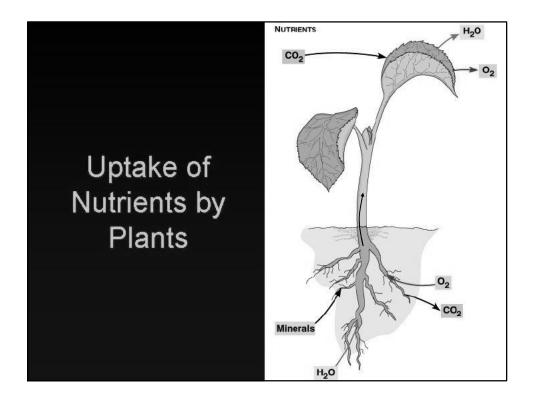
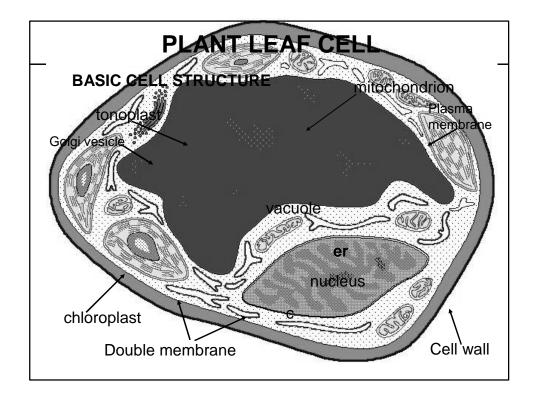
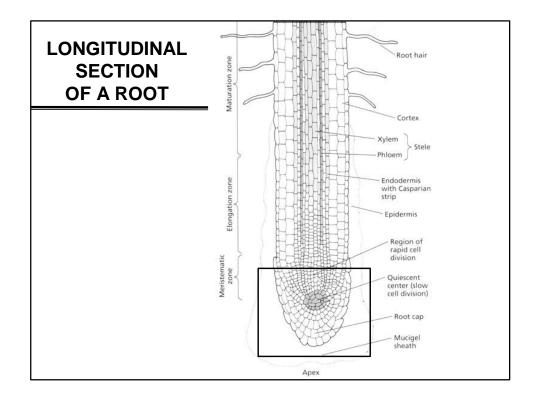
# **BIOL 695**

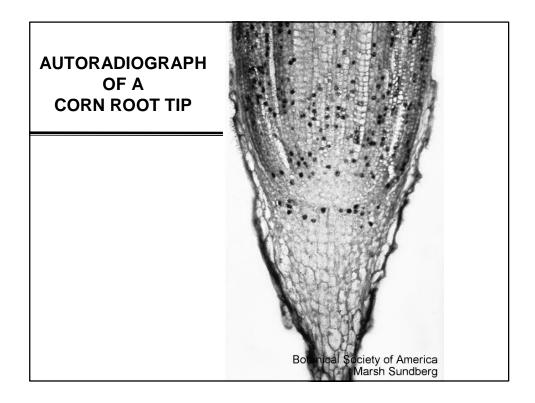
# NUTRIENT UPTAKE AND ASSIMILATION

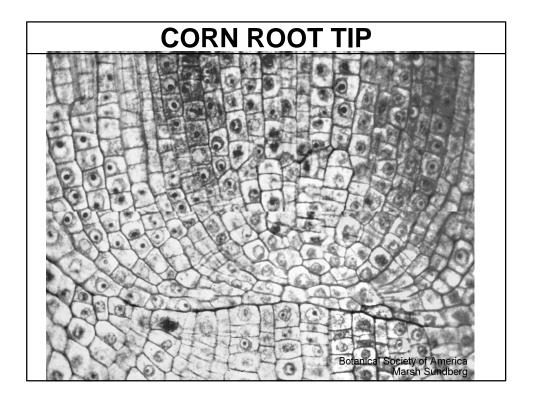
# CHAPTER 3: Mengel et al., 5th Ed





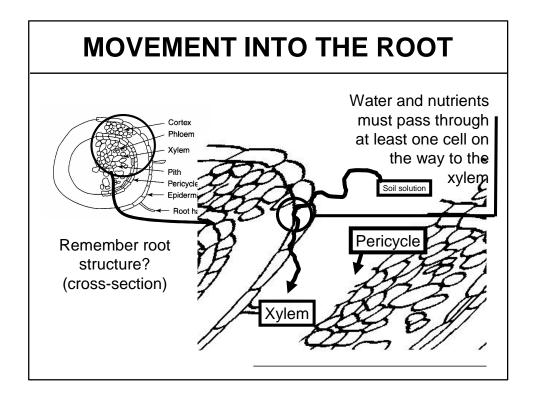


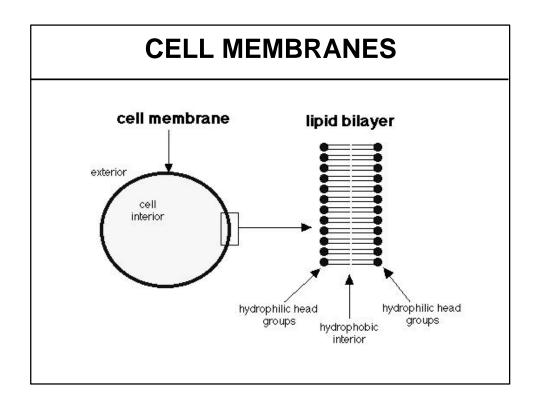


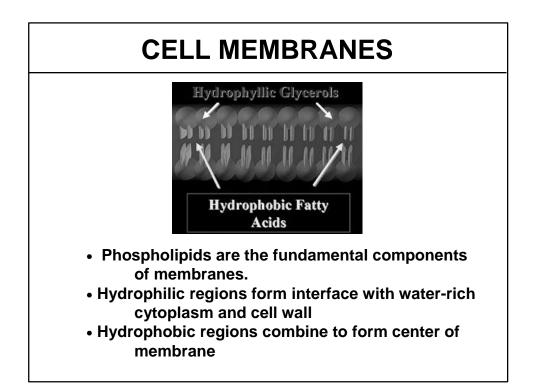


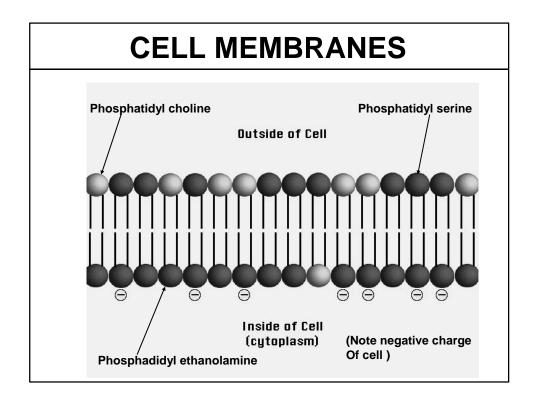
## **MINERAL UPTAKE BY PLANTS**

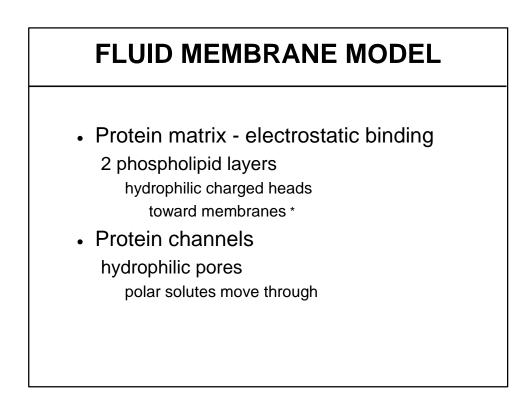
- Nutrient ions must be dissolved in soil water ("soil solution") for uptake by plants
- They move from "soil solution" to vascular center of plant root passing through at least one cell membrane (the "skin" that hold the cell's liquid contents inside)
- This movement, across the membrane, may be passive or active





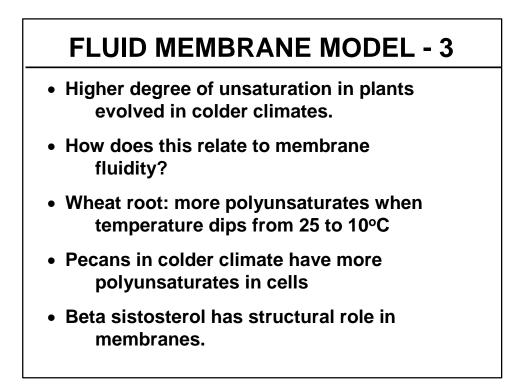


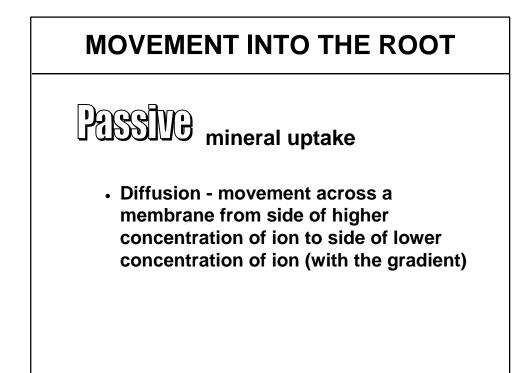


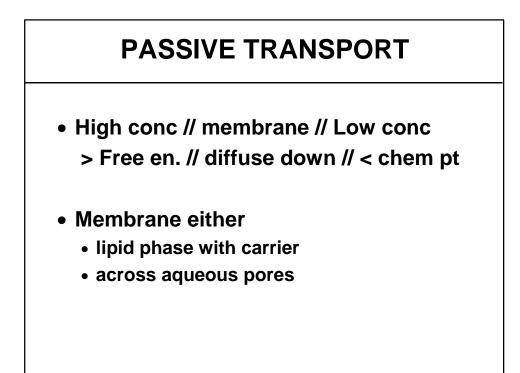


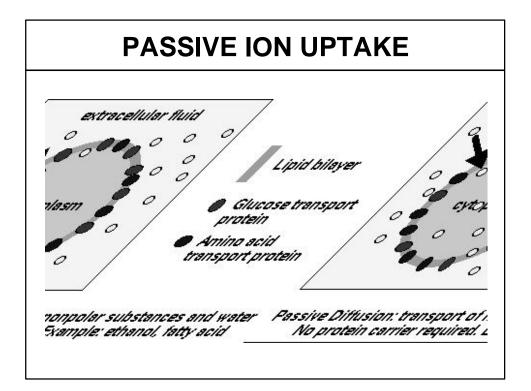
### **FLUID MEMBRANE MODEL - 2**

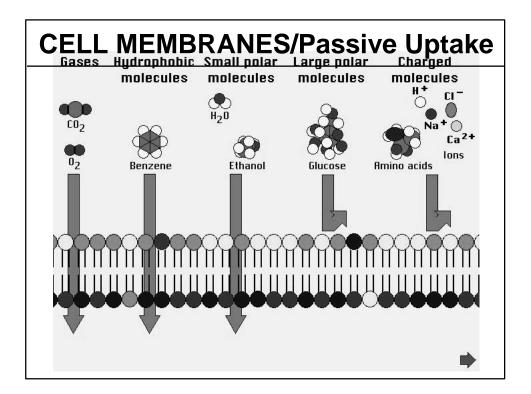
- Membranes also composed of polar lipids.
   Phospholipids \*
   Glucolipids
   Sulfolipids
- Long chain fatty acids are the hydrophobic tails oriented inwardly
- Variation in length & degree of unsaturation influence the melting point. \*

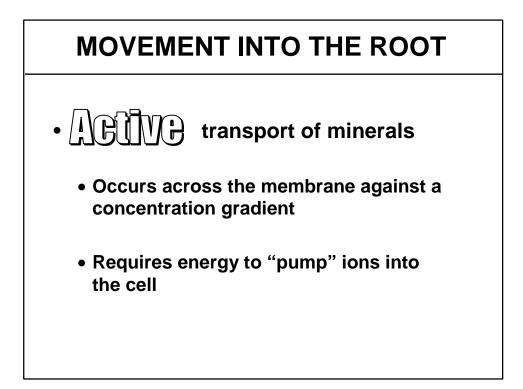


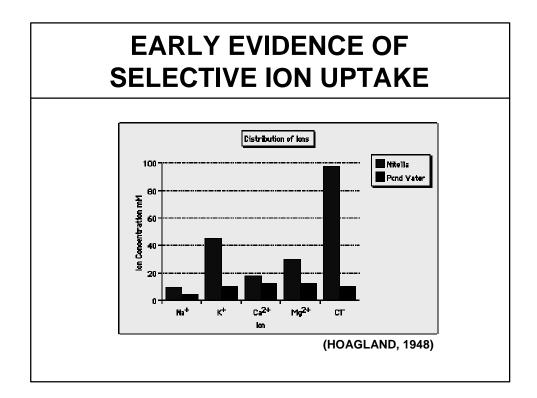






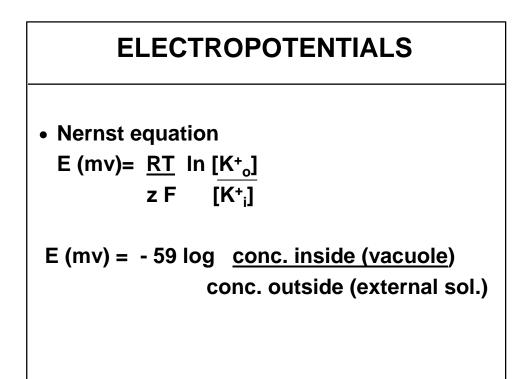






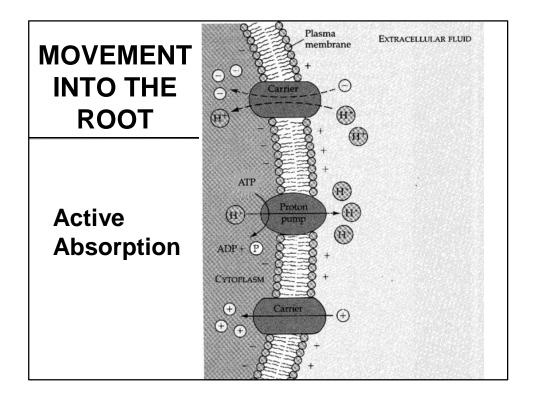
## **ACTIVE TRANSPORT**

- Transport uphill against energy gradient. (Pump in membrane)
  - Chemical Potential Gradient
    - Conc of ions on either side of membrane
  - Electrical Potential Gradient
    - Differential of milivolts across membrane





- K<sup>+</sup> in equilibrium if conc. in vacuole is 10 times > than in external solution.
- CI- in equilib if conc in vacuole is 10 times < than in external solution.
- Divalent ions are 100 times or more different between vacuole and external solution



# Two Classes of ATP Driven Proton Pumps

- Plasma membrane H<sup>+</sup>-ATPase
  - Stimulated by monovalent cations
    - K<sup>+</sup> > NH<sub>4</sub><sup>+</sup> > Na<sup>+</sup>
    - Insensitive to anions
- Tonoplast H<sup>+</sup>-ATPase
  - Insensitive to monovalent cations
  - Stimulated by anions

# Two Classes of ATP Driven Proton Pumps, con't

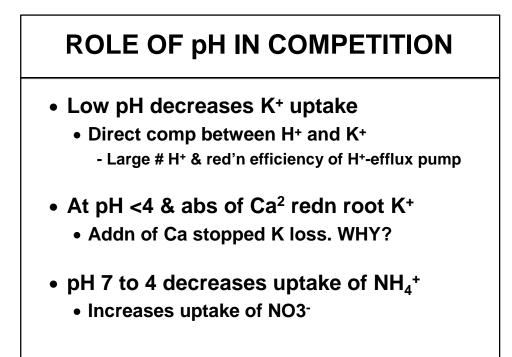
- These 2 classes of proton pumps present in all vacuolated cells
- Plasma membrane H+-ATPase
  - Light dependent in leaves
  - May be resp light stimulated K up
- Main driving force -
  - H+-ATPase Fig. 3.10

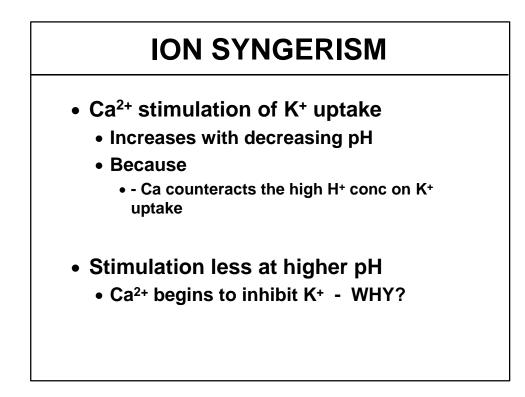
### **COMPETITION BETWEEN IONS**

- Comp between <sup>42</sup>K and K
- Comp between K<sup>+</sup> and Rb<sup>+</sup>
- Ca <sup>2+</sup> does not comp with K<sup>+</sup>
- NH<sub>4</sub><sup>+</sup> comp with K<sup>+</sup> but not reverse
  - Conversion to NH<sub>3</sub> leaves H<sup>+</sup> comp K<sup>+</sup>
- Mg<sup>2+</sup> binding weak
  - Replaced by Ca<sup>2+</sup> and Mn<sup>2+</sup>

### **ANION COMPETITION**

- Cl<sup>-</sup> can reduce excessive NO<sub>3</sub><sup>-</sup> accumulation in spinach, barley
- In saline soils, Cl<sup>-</sup> may impair N nutrition by reducing NO<sub>3</sub><sup>-</sup> uptake





# **ION UPTAKE ALONG ROOT**

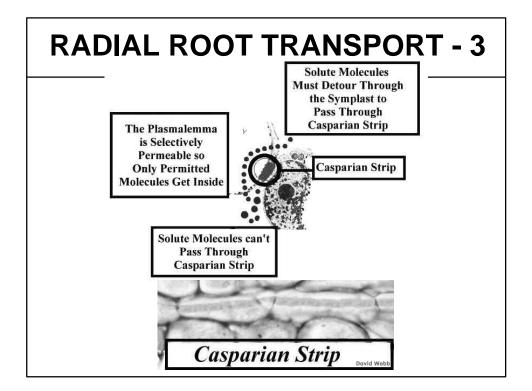
- Higher uptake in apical zone
  - Increase in suberin in older roots
  - Formation of barriers to radial translocation
- Only chance for Ca because lack of phloem mobility of Ca
- Ions react differently Fe apical
  - Basal roots absorb more P than apical
- Apical cells use more K than basal cells

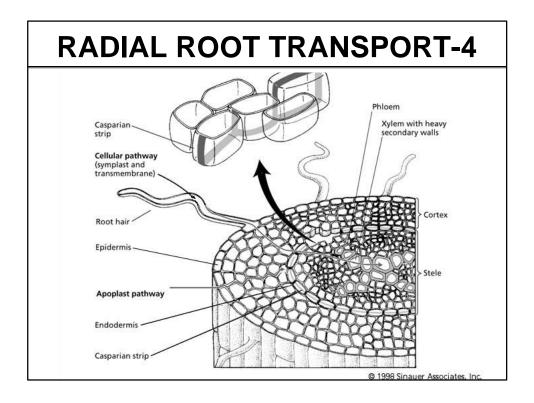
### **RADIAL ROOT TRANSPORT -1**

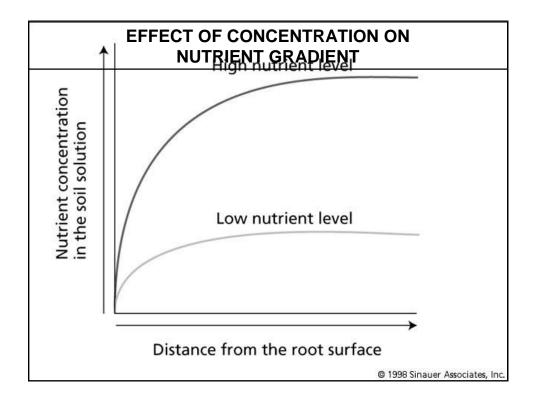
- Apoplastic
  - lons move through cell walls & intercellular spaces
    - - Exception to termination at Casparian strip
      - (forms in few days old roots)
      - \* Apical areas suberization increases
      - \* Basal areas lateral root development
  - Leaky areas important for Ca, AI, Mg

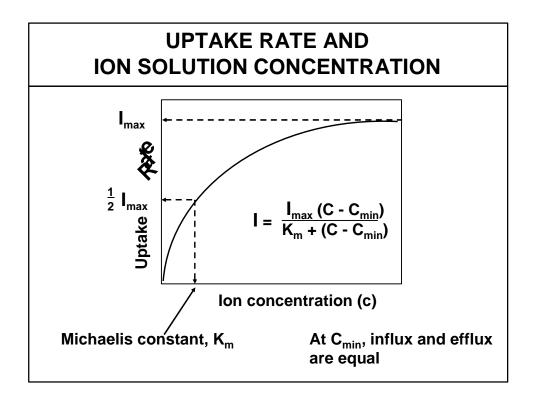
# **RADIAL ROOT TRANSPORT - 2**

- Symplastic
  - lons passing from cell to cell thru
    - plasmodesmata bypassing vacuole
- Release of ions into xylem



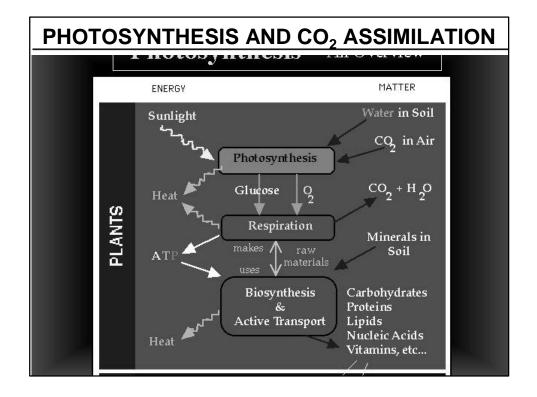


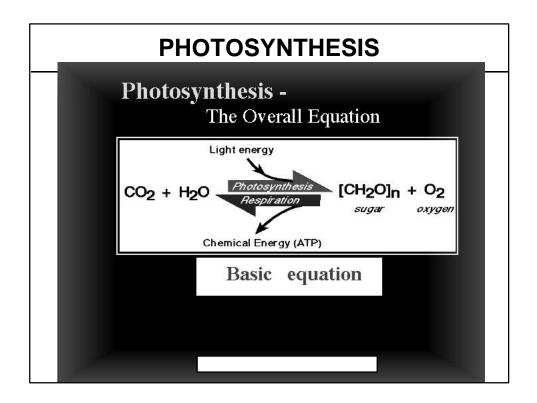


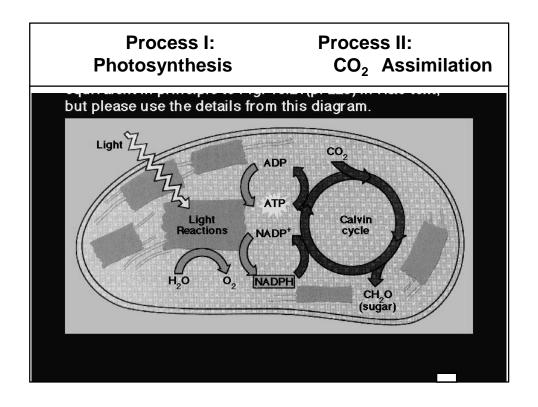


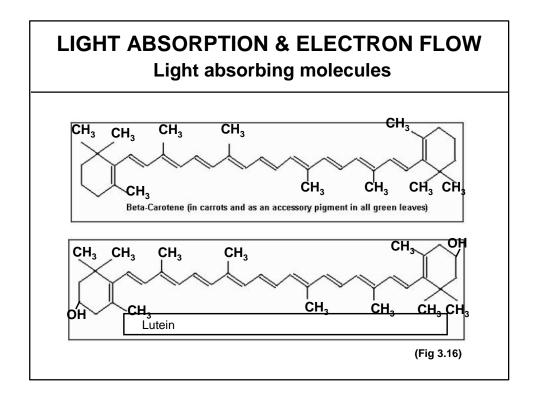
### PHOTOSYNTHESIS AND CO<sub>2</sub> ASSIMILATION GENERAL OBSERVATIONS

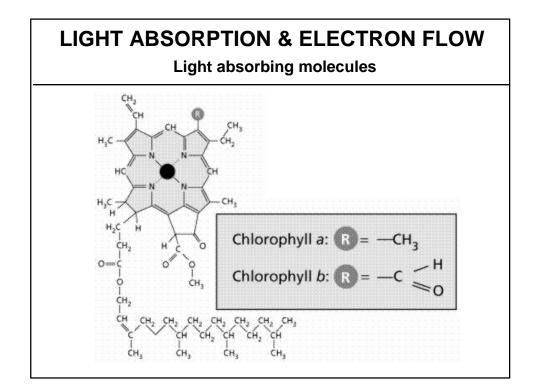
- Many, many reactions are interrelated
- Excess products from one reaction used in another
   minimizes potential toxicity (origin of reaction?)
   product or pH change often initiates reactions
- Note the smooth flow of energy from one reaction to another, even in different cell structures
- Efficient recycling of energy-bearing compounds these often participate in two sets of reactions; one in each "direction"

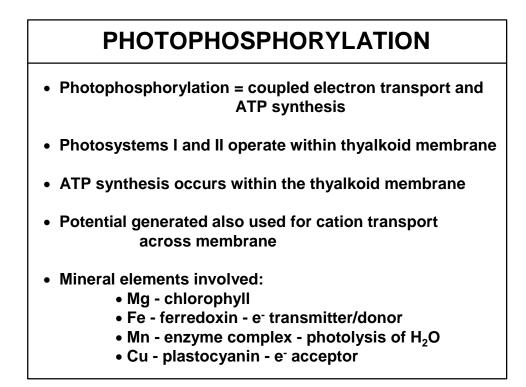


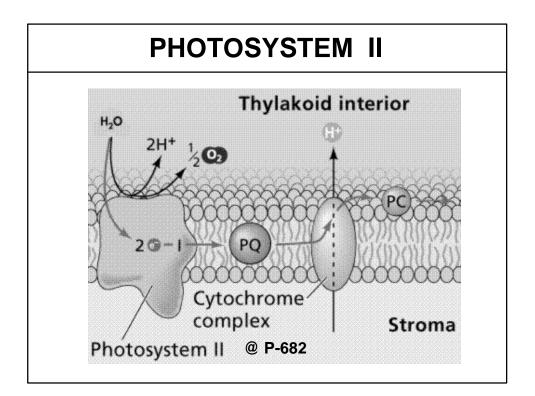


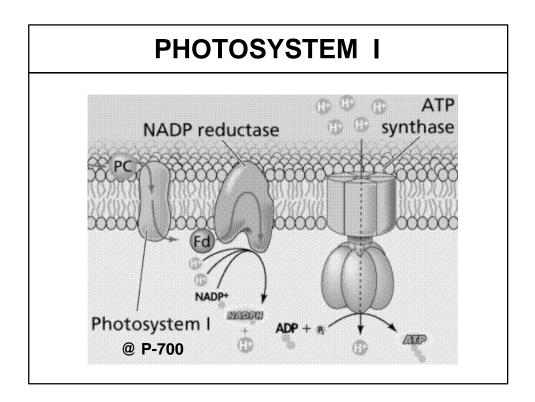


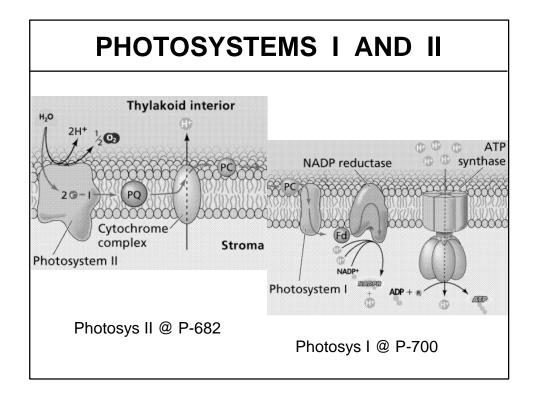


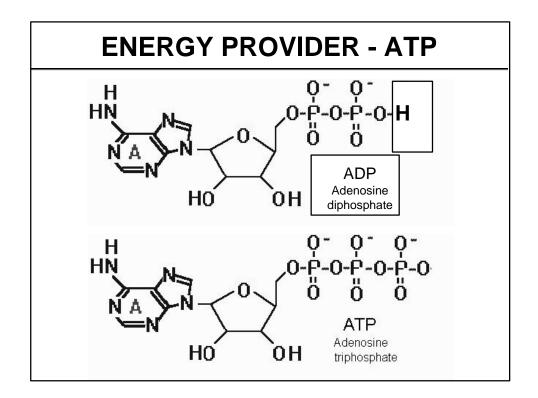


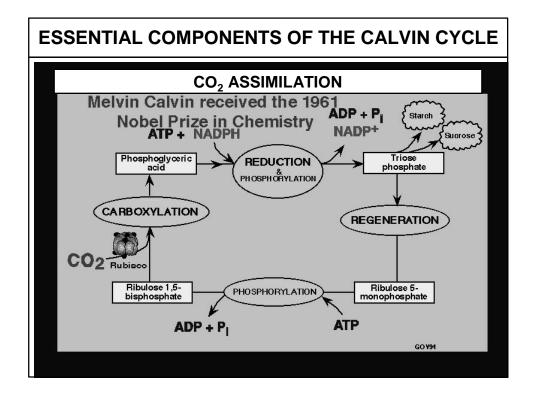


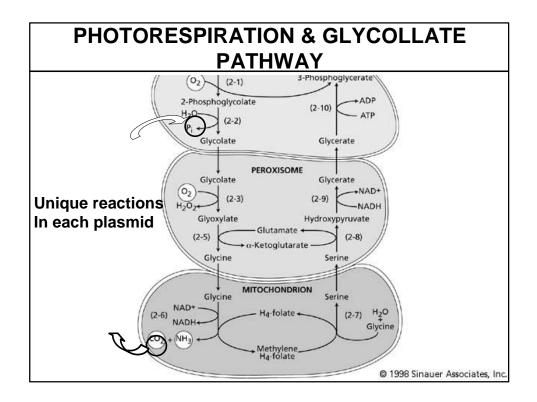


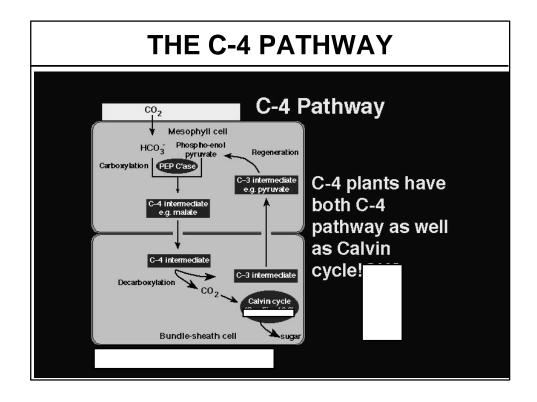


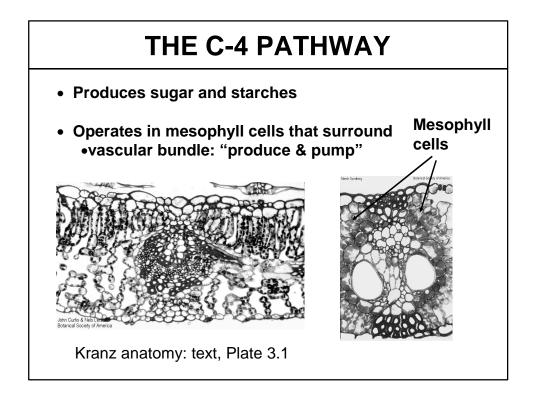


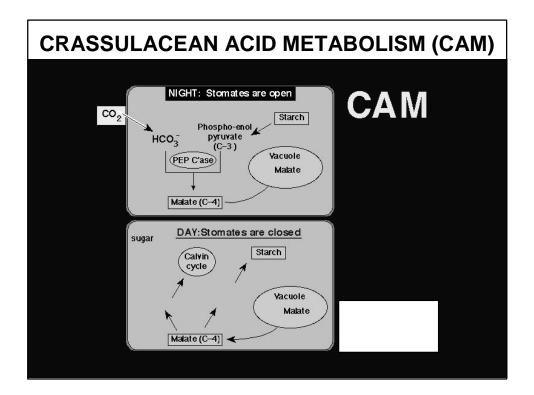




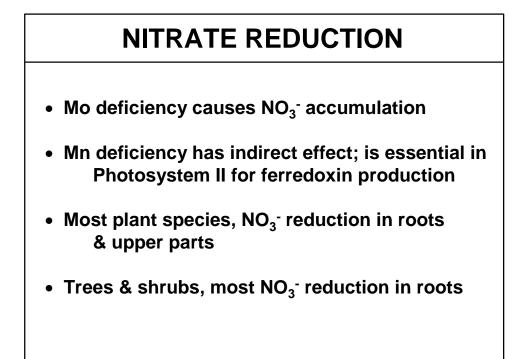








# NITRATE REDUCTION NO<sub>3</sub><sup>-</sup> is common form available to plants Must be reduced to NH<sub>3</sub> before being metabolized Two steps: nitrate to nitrite reduction NADH important nitrite to ammonia reduction ferredoxin important



NITRATE REDUCTION	
NO3 - + NAD(P)H + H + + 2e-	
NO2 + NAD(P) + H20	
Reductase in cytosol	(Cytoplasm)
ed + 8 H++6 e-	NO2 + 6 FL
6 FD <sub>ox</sub> + 2 H <sub>2</sub> 0	₩ NH4 ++
astids, chloroplasts   Nitrite Red	ductase in pl

### NITROGEN FIXATION

- Conducted by bacteria
  - free-living
  - living in symbiosis with higher plant
- *Rhizobium* bacteria legume assoc'n esp. impt.
  - bacteroid enveloped in membrane and embedded in host cell
  - anaerobic environment in bacteroid
  - ATP and ferredoxin important in process
  - NH<sub>3</sub> released into cells for utilization

