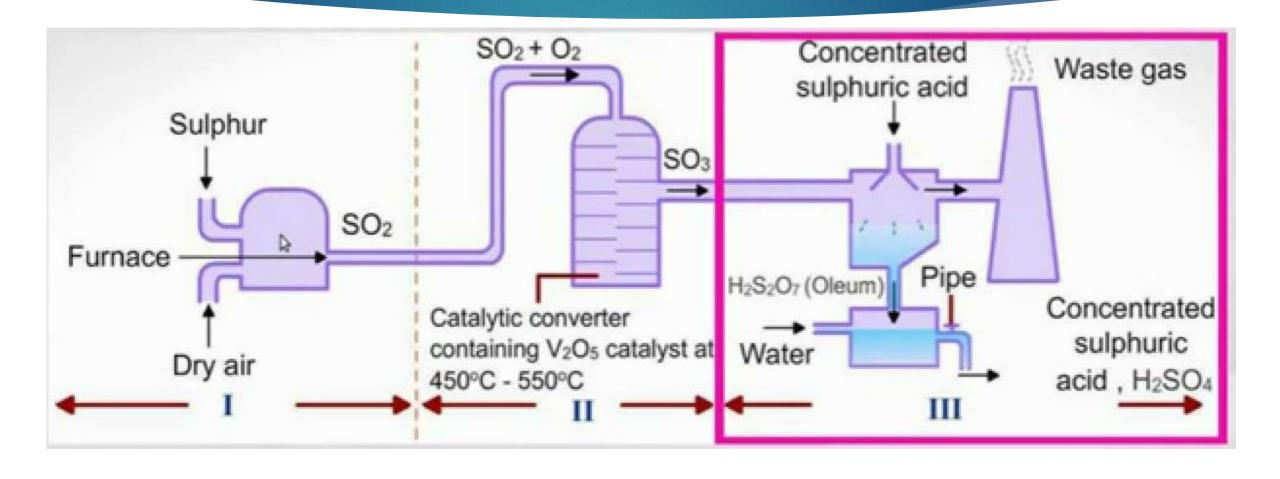


MANUFACTURING PROCESS OF SULFURIC ACID

MANUFACTURING PROCESS TYPES

- ► There are Two Types of MANUFACTURING Processes
- ▶ 1) Contact Process
- 2) Lead Chamber Process

MANUFACTURING CHART



DETAILS OF CONTACT PROCESS

1)PREPARATION OF SO₂.

 SO_2 is obtained by burning sulphur or by heating iron pyrite (FeS₂) in pyrite burner.

$$S + O_2 \longrightarrow SO_2$$

 $4FeS_2 + 11O_2 \longrightarrow 2Fe_2O_3 + 8SO_2$

2) PURIFICATION OF SO₂.

DUST CHAMBER:

 SO_2 is first passed through the dust chamber where steam is spread over the gas to remove dust particles, which settle down. Fe(OH)₃ also sprayed over to remove oxides of Arsenic.

WASHING TOWER:

SO₂ is then passed through a washing tower after cooling. Here it is sprayed by water to remove any other soluble impurities.

DRYING TOWER:

The gas is now dried by passing through drying tower where conc. H_2SO_4 (dehydrating agent) is sprayed. H_2SO_4 removes moisture from SO_2

DETAILS OF CONTACT PROCESS

3)OXIDATION OF SO₂ TO SO₃

CONTACT TOWER:

Oxidation of SO_2 is carried out in contact tower where V_2O_5 is filled in different pipes. SO_2 here reacts with air (O_2) to produce SO_3 . Under above conditions 98% SO_2 is converted into SO_3 .

$$2SO_2 + O_2 \longrightarrow 2SO_3 + 45Kcal$$

CONDITIONS NECESSARY FOR MAXIMUM YIELD OF SO₃:

Oxidation of SO_2 is a reversible and exothermic process in which volume of product is less than the volumes of reactants. In order to obtain maximum amount of SO_3 ,

4) ABSORPTION OF SO₃ IN H₂SO₄

 SO_3 is not directly passed in water, because a dense fog of minute particles of H_2SO_4 is produced. It is therefore, dissolved in conc. H_2SO_4 to form pyrosulphuric acid (oleum).

$$SO_3 + H_2SO_4 \longrightarrow H_2S_2O_7(OLEUM)$$

5) DILUTION OF OLEUM

Oleum is now diluted with water to form H₂SO₄ of required concentration.

$$H_2S_2O_7 + H_2O \longrightarrow 2H_2SO_4$$

PHYSICAL PROPERTIES OF SUPHURIC ACID

PHYSICAL PROPERTIES

- ► Its specific gravity is 1.834
- Its freezing point is 10.5°C
- It is very corrosive to skin and burns the animal tissues

PHYSICAL PROPERTIES OF SUPHURIC ACID

PHYSICAL PROPERTIES

- Its boiling point is 315 338 `C
- Its melting point is 10.35°C
- Its heat capacity is 1.414j/Kj

CHEMICAL PROPERTIES OF SUPHURIC ACID

REACTION WITH METALS

If sulfuric acid is react with metals then it produces hydrogen gas.

$$H_2SO_4 + Z_n \longrightarrow ZNSO_4 + H2O$$

$$H_2SO_4 + Mg \longrightarrow MgSO_4 + H2O$$

$$H_2SO_4$$
 +Fe FeSO₄ +H2O

CHEMICAL PROPERTIES OF SUPHURIC ACID

REACTION WITH GASES

▶ Gases such as NH3 and SO3 react with H2SO4 to give additional products

► 2NH3 + H2SO4 — (NH4)2SO4

CHEMICAL PROPERTIES OF SUPHURIC ACID

H₂SO₄ As an Oxidizing Agent

► H₂SO₄ is a strong Oxidizing Agent

USES OF SULFURIC ACID

USES

- ▶ By far the largest amount of sulfuric acid is used to make phosphoric acid, used, in turn, to make the <u>phosphate fertilizers</u>, calcium dihydrogen phosphate and the ammonium phosphates. It is also used to make ammonium sulfate, which is a particularly important fertilizer
- Sulfuric Acid is used as a laboratory reagent

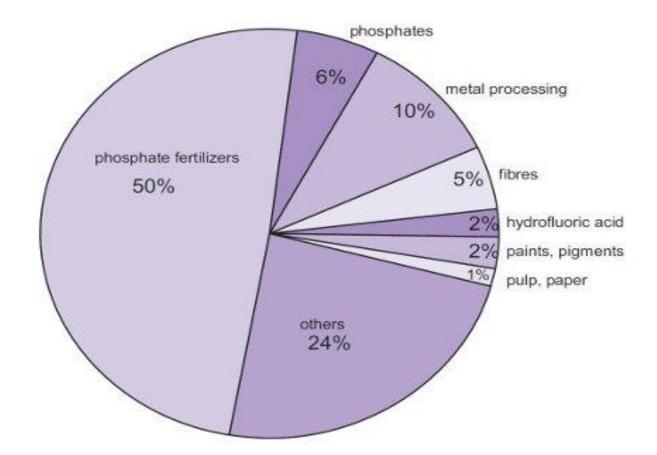
USES OF SULFURIC ACID

USES

- ► It is used as an oxidizing agent
- It is used in refining of petroleum
- ▶ It is Used in manufacturing of many chemicals, dyes, drugs, paints etc
- ▶ It is Used in prepration of explosives
- It is used in manufacturing of Fertilizers

USES OF SULFURIC ACID

CHART



ANNUAL PRODUCTION OF SULFURIC ACID

World	231 million tones
China	74 million tones
U.S.	37 million tones
India	16 million tones
Russia	14 million tones
Morocco	7 million tones

List of industries producing sulfuric acid

- Exide Pakistan
- Tufail Chemicals
- Siraj chemicals
- Itehad Chemicals
- Internation Chemicals
- Hazara fertilizers
- Amber Chemicals
- Akbari Chemicals

MANUFACTURING OF SULFURIC ACID IN PAKISTAN

▶ Following are the industries that produces sulfuric acid in Pakistan

Exide Pakistan 16500 MTPA

Tufail Chemicals 20000 TPA

Siraj chemicals 75000 STPA

Itehad Chemicals 33000 MTPA

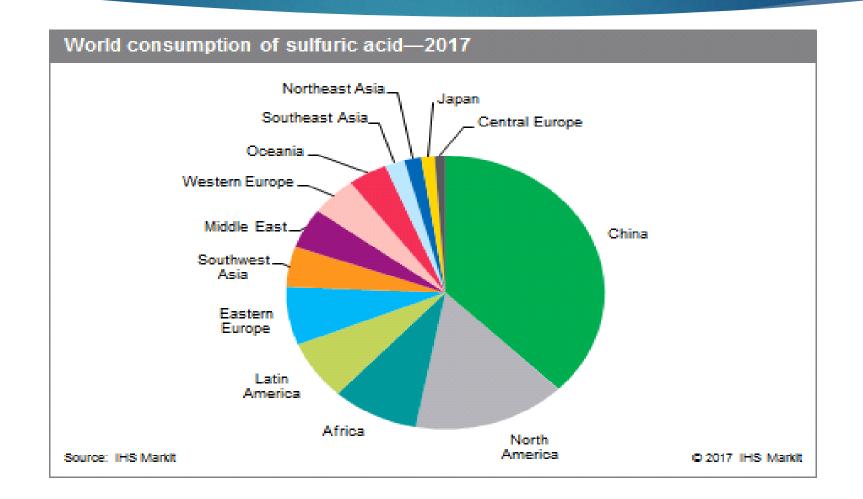
Internation Chemicals 300 MTPD

Hazara fertilizers 110 MTPD

Amber Chemicals 50 MTPD

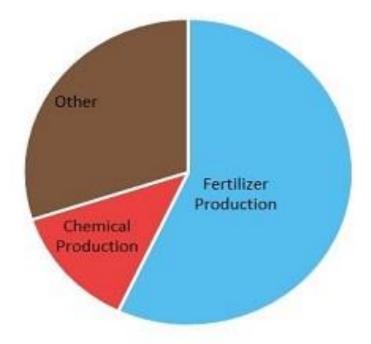
Akbari Chemicals 25 MTPD

GLOBAL MANUFACTURING OF SULFURIC ACID



GLOBAL MANUFACTURING OF SULFURIC ACID

Global Sulfuric Acid Demand by Application



THE END